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DOI:

[10.1016/j.geodrs.2023.e00697](https://doi.org/10.1016/j.geodrs.2023.e00697)

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*Document Version*

Publisher's PDF, also known as Version of record

*Citation for published version (Harvard):*

Guo, Y, Saiz, E, Radu, A, Sonkusale, S & Ullah, S 2023, 'Prolonged flooding followed by drying increase greenhouse gas emissions differently from soils under grassland and arable land uses', *Geoderma Regional*, vol. 34, e00697. <https://doi.org/10.1016/j.geodrs.2023.e00697>

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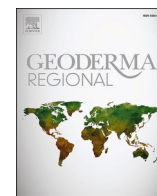
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# Prolonged flooding followed by drying increase greenhouse gas emissions differently from soils under grassland and arable land uses

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## ARTICLE INFO

### Keywords:

Prolonged flooding  
Flooding-drying  
Arable soil  
Grassland soil  
Greenhouse gas  
Net zero greenhouse gas emissions

## ABSTRACT

Several climate change scenarios have predicted that heavy precipitation could result in prolonged flooding (PF) and flooding–drying (FD) of soils under agriculture. The influence of PF and FD on soil greenhouse gas (GHG) fluxes and ammonium-nitrogen ( $\text{NH}_4^+\text{-N}$ ) and nitrate-nitrogen ( $\text{NO}_3^-\text{-N}$ ) dynamics of arable and grassland soils, the dominant land-use types in the UK, remain unclear. A two-month soil incubation experiment was conducted to determine the impact of PF and FD on soil N dynamics and GHG fluxes from arable and grassland soils. Arable soil emitted more  $\text{N}_2\text{O-N}$  when soil moisture exceeded 100% water-holding capacity (WHC) compared to grassland soil under PF. Grassland soils exhibited increased  $\text{N}_2\text{O-N}$  emissions than arable soils when soil moisture was lower than 100% WHC under FD. When soil moisture exceeded 100% WHC, the available  $\text{NO}_3^-\text{-N}$  in the soil contributed 58% of  $\text{N}_2\text{O-N}$  emissions potentially by denitrification from grassland. When soil moisture was lower than 100% WHC, soil  $\text{NH}_4^+\text{-N}$  and  $\text{NO}_3^-\text{-N}$  contributed 71% of  $\text{N}_2\text{O-N}$  emissions, which suggests coupling of nitrification-denitrification processes in driving high emissions from grassland soils. The  $\text{N}_2\text{O-N}$  and  $\text{CO}_2\text{-C}$  emissions increased with the incubation time under FD. Moreover, FD significantly increased  $\text{N}_2\text{O-N}$ ,  $\text{CO}_2\text{-C}$ , and  $\text{CH}_4\text{-C}$  emissions in grassland soil by 0.93, 2.15, and 37.29 times more than arable soil, respectively. These findings points to important tipping points in the source strengths of GHG fluxes from the two land use types differently. Future land use changes should consider the contribution of the changing dynamics of GHG fluxes in light of climate extremes and its implications for net zero greenhouse gas emission ambitions.

## 1. Introduction

Carbon dioxide ( $\text{CO}_2$ ), methane ( $\text{CH}_4$ ), and nitrous oxide ( $\text{N}_2\text{O}$ ) are the primary greenhouse gases (GHGs) present in the Earth's atmosphere (IPCC, 2022). The atmospheric  $\text{CO}_2$  concentration has increased to approximately 420 ppm and future rapid increase is expected to reach 550 ppm by mid-century and 1000 ppm by the end of this century (IPCC, 2022). The  $\text{N}_2\text{O}$  is a long-lived GHG with a long-term global warming potential 300 times greater than  $\text{CO}_2$  (Carneiro et al., 2010) and is a strong stratospheric ozone-depleting substance (Thompson et al., 2019). The  $\text{N}_2\text{O}$  concentration in the atmosphere has also risen steadily since the mid-twentieth century (IPCC, 2022), from approximately 290 ppb in 1940 to 330 ppb in 2017 (Park et al., 2012). The atmospheric  $\text{CH}_4$

continues to rise at a rate of approximately 22 Tg every year, as global sources are larger than sinks (IPCC, 2001). Agricultural soils are important source of GHGs (Liu et al., 2019). They are considered the main source of non- $\text{CO}_2$  anthropogenic GHG and are responsible for 78.6% of  $\text{N}_2\text{O}$  and 39.1% of  $\text{CH}_4$  emissions worldwide (IPCC, 2022). The increasing atmospheric GHG concentrations are inducing changes in climate including shifts in the intensity and duration of precipitation that are very likely to feedback into the production and emission of GHG (IPCC, 2022).

Nitrogen (N) is a necessary nutrient in agricultural ecosystems (Fixon and West, 2002) as it is critical for plant growth (Gilsanz et al., 2016); however, the nitrogen use efficiency of applied fertilizer N by plants is <40% (Chen et al., 2008). In agricultural soils, the applied fertilizer N

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<https://doi.org/10.1016/j.geodrs.2023.e00697>

Received 27 February 2023; Received in revised form 20 June 2023; Accepted 22 August 2023

Available online 23 August 2023

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contributes to  $\text{N}_2\text{O}$  emissions. The  $\text{N}_2\text{O}$  can be produced from several biological processes, including nitrification, denitrification, codenitrification, dissimilatory nitrate reduction to ammonia, and chemo denitrification, which involve different microbial groups in soil (Harter et al., 2014). The two main microbial processes that produce  $\text{N}_2\text{O}$  are nitrification and denitrification (Hu et al., 2015; Sgouridis and Ullah, 2017). Nitrifying microbes undertake biological oxidation of ammonium ( $\text{NH}_4^+$ ) to nitrite ( $\text{NO}_2^-$ ) and further to nitrate ( $\text{NO}_3^-$ ) and can produce  $\text{N}_2\text{O}$  by nitrification under aerobic soil conditions (Wrage et al., 2001). Denitrifying microbes reduce  $\text{NO}_3^-$  to  $\text{NO}_2^-$ , nitric oxide ( $\text{NO}^-$ ),  $\text{N}_2\text{O}$ , and molecular nitrogen ( $\text{N}_2$ ) by denitrification under anaerobic conditions (Köster et al., 2013). The majority of naturally occurring emissions of  $\text{CH}_4$  can be attributed to the biogenic processes of methanogens (methane-emitting microorganisms), as a final step in the anaerobic decomposition of organic matter (Gütlein et al., 2018). These microorganisms are predominant in anaerobic areas rich in organic carbon (Conrad, 2009). Carbon dioxide emissions come from the soil mineralization of soil organic carbon (SOC) (Guo et al., 2019). In intact soil systems where plants are present,  $\text{CO}_2$  emissions originate from both microbial and plant respiration.

Soil  $\text{CO}_2$ ,  $\text{N}_2\text{O}$ , and  $\text{CH}_4$  exchanges are driven by aerobic (when soil moisture (SM) was lower) and anaerobic (when SM was higher) microbial processes (Gütlein et al., 2018), which are in turn influenced by soil properties and environmental factors (temperature, precipitation), soil physical (texture and oxygen concentration) and chemical properties (pH and nutrient availability) (Miller et al., 2020). Soil pH manipulates the microbial community structure, and therefore, the decomposition or accumulation of SOC (Malik et al., 2018) to influence the GHG emissions.

Land-use change affects soil GHG emissions due to changes in vegetation, soil hydrology, and nutrient management (Gütlein et al., 2018). Grassland dominates the landscape in the United Kingdom (40%) (Caroline et al., 2017). A total of 56,506  $\text{km}^2$  of land is classed as arable and 96,949  $\text{km}^2$  is classed as grasslands (improved, neutral, calcareous, and acid) in the UK (Rowland et al., 2017). Moreover, GHG emissions from soils are also linked to the hydrological conditions, with deep water tables favouring  $\text{CO}_2$  emissions, shallow water tables (less than  $\sim 20$  cm) favouring  $\text{CH}_4$  emissions, and fluctuating water tables potentially being conducive to  $\text{N}_2\text{O}$  emissions (Petersen et al., 2012; Poyda et al., 2016; Wilson et al., 2016).

Climate change is predicted to cause major changes in precipitation patterns with increased frequency and intensity of large rainfall events (IPCC, 2007). Flooding–drying (FD) cycles of soil may expose unavailable (physically protected) soil organic matter to microbes through breakdown of soil aggregates (Zhang et al., 2020) as well as changing the redox conditions in soils with implications for shifts in net  $\text{N}_2\text{O}$ ,  $\text{CH}_4$ , and  $\text{CO}_2$  emissions into the atmosphere. Changes in flooding frequency of soils following future climate change will likely affect the timing and magnitude of  $\text{N}_2\text{O}$  emissions from the soil to the atmosphere (Jørgensen and Elberling, 2012). Production and emission of GHGs may vary depending on soil physical properties, SM status as influenced by precipitation under climate change extremes (Fay et al., 2010), and fertilization (Mazza et al., 2018). Therefore, changes in soil saturation following extreme precipitation events in the future are likely to change the timing and extent of GHG emissions from soils. In fact, soil saturation is more likely to be severe in agricultural soils, where precipitation interception is low compared with forest soils.

In this study, we conducted an incubation experiment to identify the effect of prolonged flooding (PF) and FD (extreme SM condition) of soils on soil  $\text{N}_2\text{O}$ -N,  $\text{CO}_2$ -C and  $\text{CH}_4$ -C emissions and  $\text{NH}_4^+$ -N and  $\text{NO}_3^-$ -N dynamics under grassland and arable land uses. We hypothesized that: (1) arable soil will result in high  $\text{N}_2\text{O}$ -N,  $\text{CO}_2$ -C and  $\text{CH}_4$ -C emissions and GWP under PF and FD due to high N fertilizer addition; and (2) FD will lead to high  $\text{N}_2\text{O}$ -N,  $\text{CO}_2$ -C and  $\text{CH}_4$ -C emissions and GWP in grassland and arable soil due to faster mineralization of soil organic matter. The objectives of this study were to determine soil  $\text{N}_2\text{O}$ -N,  $\text{CO}_2$ -C, and  $\text{CH}_4$ -C

emissions as influenced by environmental (flooding and draining) and edaphic factors ( $\text{NH}_4^+$ ,  $\text{NO}_3^-$ , and total dissolved nitrogen contents) in fertilized grassland and arable soil in the UK. The result could help us to understand the GHG emissions after conversion of arable land use to grasslands (in order to improve soil health and sequester carbon) under climate change scenarios (heavy rain).

## 2. Materials and methods

### 2.1. Study site description and soil collection

Soil samples were taken by a soil auger (core sampler) (up to 10 cm) from an experimental demonstration mixed land use farm at Honeydale Farm ( $51^\circ 51' \text{N}$ ,  $1^\circ 35' \text{W}$ ), located in the Evenlode Valley in the heart of the Cotswolds, UK. Honeydale Farm is a 107-acre (43 ha) farm. The nearest climate station is Little Rissington ( $51^\circ 51' \text{N}$ ,  $1^\circ 41' \text{W}$ ) which is 8 miles away. The annual monthly maximum temperature is  $13.41^\circ \text{C}$  and monthly minimum temperature is  $5.89^\circ \text{C}$ . Air frost occurs 46.14 days of the year. The cumulative number of sunshine hours is 1631.53 and the average rainfall is 809.63 mm per year. Monthly mean wind speed at 10 m is 10.30 knots. General cropping is over chalk; spring and autumn cereals can be grown. The poor SM is likely to be the main limiting factor on crop yield (Cranfield soil and agrifood institute, 2023).

Grassland soil was restored for 4 years with 'HERBAL' (Cotswold Grass Seeds, Cotswold Business Village, Moreton-in-Marsh, UK) grazing ley and used for grazing. Previously, the land was used for arable cultivation (wheat and oats) over years for decades before changing to grass. HERBAL all-round mixture (20 species) provides wholesome and substantial forage for grazing and occasional cutting. It can also provide grazing for early turnout and continues to produce forage right through the summer and autumn. Containing deep-rooting ingredients, this ley not only improves soil structure but also draws up essential vitamins and minerals for the ruminant animal (<https://www.cotswoldseeds.co.uk/products/542/herbal-grazing-ley-four-year-drought-resistant-ley>). The arable soil was a control plot with winter wheat planted in October. The soil N fertilization rate in the Honeydale farm ranges between 120 and 150 kg N per ha depending on the year of production in light of N fertilization rates recommended by DEFRA. This soil has been under wheat or barley cultivation for 30 years. The parent material of both the grassland and arable soil is Oolitic limestone. Arable soil and grassland soil are inceptisols (Soil Taxonomy-USDA).

Soil samples were carefully collected from 0 to 20 cm soil depth from the grassland and arable plots on May 18, 2022 (around 20 soil samples were taken in both grassland and arable plots and then mixed to give one grassland and one arable soil sample). Fresh soil samples were stored in a refrigerator at  $4^\circ \text{C}$  after the removal of plant residues and stones and passing through a 2-mm sieve for homogenization. The main soil

**Table 1**

Selected soil physical and chemical properties before the experiment. TOC is total dissolved organic carbon, TDN is total dissolved nitrogen, TC is total carbon, TN is total nitrogen, and WHC is water-holding capacity. Means of the same index and indicated by the same lower-case letter are not significantly different at  $P < 0.05$  on the basis of Tukey's HSD test.

Soil type	$\text{NH}_4^+$ -N $\text{mg kg}^{-1}$	$\text{NO}_3^-$ -N $\text{mg kg}^{-1}$	TOC-C $\text{mg kg}^{-1}$	TDN-N $\text{mg kg}^{-1}$	TC g $\text{kg}^{-1}$	TN g $\text{kg}^{-1}$
Grassland	0.63a ( $\pm 0.07$ )	10.03a ( $\pm 0.76$ )	88.67b ( $\pm 1.10$ )	22.82a ( $\pm 0.26$ )	69.81b ( $\pm 0.95$ )	4.22b ( $\pm 0.13$ )
Arable	0.34a ( $\pm 0.06$ )	277.28b ( $\pm 4.65$ )	61.49a ( $\pm 1.76$ )	254.19b ( $\pm 4.24$ )	53.88a ( $\pm 0.58$ )	3.71a ( $\pm 0.21$ )
Soil type	Clay (%)	Silt (%)	Sand (%)	pH	WHC (%)	C/N
Grassland	46.54a ( $\pm 0.47$ )	33.23a ( $\pm 0.70$ )	20.23b ( $\pm 1.01$ )	7.36b ( $\pm 0.04$ )	72.42a ( $\pm 0.44$ )	16.66a ( $\pm 0.39$ )
Arable	46.12a ( $\pm 0.74$ )	39.45b ( $\pm 0.86$ )	14.43a ( $\pm 0.80$ )	6.90a ( $\pm 0.01$ )	70.52a ( $\pm 0.21$ )	14.95a ( $\pm 0.98$ )

Note: Average value ( $\pm$  relative standard deviation),  $n = 4$ .

properties are given in Table 1.

## 2.2. Incubation experiment

Grassland and arable soils were each packed into a 1 L Mason Jar pot to achieve a bulk density of  $0.9 \text{ g cm}^{-3}$  (field bulk density) with a 10 cm height. Three replications were set for each land use type.

All pots were fertilized with  $100 \text{ mg NH}_4^+\text{-N kg}^{-1}$  soil ( $180 \text{ kg NH}_4^+\text{-N ha}^{-1}$ ) using ammonium sulphate salt ( $(\text{NH}_4)_2\text{SO}_4$ ) in order to determine the response of fertilized soils to saturation. Deionized water was added to keep SM at 110% water-holding capacity (WHC) and the pots were incubated for 4 weeks. Soil moisture was kept constant by weighing the pot each day to simulate PF (Fig. 1).

All pots were fertilized with  $100 \text{ mg NH}_4^+\text{-N kg}^{-1}$  soil again using  $(\text{NH}_4)_2\text{SO}_4$  after 4 weeks of incubation under PF. This was followed by adjustment of moisture to 130% WHC for 4 weeks to mimic another intense precipitation event. Water loss (e.g., evaporation) was not replaced over time to simulate drying following flooding (FD) (Fig. 1).

## 2.3. Collection and measurement of GHGs

The  $\text{CO}_2\text{-C}$ ,  $\text{N}_2\text{O-N}$ , and  $\text{CH}_4\text{-C}$  emission rates were calculated by measuring the concentration of these gases in the headspace of closed pots at different times. The  $\text{CO}_2\text{-C}$ ,  $\text{N}_2\text{O-N}$ , and  $\text{CH}_4\text{-C}$  samples were collected once a day during the first week, once per two days during the second week, and once per three days during the last two weeks (Guo et al., 2021a). The pots were first closed by adding lids, gases were mixed by collecting and injecting the gas in and out of the pots a few times with a syringe, and then samples were collected from the headspace at 0, 30, and 60 min after lid closure (Guo et al., 2021b). A 20-ml syringe and hypodermic needle was used to collect the gas samples and inject them into pre-evacuated 12-ml glass headspace exetainer vials fitted with a chloro-butyl rubber septum (Guo et al., 2022a). Gas sampling was carried out between 09:00 and 10:00 a.m. The pots were left open outside of the sampling periods.

The concentrations of  $\text{CO}_2\text{-C}$ ,  $\text{N}_2\text{O-N}$ , and  $\text{CH}_4\text{-C}$  in the 12-ml glass exetainer vials were measured by gas chromatography (Agilent 7890A GC, Agilent, CA, USA). An electron capture detector (ECD) was used to detect the  $\text{N}_2\text{O}$ , with a temperature of  $350^\circ\text{C}$  and an  $\text{N}_2$  as carrier gas. A flame ionization detector (FID) was used to detect the  $\text{CH}_4$  and  $\text{CO}_2$  concentrations, with a temperature of  $250^\circ\text{C}$  and  $\text{N}_2$  as the carrier gas. The  $\text{CH}_4$  was eluted and analysed first. Secondly,  $\text{CH}_4$  (equal to  $\text{CO}_2$ ) was

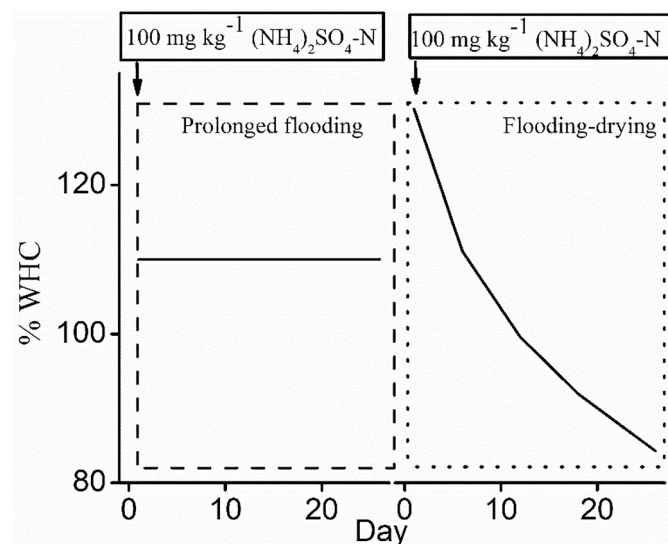


Fig. 1. Soil water content as percentage water-holding capacity (WHC) during the prolonged flooding and flooding-drying incubation.

analysed after the sample was passed through a methanizer to convert the  $\text{CO}_2$  to  $\text{CH}_4$  for detection.

Total  $\text{CO}_2\text{-C}$ ,  $\text{N}_2\text{O-N}$ , and  $\text{CH}_4\text{-C}$  emissions during the experimental period were calculated from the daily emissions of the gases (Guo et al., 2022b). We used the fluxes of  $\text{CO}_2\text{-C}$ ,  $\text{N}_2\text{O-N}$ , and  $\text{CH}_4\text{-C}$  multiplied by the number of hours during sampling (24 h in the first week, 48 h in the second week, and 72 h in the last six weeks) then added them together to obtain the total emissions ( $\text{mg kg}^{-1}$  or  $\mu\text{g kg}^{-1}$ ). The global warming potential (GWP) was used to quantitatively assess the relative impacts of  $\text{N}_2\text{O}$ ,  $\text{CH}_4$ , and  $\text{CO}_2$  on climate change by summing the 100-year radiative forcing associated with each of the three measured GHGs (Hawthorne et al., 2017). The conversion factors for the assessment of GWP for soil  $\text{N}_2\text{O}$ ,  $\text{CH}_4$ , and  $\text{CO}_2$  are 298, 25, and 1 respectively (IPCC, 2021). The GWP was calculated as follows:

$$\text{GWP} = \text{Total}_{\text{CO}_2} \times 1 + \text{Total}_{\text{CH}_4} \times 25 + \text{Total}_{\text{N}_2\text{O}} \times 298 \quad (1)$$

where  $\text{Total}_{\text{CO}_2}$  is the total emissions of  $\text{CO}_2$ ,  $\text{Total}_{\text{CH}_4}$  is the total emissions of  $\text{CH}_4$ , and  $\text{Total}_{\text{N}_2\text{O}}$  is the total emissions of  $\text{N}_2\text{O}$ .

## 2.4. $\text{NH}_4^+\text{-N}$ and $\text{NO}_3^-\text{-N}$ measurement

Three types (available, labile and stable) of  $\text{NH}_4^+\text{-N}$  and  $\text{NO}_3^-\text{-N}$  in soil were selected to determine their influence on GHG emissions. The  $\text{NH}_4^+\text{-N}$  and  $\text{NO}_3^-\text{-N}$  in pore water ( $\text{WNH}_4$  and  $\text{WNO}_3$ ) are available for soil microbes to use since they are present in the pore water. The  $\text{NH}_4^+\text{-N}$  and  $\text{NO}_3^-\text{-N}$  in 0.5 M HCl from ion-exchange resin membranes (IERM) ( $\text{INH}_4$  and  $\text{INO}_3$ ) are labile for soil microbes to use. The  $\text{NH}_4^+\text{-N}$  and  $\text{NO}_3^-\text{-N}$  in 2 M KCl from soil extraction ( $\text{SNH}_4$  and  $\text{SNO}_3$ ) are stable for soil microbes to use.

### 2.4.1. Soil pore water collection

Ten millilitres of soil pore water were collected once a day during the first week, once per two days during the second week, and once per three days during the last two weeks under PF. The samples were collected once a day during the first week, and once per two days during the first two weeks under FD (the pore water was collected in the FD treatment when the SM was higher than 100% WHC,  $\text{SM} > 100\%\text{WHC}$ ). The pore water was collected via a fibre microfluidic thread.

### 2.4.2. Ion-exchange resin membranes

The IERMs included anion-exchange membranes (AEMs) that sorb  $\text{NO}_3^-\text{-N}$  and cation-exchange membranes (CEMs) that sorb  $\text{NH}_4^+\text{-N}$  from soil pore water through diffusion. These membranes were inserted into a 7.5 cm depth of soil for passive sensing of the mineral N during the course of the incubation. The CEM and AEM (SNOWPURE, San Clemente, CA, USA) were put in the deionized water for 48 h at  $90^\circ\text{C}$ , followed by drying the membrane. The CEM was saturated with  $\text{H}^+$  by leaving CEM strips overnight in 2 M HCl. The AEM was saturated with  $\text{HCO}_3^-$  by leaving the AEM strips overnight in 1 M  $\text{NaHCO}_3$ . IERMs were rinsed free of excess HCl and  $\text{NaHCO}_3$  with deionized water. A further 24-h equilibration and removal of the chemicals with deionized water was required for the CEM and AEM. Prepared strips were kept moisturized in sealed plastic bags prior to installation. The CEM and AEM were inserted into the soil and replaced each week. The IERMs were carefully rinsed with deionized water until all traces of soil were removed. Desorption of ions was achieved by shaking each strip (cations and anions separated) for 2 h in 17.5 ml of 0.5 M HCl at 200 rpm.

### 2.4.3. Soil extraction

Subsoil samples were collected to a depth of 10 cm on Days 5, 12, 19, and 26 of the incubation. For soil mineral N analysis, 2 g fresh soil was extracted with 20 ml of 2 M KCl solution (1:10) for 1 h on a reciprocating shaker. The suspensions obtained were centrifuged at  $112 \times g$  for 10 min, filtered through a  $0.45\text{-}\mu\text{m}$  Syringe Filter PES, and stored at  $4^\circ\text{C}$  for analysis in the next day. The solution would be frozen for storage if the



analysis is not immediately.

The concentrations of  $\text{INH}_4$ ,  $\text{INO}_3$ ,  $\text{WNO}_3$ ,  $\text{SNH}_4$ , and  $\text{SNO}_3$  were measured using an AQ400 Discrete Analyser (SEAL Analytical Ltd., Wrexham, United Kingdom).

## 2.5. Measurement of additional soil properties

Ten grams of air-dried soil were mixed with 25 ml deionized water (1:2.5) and shaken for 30 min at 200 rpm for pH measurement. The pH of the upper clear liquid was measured using a pH meter (Mettler Toledo, FiveEasy™, FE20). Soil total dissolved organic carbon (TOC) and soil total dissolved nitrogen (TDN) were extracted with 2 M KCl solution (1:10) and measured by TOC-L CPH with ASI-L (Shimadzu, Kyoto, Japan). Soil particle sizes were measured using a Mastersizer 2000 particle analyser (Malvern Panalytical, Malvern, United Kingdom). Soil total carbon (TC) and total nitrogen (TN) were measured using a Flash Smart elemental analyser (Thermo Scientific, Massachusetts, US).

## 2.6. Statistical analysis

Tukey's honest significant difference (HSD) test was used to compare the means of soil total  $\text{CO}_2\text{-C}$ ,  $\text{N}_2\text{O-N}$ , and  $\text{CH}_4\text{-C}$  emissions and soil GWP under different treatments if the treatment effects were significant at the  $P < 0.05$  level. Differences for soil initial index (soil properties before experiment) were compared between grassland soil and arable soil under Tukey's HSD test.

Principal component analysis (PCA) were used to test the relation among all GHGs samples. Correlation analysis were used to test the relation between GHGs and soil properties. Variation partitioning were used to found out the contribution of soil properties to GHGs emissions. All statistical analyses were performed using R statistical language (Oakland, CA, USA). PCA and variance decomposition of  $\text{CO}_2\text{-C}$ ,  $\text{N}_2\text{O-N}$ , and  $\text{CH}_4\text{-C}$  emissions were computed using the "RDA" and "VARPART" function of the "vegan" library for R (Oksanen and O'Hara, 2005).

## 3. Results

### 3.1. Soil moisture changes with time during flooding–drying

Soil moisture decreased with time and decreased to 100% WHC in 10 days under FD (Fig. 2). Soil moisture decreased faster in arable soil than in grassland soil (Fig. 2).

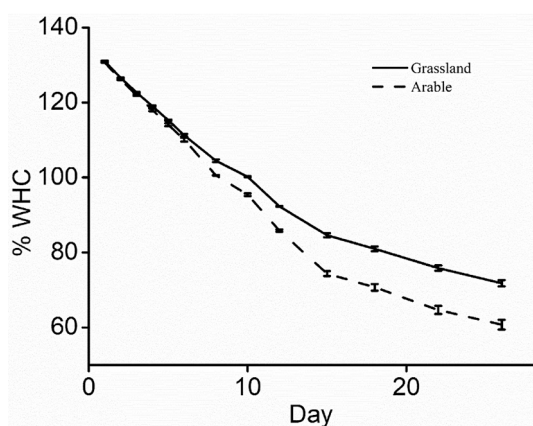


Fig. 2. Soil moisture changes with time under grassland and arable soil during the flooding–drying treatment. The data values are means of three independent pot replicates and error bars represent the standard error of the means ( $n = 3$ ).

### 3.2. Soil $\text{CO}_2\text{-C}$ , $\text{N}_2\text{O-N}$ , and $\text{CH}_4\text{-C}$ emissions

#### 3.2.1. Rate of $\text{CO}_2\text{-C}$ , $\text{N}_2\text{O-N}$ , and $\text{CH}_4\text{-C}$ emissions

The rate of  $\text{N}_2\text{O-N}$ ,  $\text{CO}_2\text{-C}$ , and  $\text{CH}_4\text{-C}$  emissions in grassland and arable soils did not change over time under PF, and that of  $\text{N}_2\text{O-N}$  and  $\text{CH}_4\text{-C}$  emissions in arable soil did not change over time under FD (Fig. 3). The arable soil had a higher rate of  $\text{N}_2\text{O-N}$  emissions than the grassland soil under PF (Fig. 3).

The rate of  $\text{N}_2\text{O-N}$  emissions in grassland soil under FD increased over time, especially after 10 days, when the SM level started to drop below 100% WHC. Arable soil had a higher rate of  $\text{N}_2\text{O-N}$  emissions than grassland soil during the first 15 days under saturated conditions, but decreased when moisture started to fall below 80% WHC (Fig. 3). Thus, arable soil sustained high  $\text{N}_2\text{O}$  emissions under water saturation conditions and grassland enhanced  $\text{N}_2\text{O}$  production following a decline in SM below 80% WHC from Day 15 onwards. The rate of  $\text{CO}_2\text{-C}$  emissions in arable and grassland soil under FD increased over time (Fig. 3). Grassland soil had a higher rate of  $\text{CO}_2\text{-C}$  emissions than arable soil under FD (Fig. 3). The rate of  $\text{CH}_4\text{-C}$  emissions in grassland soil under FD increased and then decreased over time (Fig. 3). Arable soil had a lower rate of  $\text{CH}_4\text{-C}$  emissions than grassland soil under FD (Fig. 3).

#### 3.2.2. Total $\text{CO}_2\text{-C}$ , $\text{N}_2\text{O-N}$ , and $\text{CH}_4\text{-C}$ emissions

The grassland soil under FD had the highest ( $P < 0.05$ ) total  $\text{N}_2\text{O-N}$ ,  $\text{CO}_2\text{-C}$ , and  $\text{CH}_4\text{-C}$  emissions (Table 2). The grassland soil under PF had the lowest ( $P < 0.05$ ) total  $\text{N}_2\text{O-N}$  emissions (Table 2). The grassland and arable soil under PF had the lowest ( $P < 0.05$ ) total  $\text{CO}_2\text{-C}$  emissions (Table 2). The arable soil under FD, and the grassland and arable soil under PF had the lowest ( $P < 0.05$ ) total  $\text{CH}_4\text{-C}$  emissions (Table 2).

#### 3.2.3. Global warming potential

Grassland soil had highest ( $P < 0.05$ ) GWP under FD (Table 3). Grassland soil had lowest ( $P < 0.05$ ) GWP under PF (Table 3). Arable soil had a moderate GWP under both PF and FD (Table 3).

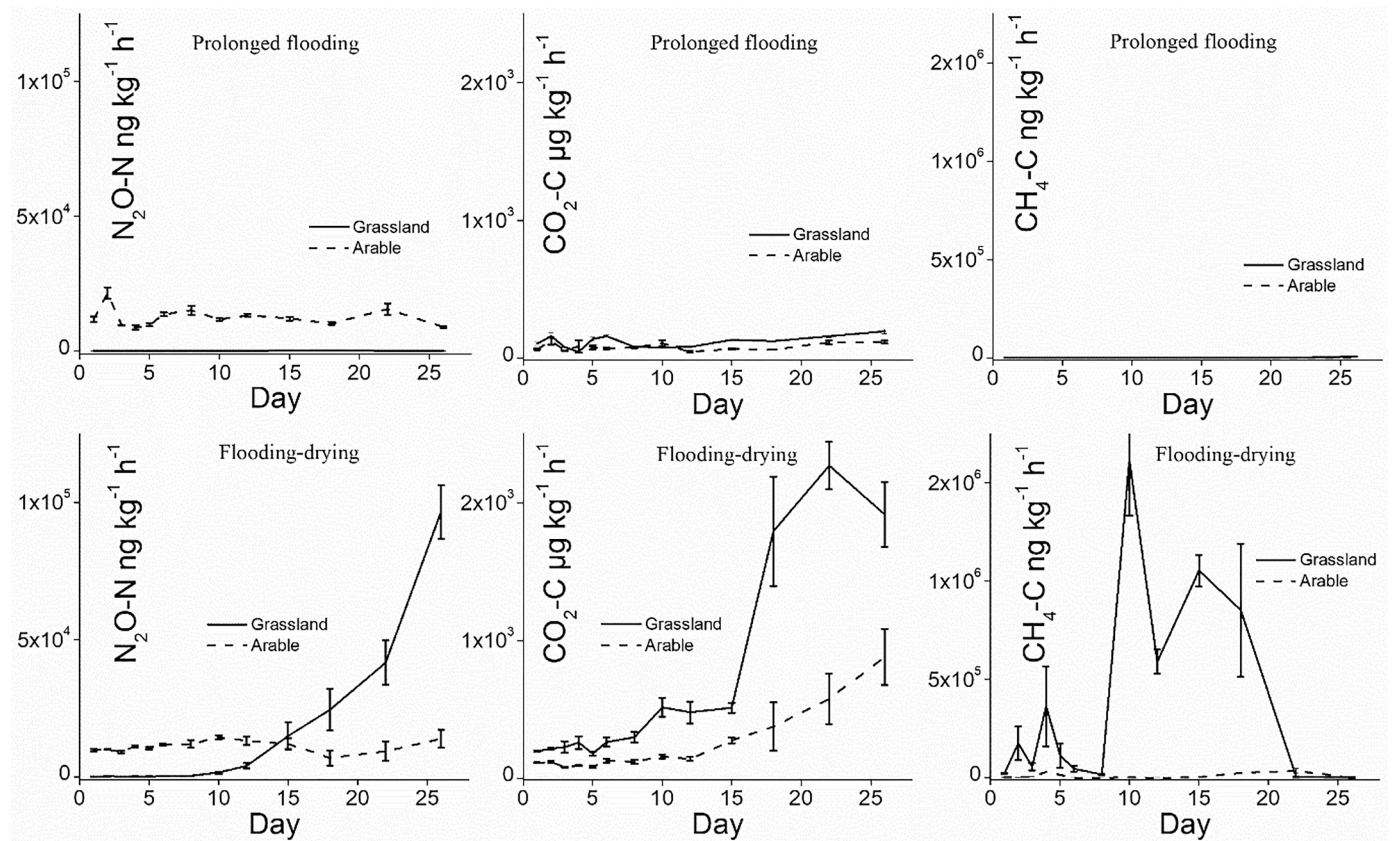
### 3.3. Soil $\text{NH}_4^+\text{-N}$ and $\text{NO}_3^-\text{-N}$ concentrations

The  $\text{WNO}_3$  concentrations of grassland soil decreased and  $\text{WNO}_3$  concentrations of grassland increased over time under PF and FD but  $\text{WNO}_3$  concentrations decreased rapidly under FD (Fig. 4a). The  $\text{WNO}_3$  concentrations of arable soil decreased and  $\text{WNO}_3$  concentrations of arable soil rapidly increased with time under PF (Fig. 4a). The  $\text{WNO}_3$  concentrations of arable soil decreased rapidly, whereas  $\text{WNO}_3$  concentrations increased slightly over time under FD (Fig. 4a).

The  $\text{INH}_4$  and  $\text{INO}_3$  concentrations of grassland soil and  $\text{INH}_4$  concentrations of arable soil were very small and did not change substantially over time under PF and FD (Fig. 4b). The  $\text{INO}_3$  concentrations of arable soils did not change over time under PF, and  $\text{INO}_3$  concentrations was highest in arable land under PF compared with other soils and treatments (Fig. 4b). The  $\text{INO}_3$  concentrations of arable soil under FD decreased in the first 20 incubation days and increased thereafter (Fig. 4b).

The  $\text{SNH}_4$  concentrations of grassland decreased and  $\text{SNO}_3$  concentrations of grassland were stable over time under PF (Fig. 4c). The  $\text{SNH}_4$  concentrations of grassland soil decreased rapidly and  $\text{SNO}_3$  concentrations of grassland soil increased slightly over time under FD (Fig. 4c). The  $\text{SNH}_4$  concentrations of arable soil decreased slightly and  $\text{SNO}_3$  concentrations of arable soil increased slightly over time under PF (Fig. 4c). The  $\text{SNH}_4$  concentrations of arable soil decreased rapidly and  $\text{SNO}_3$  concentrations of arable soil increased slightly over time under FD (Fig. 4c).

The TDN-N concentrations of grassland soil decreased and TOC-C concentrations of grassland increased over time under PF (Fig. 4d). The TDN-N and TOC-C concentrations of grassland decreased over time under FD (Fig. 4d). The TDN-N concentrations of arable decreased and TOC-C concentrations of arable soil remained stable over time under PF



**Fig. 3.** Rate of soil  $\text{CO}_2\text{-C}$ ,  $\text{N}_2\text{O-N}$ , and  $\text{CH}_4\text{-C}$  emissions over time in the grassland and arable soils. The data values are means of three independent pot replicates and error bars represent the standard error of the means ( $n = 3$ ).

**Table 2**

Soil total  $\text{CO}_2\text{-C}$ ,  $\text{N}_2\text{O-N}$ , and  $\text{CH}_4\text{-C}$  emissions. Means of the same gas with the same lower-case letter are not significantly different at the  $P < 0.05$  level on the basis of Tukey's HSD test.

	Soil type	$\text{N}_2\text{O-N}$ mg $\text{kg}^{-1}$	$\text{CO}_2\text{-C}$ mg $\text{kg}^{-1}$	$\text{CH}_4\text{-C}$ mg $\text{kg}^{-1}$
Prolonged flooding	Grassland	0.08 ( $\pm 0.02$ ) a	79.50 ( $\pm 1.50$ ) a	0.70 ( $\pm 0.05$ ) a
	Arable	7.82 ( $\pm 0.21$ ) b	53.72 ( $\pm 2.82$ ) a	0.02 ( $\pm 0.01$ ) a
Flooding-drying	Grassland	13.31 ( $\pm 0.72$ ) c	623.40 ( $\pm 30.01$ ) c	278.02 ( $\pm 43.91$ ) b
	Arable	6.89 ( $\pm 0.76$ ) b	197.89 ( $\pm 21.91$ ) b	7.26 ( $\pm 1.24$ ) a

Note: Average value ( $\pm$  relative standard deviation),  $n = 3$ .

**Table 3**

Global warming potential. Means with same lower-case letter are not significantly different at the  $P < 0.05$  level on the basis of Tukey's HSD test.

GWP (g $\text{CO}_2\text{ kg}^{-1}$ )	Prolonged flooding	Flooding-drying
Grassland	0.39 ( $\pm 0.02$ ) a	24.02 ( $\pm 2.25$ ) c
Arable	7.52 ( $\pm 0.21$ ) b	7.42 ( $\pm 0.84$ ) b

Note: Average value ( $\pm$  relative standard deviation),  $n = 3$ .

and FD (Fig. 4d).

### 3.4. Principal component analysis of soil variation in soil GHG emission

Principal Component 1 (PC1) explained 37.2% of variation in soil GHG emission rates and Principal Component 2 (PC2) explained 26.3%

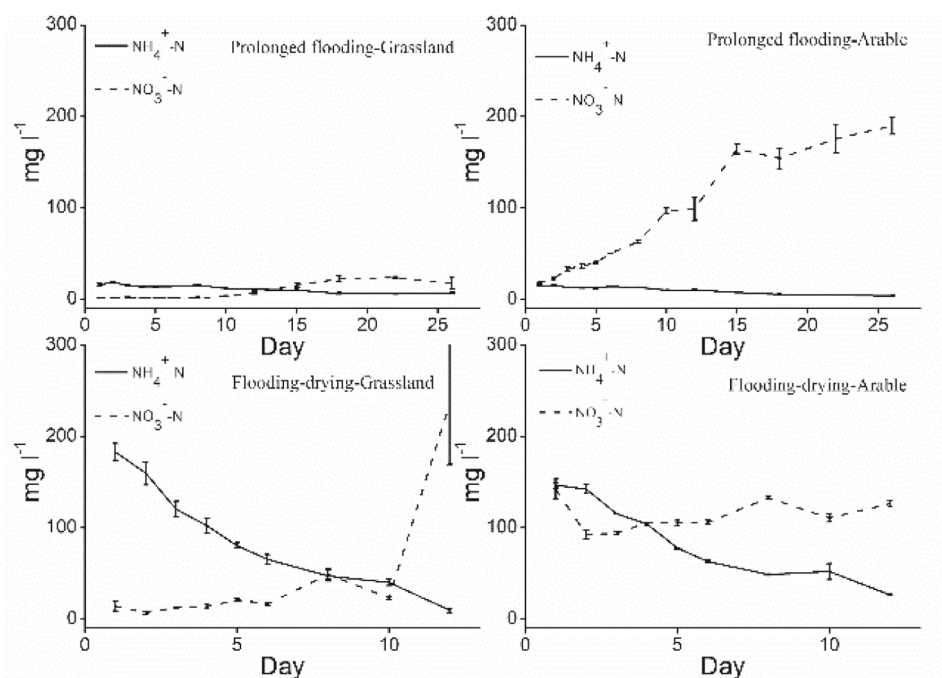
(Fig. 5). Samples of grassland and arable soils with  $\text{SM} > 100\%$  WHC, and grassland and arable soils with  $\text{SM} < 100\%$  WHC were clearly separated from each other (Fig. 5). Grassland and arable soils, with  $\text{SM} > 100\%$  WHC showed a slight relation to soil  $\text{N}_2\text{O-N}$  emissions rate. Grassland and arable soils with  $\text{SM} < 100\%$  WHC had a strong relation to  $\text{N}_2\text{O-N}$  emissions rate, which varied overall and peaked on different days in the two land use types (Fig. 5). The WHC had a strong negative relation to soil  $\text{N}_2\text{O-N}$  emissions rate (increased WHC decreased the soil  $\text{N}_2\text{O-N}$  emissions rate). Soil  $\text{CO}_2\text{-C}$  emissions rate had a strong positive relation to soil  $\text{N}_2\text{O-N}$  emissions rate (Fig. 5).

### 3.5. Correlation analysis of soil variation in soil GHG emissions

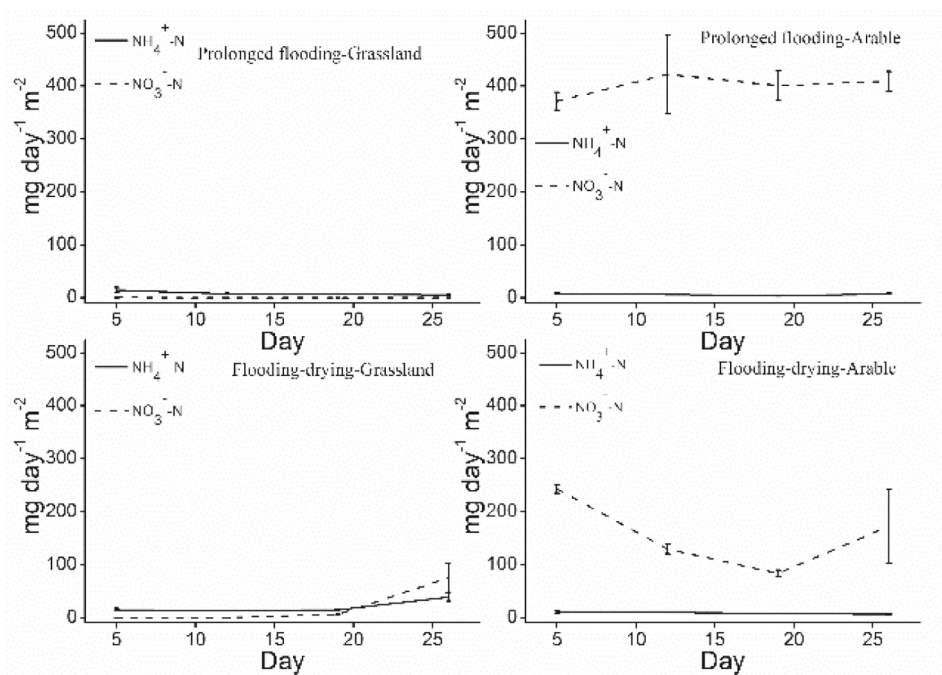
We separated the date between  $\text{SM} > 100\%$  WHC and  $\text{SM} < 100\%$  WHC to conduct the correlation analysis of soil variation since samples of  $\text{SM} > 100\%$  WHC and  $\text{SM} < 100\%$  WHC were clearly separated with each other in the PCA (Fig. 5).

The rate of soil  $\text{N}_2\text{O-N}$  emissions in grassland soils under  $\text{SM} > 100\%$  WHC showed a significant correlation with soil WHC and  $\text{WNO}_3$  (Table 4). The rate of soil  $\text{N}_2\text{O-N}$  emissions from grassland soils under  $\text{SM} < 100\%$  WHC had a significant correlation with soil WHC,  $\text{INH}_4$ ,  $\text{INO}_3$ , and  $\text{SNO}_3$  (Table 4).

The rate of soil  $\text{CO}_2\text{-C}$  emissions in grassland soil under  $\text{SM} > 100\%$  WHC showed a significant correlation with  $\text{INO}_3$ ,  $\text{WNO}_3$ , and  $\text{TOC-C}$  (Table 4). The rate of soil  $\text{CO}_2\text{-C}$  emissions from grassland soil under  $\text{SM} < 100\%$  WHC showed a significant correlation with  $\text{SNO}_3$  and  $\text{TOC-C}$  (Table 4). The rate of soil  $\text{CO}_2\text{-C}$  emissions in arable soil under  $\text{SM} > 100\%$  WHC showed a significant correlation with  $\text{INH}_4$  and  $\text{INO}_3$  (Table 4). The rate of soil  $\text{CO}_2\text{-C}$  emissions from arable soil under  $\text{SM} < 100\%$  WHC showed a significant correlation with soil WHC,  $\text{INH}_4$ ,  $\text{SNH}_4$ , and  $\text{TDN-N}$  (Table 4). The rate of soil  $\text{CH}_4\text{-C}$  emissions in arable soil



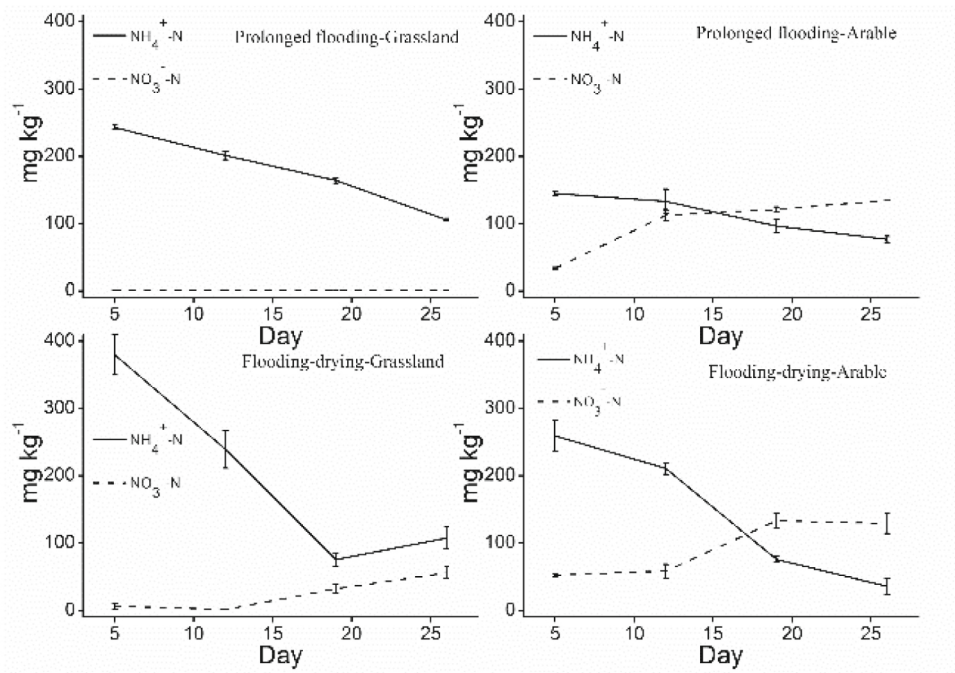
a



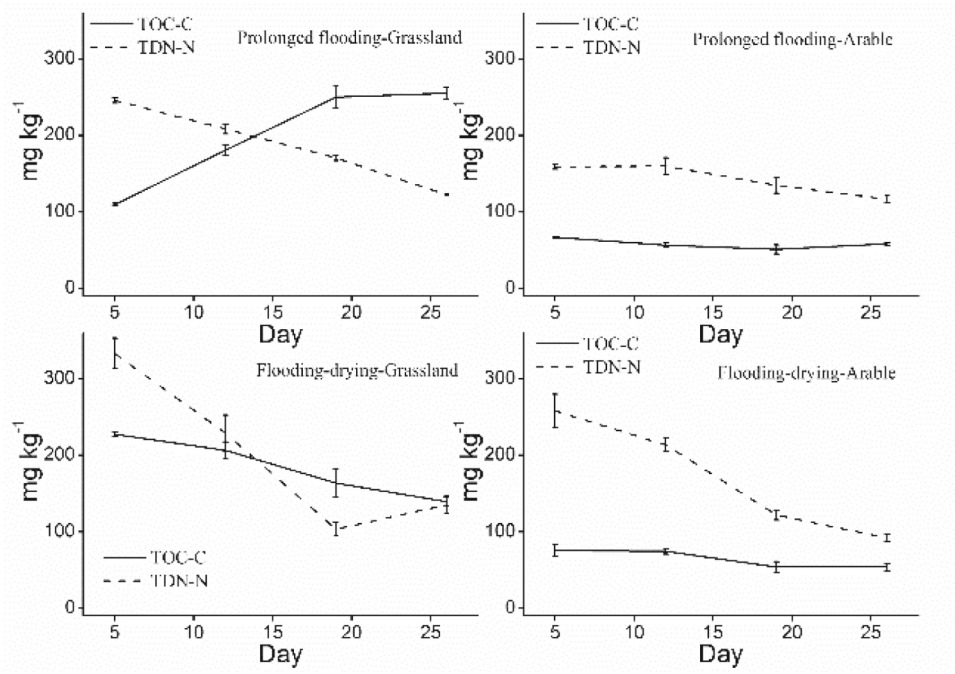
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**Fig. 4.** Soil  $\text{NH}_4^+\text{-N}$  and  $\text{NO}_3^-\text{-N}$  concentrations from soil pore water (a), ion-exchange resin membranes (b), soil extraction by KCl (c), and extracted total dissolved organic carbon (TOC-C) and total dissolved nitrogen (TDN-N) concentrations (d) over time under grassland and arable land use. The data values are means of three independent pot replicates and error bars represent the standard error of the means ( $n = 3$ ).





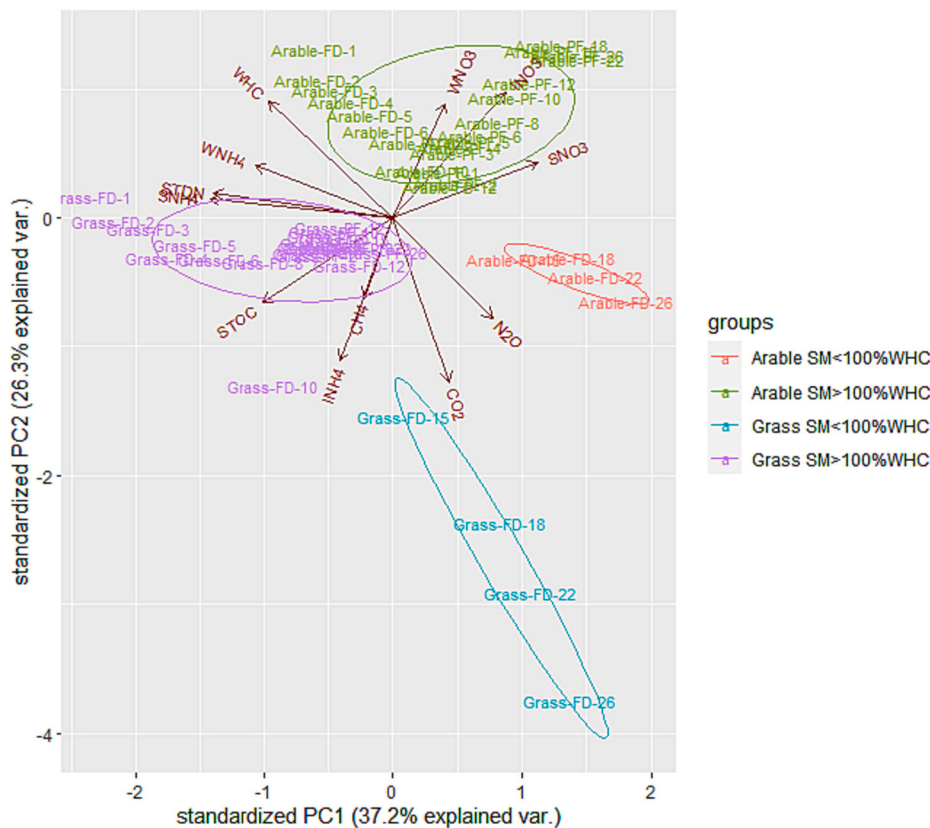
C



d

Fig. 4. (continued).





**Fig. 5.** Principal component analysis of variation in soil GHG emission under grassland and arable soil. N<sub>2</sub>O, rate of soil N<sub>2</sub>O-N emissions; CO<sub>2</sub>, rate of soil CO<sub>2</sub>-C emissions; CH<sub>4</sub>, rate of soil CH<sub>4</sub>-C emissions; WHC, soil water-holding capacity (%); INH<sub>4</sub>, soil NH<sub>4</sub><sup>+</sup>-N concentrations from ion-exchange resin membranes; INO<sub>3</sub>, soil NO<sub>3</sub><sup>-</sup>-N concentrations from ion-exchange resin membranes; WNH<sub>4</sub>, soil NH<sub>4</sub><sup>+</sup>-N concentrations from soil pore water; WNO<sub>3</sub>, soil NO<sub>3</sub><sup>-</sup>-N concentrations from soil pore water; SNH<sub>4</sub>, soil NH<sub>4</sub><sup>+</sup>-N concentrations from soil extraction by KCl; SNO<sub>3</sub>, soil NO<sub>3</sub><sup>-</sup>-N concentrations from soil extraction by KCl; STOC, soil total dissolved organic carbon-carbon; TDN, soil total dissolved nitrogen-nitrogen; SM, soil moisture.

**Table 4**

Correlation analysis of soil variation in soil GHG emission rates under grassland and arable soil when soil moisture >100% WHC and < 100% WHC. N<sub>2</sub>O-N, rate of soil N<sub>2</sub>O-N emissions; CO<sub>2</sub>-C, rate of soil CO<sub>2</sub>-C emissions; CH<sub>4</sub>-C, rate of soil CH<sub>4</sub>-C emissions; WHC, % of soil water-holding capacity; INH<sub>4</sub>, soil NH<sub>4</sub><sup>+</sup>-N concentrations from ion-exchange resin membranes; INO<sub>3</sub>, soil NO<sub>3</sub><sup>-</sup>-N concentrations from ion-exchange resin membranes; WNH<sub>4</sub>, soil NH<sub>4</sub><sup>+</sup>-N concentrations from soil pore water; WNO<sub>3</sub>, soil NO<sub>3</sub><sup>-</sup>-N concentrations from soil pore water; SNH<sub>4</sub>, soil NH<sub>4</sub><sup>+</sup>-N concentrations from soil extraction by KCl; SNO<sub>3</sub>, soil NO<sub>3</sub><sup>-</sup>-N concentrations from soil extraction by KCl; TOC-C, soil total dissolved organic carbon-carbon; TDN-N, soil total dissolved nitrogen-nitrogen; SM, soil moisture.

Soil type		SM > 100% WHC 1–12 (130)			SM < 100% WHC		
		N <sub>2</sub> O-N	CO <sub>2</sub> -C	CH <sub>4</sub> -C	N <sub>2</sub> O-N	CO <sub>2</sub> -C	CH <sub>4</sub> -C
Grassland	N <sub>2</sub> O-N	1.00	0.78**	0.59**	1.00	0.65	-0.73
	CO <sub>2</sub> -C	0.78**	1.00	0.81**	0.65	1.00	-0.69
	CH <sub>4</sub> -C	0.59**	0.81**	1.00	-0.73	-0.69	1.00
	WHC	-0.59**	-0.36	-0.37	-0.89*	-0.86	0.66
	INH <sub>4</sub>	0.01	0.13	0.10	0.97**	0.67	-0.86
	INO <sub>3</sub>	-0.39	-0.54**	-0.34	0.97**	0.68	-0.86
	WNH <sub>4</sub>	-0.09	0.22	0.07	-	-	-
	WNO <sub>3</sub>	0.95**	0.64**	0.32	-	-	-
	SNH <sub>4</sub>	-0.02	0.26	0.12	-0.52	-0.86	0.28
	SNO <sub>3</sub>	-0.07	0.27	0.07	0.89*	0.89*	-0.69
	TOC-C	0.22	0.43*	0.20	-0.86	-0.91*	0.65
	TDN-N	-0.05	0.21	0.09	-0.47	-0.84	0.23
	N <sub>2</sub> O-N	1.00	0.29	-0.17	1.00	0.15	-0.84
	CO <sub>2</sub> -C	0.29	1.00	-0.02	0.15	1.00	0.10
Arable	CH <sub>4</sub> -C	-0.17	-0.02	1.00	-0.84	0.10	1.00
	WHC	-0.34	-0.31	0.30	0.16	-0.93*	-0.36
	INH <sub>4</sub>	-0.14	0.43*	0.41	0.12	-0.96**	-0.33
	INO <sub>3</sub>	0.01	-0.61**	-0.32	0.77	0.68	-0.43
	WNH <sub>4</sub>	-0.31	0.26	0.52*	-	-	-
	WNO <sub>3</sub>	-0.18	0.31	0.07	-	-	-
	SNH <sub>4</sub>	-0.21	0.34	0.51*	0.32	-0.88*	-0.47
	SNO <sub>3</sub>	-0.09	0.00	-0.10	-0.53	0.74	0.60
	TOC-C	-0.09	0.39	0.38	0.50	-0.77	-0.58
	TDN-N	-0.24	0.34	0.53*	0.31	-0.89*	-0.46

\* Significant at  $P < 0.05$ .

\*\* Significant at  $P < 0.01$ .

under SM > 100% WHC showed a significant correlation with  $\text{WNH}_4$ ,  $\text{SNH}_4$ , and TDN-N (Table 4).

### 3.6. Variation partitioning of soil $\text{N}_2\text{O}$ -N emissions

WHC and  $\text{WNO}_3$  were selected to conduct the variation partitioning of  $\text{N}_2\text{O}$ -N emissions when SM > 100% WHC in grassland soil, since WHC and  $\text{WNO}_3$  showed a significant correlation with soil  $\text{N}_2\text{O}$ -N emissions (Table 4). WHC,  $\text{INH}_4$ ,  $\text{INO}_3$ , and  $\text{SNO}_3$  were selected to conduct the variation partitioning of  $\text{N}_2\text{O}$ -N emissions when SM < 100% WHC in grassland soil since WHC,  $\text{INH}_4$ ,  $\text{INO}_3$ , and  $\text{SNO}_3$  showed a significant correlation with soil  $\text{N}_2\text{O}$ -N emissions (Table 4).

$\text{WNO}_3$  of grassland soil had the highest contribution to  $\text{N}_2\text{O}$ -N emissions (58%) when SM > 100% WHC (Fig. 6a). The combined effect of  $\text{SNO}_3$  and  $\text{INH}_4$  had the highest contribution to  $\text{N}_2\text{O}$ -N emissions (71%) when SM < 100% WHC in grassland soil (Fig. 6b).

## 4. Discussion

### 4.1. The GHG emissions and $\text{NH}_4^+$ -N and $\text{NO}_3^-$ -N dynamics under prolonged flooding

The  $\text{SNH}_4$  of grassland and arable decreased and  $\text{SNO}_3$  of grassland and arable soils remained stable or increased slightly over time under PF, whereas the  $\text{WNH}_4$  of grassland and arable soils decreased and  $\text{WNO}_3$  of grassland and arable soils increased over time under PF (Fig. 4). This could be because small amounts of  $\text{NH}_4^+$ -N that were absorbed by soil particles (stable) were slowly released to soil pore water (available). The available  $\text{NH}_4^+$ -N and TDN-N seem to have been key substrates for nitrification ( $\text{NO}_3^-$ -N production) in our grassland and arable soil. The dynamics of  $\text{SNH}_4$ ,  $\text{SNO}_3$ , and  $\text{WNO}_3$  concentration combined with the lower  $\text{N}_2\text{O}$  emissions and GWP under PF (Fig. 3 and Table 3) showed that small amounts of  $\text{NO}_3^-$ -N were denitrified to  $\text{N}_2\text{O}$ -N and most of the  $\text{NO}_3^-$ -N was denitrified to  $\text{N}_2$  under the 110% WHC conditions of PF. This supports the finding of Wu et al. (2017) that excessive moisture condition can suppress  $\text{N}_2\text{O}$  emissions through its reduction to  $\text{N}_2$  during complete denitrification. Previous studies have reported low  $\text{N}_2\text{O}$  emissions from flooded fields (Shaaban et al., 2018; Song et al., 2021; Xu et al., 2022), which is commensurate with our finding that PF had lower  $\text{N}_2\text{O}$  emissions than FD. Flooding generally induce soil anaerobic conditions, consequently limiting mineralization of organic C and N, and ultimately resulting in less substrates for  $\text{N}_2\text{O}$  production via nitrification and denitrification (Li et al., 2022; Neubauer and Megonigal, 2021). These conditions are favourable for complete denitrification producing  $\text{N}_2$  rather than  $\text{N}_2\text{O}$ , and therefore low  $\text{N}_2\text{O}$  emissions (Cai et al., 2013; Mazza et al., 2018). Moreover, the decrease in  $\text{N}_2\text{O}$  emissions during continuously flooded soil can also rapidly deplete and entrap  $\text{NO}_3^-$  within the micropores, possibly limiting further denitrification to produce  $\text{N}_2\text{O}$  and  $\text{N}_2$  gases (McNicol and Silver, 2014).

Arable soil had a higher rate of  $\text{N}_2\text{O}$ -N emissions than grassland soil under PF (Fig. 3), and the arable soil had significantly higher total  $\text{N}_2\text{O}$ -N emissions and GWP than grassland soil under PF (Tables 2 and 3) with a higher  $\text{INO}_3$  concentration (Fig. 4b) and decomposition of TDN-N (Fig. 4d). This could be because arable soil had a smaller pH compared with grassland soil and arable soil had more nitrate compared with grassland soil from the beginning (Table 1). Čuhel et al. (2010) found that the relative importance of  $\text{N}_2\text{O}$  as a product of denitrification is higher at low pH. The functionality of the *nosZ* gene for synthesizing  $\text{N}_2\text{O}$ -reducing enzyme ( $\text{N}_2\text{O}$  reductase) is limited under low pH, leading to higher  $\text{N}_2\text{O}$  emissions (Shaaban et al., 2018). High soil pH values also typically shift the denitrification end product ratio towards  $\text{N}_2$  instead of  $\text{N}_2\text{O}$  (Simek et al., 2002) and lower  $\text{N}_2\text{O}$  formation during nitrification (Mørkved et al., 2007). Bell et al. (2015) also found that the annual emissions factor for a Scottish arable soil was 3–5 times higher than grassland sites elsewhere in the UK, which supports our findings of greater  $\text{N}_2\text{O}$ -N emissions from arable than grassland soil.

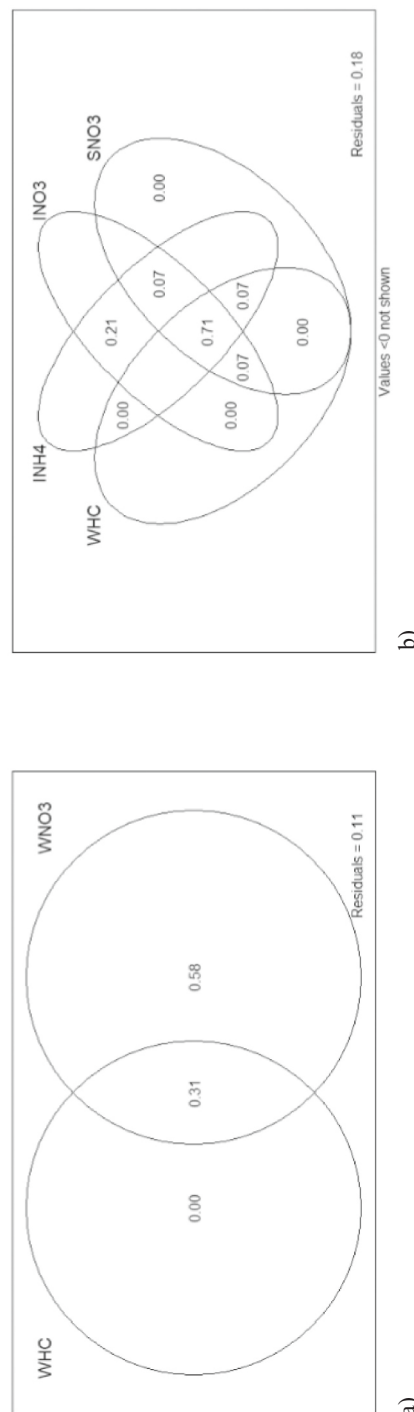


Fig. 6. Variation partitioning of  $\text{N}_2\text{O}$ -N emissions for soil moisture exceeding 100% water-holding capacity (a) and soil moisture lower than 100% water-holding capacity (b) in grassland soil. WHC, soil water-holding capacity (%);  $\text{INH}_4$ , soil  $\text{NH}_4^+$ -N concentrations from ion-exchange resin membranes;  $\text{INO}_3$ , soil  $\text{NO}_3^-$ -N concentrations from ion-exchange resin membranes;  $\text{WNO}_3$ , soil  $\text{NO}_3^-$ -N concentrations from soil pore water;  $\text{SNO}_3$ , soil  $\text{NO}_3^-$ -N concentrations from soil extraction by KCl.

The CO<sub>2</sub> and CH<sub>4</sub> emissions were low in both grassland and arable soil under PF (Fig. 3 and Table 2). This is likely to be because the high-water content limited the decomposition of soil organic matter and dissolved the CH<sub>4</sub>. Devivre and Horwath (2000) and Khalid et al. (2019) showed that mineralization of SOC was restrained under waterlogged conditions by restricting microbial growth and activities, and thereby impeded CO<sub>2</sub> production. The unexpected low CH<sub>4</sub> emissions seem to have resulted from low diffusion into air under flooding/saturated soils conditions. Moreover, under saturated conditions, interaction of methanogens and nitrate reducers, which utilize acetate and H<sub>2</sub> more efficiently than methanogens may have further reduced potential CH<sub>4</sub> fluxes (Conrad, 2002) for arable soil. Another possible explanation for the low CH<sub>4</sub> emission might be the use of ammonium sulphate as fertilizer, where ammonium competitively inhibits methanography (Ullah et al., 2008) in rice fields (Linquist et al., 2012).

#### 4.2. The GHG emissions and NH<sub>4</sub><sup>+</sup>-N and NO<sub>3</sub><sup>-</sup>-N dynamic under flooding–drying

The WNH<sub>4</sub> of grassland and arable soils decreased and WNO<sub>3</sub> of grassland and arable soils increased over time under FD, while the SNH<sub>4</sub> of grassland and arable soils strongly decreased and SNO<sub>3</sub> of grassland and arable increased slightly over time under FD (Fig. 4c). When SM > 100% WHC, lower N<sub>2</sub>O emissions were detected under FD in grassland and arable soils (Fig. 3). This is similar to the findings for PD described in Section 4.1. When SM < 100% WHC, N<sub>2</sub>O-N emissions were high under FD in grassland soil and N<sub>2</sub>O-N emissions increased as SM decreased (Fig. 3). This was mainly because nitrification-denitrification and denitrification increased over time in grassland soil. Freibauer et al. (2004) and Goldberg et al. (2010) also found that degradation of soil organic matter as a result of drainage and cultivation will stimulate net N mineralization and N transformations via nitrification and denitrification, leading to N<sub>2</sub>O production. The soil pore structure is important for GHG production and diffusion at the soil-air interface. At the capillary fringe above the water level, the soil is characterized by having close to sub-saturated SM and mixed aerobic and anaerobic conditions, promoting the environmental conditions favourable for both N<sub>2</sub>O production via denitrification and NO<sub>3</sub><sup>-</sup> reduction via DNRA (Megonigal et al., 2003). Several researchers (Congreves et al., 2019; Miller et al., 2020) have found that the products of N<sub>2</sub>O by denitrification increase with decreasing water-filled pore space from 90% to 60%. Moreover, the N<sub>2</sub>O products peaked at optimum water filled pore space of 65%–70% (Pärn et al., 2018; van Lent et al., 2015; Werner et al., 2007). Denitrification becomes more prevalent at higher water contents, leading to maximum emissions at around 80% WHC (Shepherd, 2009). In agricultural soils, the total N<sub>2</sub>O emissions in a season are strongly related to the episodes of large N<sub>2</sub>O pulses observed after irrigation and rainfall events, which are primarily derived from denitrification (Trost et al., 2013). This is consistent with our finding that N<sub>2</sub>O-N emissions increased with the decrease of SM towards the end of the experiment (SM was still higher than 60% WHC).

The TDN-N and TOC-C of grassland decreased with time under FD (Fig. 4d). However, TDN-N decreased but TOC-C remained stable with time in arable soils under FD (Fig. 4d). Arable soil also had a lower N<sub>2</sub>O-N emissions rate than grassland soil when SM was below 80% WHC under FD (Fig. 3) and the grassland soil had higher total N<sub>2</sub>O-N emissions and GWP than arable soil under FD (Tables 3 and 4). This could be because soil N, C, and microbes in arable soil limited the nitrification and denitrification, and therefore lowered the production of N<sub>2</sub>O-N under SM < 100% WHC. The C-substrate availability and N<sub>2</sub>O emissions from denitrification and nitrification are always positively related to each other (Li et al., 2005; Wan et al., 2009) because nitrification is primarily an autotrophic process and heterotrophic nitrification only accounts for 20% or less under low-pH conditions (Liu et al., 2015). Gütlein et al. (2018) also found that N<sub>2</sub>O emissions correlated positively with SM and total soil nitrogen content. Short and temporary drying-

rewetting frequency enhanced denitrifier activity by availing physically protected organic matter (Fierer and Schimel, 2002). The N<sub>2</sub>O derived from soil organic matter decomposition dominates overall fluxes (Maljanen et al., 2010). Abbasi et al. (2011) also found that the process of denitrification and production of N<sub>2</sub>O was smaller in arable soil deficient in organic matter compared with grassland soil because of the lower availability of organic C. Another reason is that grassland soil has a higher pH (Table 1), which then leads to higher N<sub>2</sub>O-N emissions when SM < 100% WHC (Fig. 3 and Table 2). Fan et al. (2018) reported significantly higher N<sub>2</sub>O emissions rates from three alkaline soils (pH 7.6–8.2) as compared with an acidic soil with a pH of 5.6. Oxidation of NH<sub>4</sub><sup>+</sup> is completely inhibited at pH 5 but increases as pH increases (Wang et al., 2018) through a pH-driven shift in the microbial community structure and/or microbial activities (Ottosen et al., 2009).

The soil TC and TOC were mainly anaerobically digested to CH<sub>4</sub> before 15 days (especially when SM was between 110% and 80%) and aerobically digested to CO<sub>2</sub> after 15 days in FD grassland soil when SM was lower than 80% WHC (Fig. 3). Methane production is expected to mainly occur below the groundwater table (Segers, 1998). Anaerobic conditions during the flooding stage of the experiment promoted methanogenic activities while suppressing methanotrophic activities, leading to CH<sub>4</sub> production (Shaaban et al., 2022). Drainage will limit the production of CH<sub>4</sub> because the highest oxidation potentials are found near the oxic–anoxic interface (Petersen et al., 2012). However, draining can also increase the potential for CH<sub>4</sub> oxidation during passage through the unsaturated zone to the atmosphere (Petersen et al., 2012). In general, CO<sub>2</sub> fluxes are dominant at drained sites (Kandel et al., 2018) and show a decreasing trend following an increase of soil water content (Smith et al., 2003; McNicol and Silver, 2014). This is because of a lower O<sub>2</sub> availability and the consequent inhibition of aerobic respiration when a large proportion of pores are saturated under flooding (Mazza et al., 2018). Khalid et al. (2019) showed that CO<sub>2</sub> emissions were significantly larger in a flooded soil converted to wet soil treatment compared with continuously flooded and wet soil treatments, which they suggested was due to increased mineralization and C contents. In our study, the CH<sub>4</sub> emissions after 100% WHC (Fig. 3) may be due to the gas entrapment into microaggregates and a delayed release (either as emissions or leaching) (Mazza et al., 2018). The CH<sub>4</sub> emissions increased during the second flooding event (FD) relative to the first (PF) (Fig. 3), which could relate to some mechanism of adaptation by microbial communities (Lagomarsino et al., 2016).

The grassland soil had a higher rate of CO<sub>2</sub>-C and CH<sub>4</sub>-C emissions than the arable soil under FD (Fig. 3). The grassland soil under FD had significantly higher total CO<sub>2</sub>-C and CH<sub>4</sub>-C emissions and GWP than arable soil (Tables 3 and 4). The organic matter and TN contents were higher in grassland soils than in arable soils (Table 1), leading to higher potential for mineralization after disturbance (drainage) (Eickenscheidt et al., 2014). Volpi et al. (2017) found that CO<sub>2</sub> emissions are positively linked to SOC content and denitrification activity, which also increases CO<sub>2</sub> efflux (Groffman and Crawford, 2003). Therefore, grassland soil under FD had significantly higher total CO<sub>2</sub>-C and CH<sub>4</sub>-C emissions and GWP than arable soil under FD. Thomson et al. (2010) detected significantly higher respiration rates in dried and rewetted microcosms. When the SM content decreased from flooded to 60% WFPS, CH<sub>4</sub> emissions decreased, but N<sub>2</sub>O and CO<sub>2</sub> emissions substantially increased because the flooded soil released N and C and triggered C and N cycling (Shaaban et al., 2022). We similarly found higher GHG emissions and GWP under FD than PF in grassland soil (Tables 2 and 3).

#### 4.3. The influence of different types on soil NH<sub>4</sub><sup>+</sup>-N and NO<sub>3</sub><sup>-</sup>-N contents on GHG emissions

Grassland and arable soil sample of N<sub>2</sub>O-N emissions were clearly separate between SM > 100% WHC and SM < 100% WHC (Fig. 5). Senbayram et al. (2009) found that the high NO<sub>3</sub><sup>-</sup> availability as a result of nitrification, together with labile C under anaerobic soil conditions,



serves as a driving force for  $\text{N}_2\text{O}$  emissions. We also found that the  $\text{N}_2\text{O}$ -N and  $\text{CO}_2$ -C emissions were mainly controlled by soil available  $\text{NO}_3^-$ -N when SM exceeded 100% WHC, whereas  $\text{N}_2\text{O}$ -N emissions were controlled by soil stable  $\text{NH}_4^+$ -N and  $\text{NO}_3^-$ -N when SM was lower than 100% WHC in the grassland soil (Fig. 6).

## 5. Conclusion

Changes in soil moisture as influenced by PF and FD as proxies of climate change significantly affected the GHG flux of arable and grassland soils differently. Higher  $\text{N}_2\text{O}$ -N,  $\text{CO}_2$ -C, and  $\text{CH}_4$ -C emissions and large GWP were found in grassland soil under FD, and these fluxes were mainly controlled by soil  $\text{NH}_4^+$ -N,  $\text{NO}_3^-$ -N, and TOC concentrations when soil moisture was lower than 100% WHC. Arable soils exhibited similar GWP both under PF and FD. Compared to grassland, arable soil had higher GWP than grassland soils under PF, which switched significantly under FD where the GWP of grassland soils was order of magnitude higher than arable soils. Future climate change will most likely result in intense precipitation resulting in soils saturation followed by drying and under such conditions, temperate grasslands will emit more GHG than arable soils. This suggests that land use conversion and/or rotations from arable to grassland soils for improving soil health or carbon sequestration shall consider the changing source-strength of GHG fluxes in the overall carbon and GHG budgets under national GHG and carbon sequestration inventories and the net zero GHG ambitions. Repeating this experiment in the field with undisturbed soil samples will carry out for further study.

## Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

## Data availability

Data will be made available on request.

## Acknowledgements

This research was supported by the UKRI-NERC under the joint project “Signals in the Soils”. Large Area Distributed Real Time Soil (DIRTS) Monitoring and BBSRC (BB/R021716/1). The authors would like to thank Gianni Micucci (PhD student) for support in field campaigns, James Gore (Research Technician) for laboratory training, and Andrea Rabbai, Manon Rumeau, Thanawan Buacharoen, Jessica Chadwick (PhD students) and Johanna Pihlbald (Research Fellow) for help in laboratory work and instrumental analysis.

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