

# Impact of cloud process in the mixing state and microphysical properties of soot particles

Fu, Yuzhen; Peng, Xiacong; Sun, Wei; Hu, Xiaodong; Wang, Dian; Yang, Yuxiang; Guo, Ziyong; Wang, Yuanyuan; Zhang, Guohua; Zhu, Jianxi; Ou, Jie; Shi, Zongbo; Wang, Xinming; Bi, Xinhui

DOI:  
[10.1029/2022JD037169](https://doi.org/10.1029/2022JD037169)

License:  
None: All rights reserved

*Document Version*  
Peer reviewed version

*Citation for published version (Harvard):*  
Fu, Y, Peng, X, Sun, W, Hu, X, Wang, D, Yang, Y, Guo, Z, Wang, Y, Zhang, G, Zhu, J, Ou, J, Shi, Z, Wang, X & Bi, X 2022, 'Impact of cloud process in the mixing state and microphysical properties of soot particles: implications in light absorption enhancement', *Journal of Geophysical Research: Atmospheres*, vol. 127, no. 21, e2022JD037169. <https://doi.org/10.1029/2022JD037169>

[Link to publication on Research at Birmingham portal](#)

## **Publisher Rights Statement:**

An edited version of this paper was published by AGU. Copyright (year) American Geophysical Union.

Fu, Y., Peng, X., Sun, W., Hu, X., Wang, D., Yang, Y., et al. (2022). Impact of cloud process in the mixing state and microphysical properties of soot particles: Implications in Light absorption enhancement. *Journal of Geophysical Research: Atmospheres*, 127, e2022JD037169. DOI: 10.1029/2022JD037169. To view the published open abstract, go to: <https://doi.org/10.1029/2022JD037169>.

## **General rights**

Unless a licence is specified above, all rights (including copyright and moral rights) in this document are retained by the authors and/or the copyright holders. The express permission of the copyright holder must be obtained for any use of this material other than for purposes permitted by law.

- Users may freely distribute the URL that is used to identify this publication.
- Users may download and/or print one copy of the publication from the University of Birmingham research portal for the purpose of private study or non-commercial research.
- User may use extracts from the document in line with the concept of 'fair dealing' under the Copyright, Designs and Patents Act 1988 (?)
- Users may not further distribute the material nor use it for the purposes of commercial gain.

Where a licence is displayed above, please note the terms and conditions of the licence govern your use of this document.

When citing, please reference the published version.

## **Take down policy**

While the University of Birmingham exercises care and attention in making items available there are rare occasions when an item has been uploaded in error or has been deemed to be commercially or otherwise sensitive.

If you believe that this is the case for this document, please contact [UBIRA@lists.bham.ac.uk](mailto:UBIRA@lists.bham.ac.uk) providing details and we will remove access to the work immediately and investigate.

1 **Impact of Cloud Process in the Mixing State and Microphysical**  
2 **Properties of Soot Particles: Implications in Light Absorption**  
3 **Enhancement**

4 Yuzhen Fu<sup>1,2</sup>, Xiacong Peng<sup>1,2,3</sup>, Wei Sun<sup>1,2,3</sup>, Xiaodong Hu<sup>1,2,3</sup>, Dian Wang<sup>1,2,3</sup>,  
5 Yuxiang Yang<sup>1,2</sup>, Ziyong Guo<sup>1,2,3</sup>, Yuanyuan Wang<sup>4</sup>, Guohua Zhang<sup>1,2,5,\*</sup>, Jianxi Zhu<sup>2,6</sup>,  
6 Jie Ou<sup>7</sup>, Zongbo Shi<sup>8</sup>, Xinming Wang<sup>1,2,5</sup>, Xinhui Bi<sup>1,2,5,\*</sup>

7 <sup>1</sup> State Key Laboratory of Organic Geochemistry and Guangdong Key Laboratory of  
8 Environmental Protection and Resources Utilization, Guangzhou Institute of  
9 Geochemistry, Chinese Academy of Sciences (CAS), Guangzhou, PR China

10 <sup>2</sup> CAS Center for Excellence in Deep Earth Science, Guangzhou, PR China

11 <sup>3</sup> University of Chinese Academy of Sciences, Beijing, PR China

12 <sup>4</sup> Department of Atmospheric Science, School of Earth Sciences, Zhejiang University,  
13 Hangzhou, PR China

14 <sup>5</sup> Guangdong-Hong Kong-Macao Joint Laboratory for Environmental Pollution and  
15 Control, Guangzhou, PR China

16 <sup>6</sup> CAS Key Laboratory of Mineralogy and Metallogeny and Guangdong Provincial Key  
17 Laboratory of Mineral Physics and Materials, Guangzhou Institute of Geochemistry,  
18 Chinese Academy of Sciences, Guangzhou, PR China

19 <sup>7</sup> Shaoguan Environmental Monitoring Center, Shaoguan, PR China

20 <sup>8</sup> School of Geography, Earth and Environmental Sciences, University of Birmingham,  
21 Birmingham, UK

22 \*Correspondence author: Guohua Zhang (zhanggh@gig.ac.cn) and Xinhui Bi  
23 (bixh@gig.ac.cn)

24 **Key Points:**

- 25 ● Cloud processing contributes to the increase of organic coating on soot aggregates,  
26 especially for those with fully embedded structure.
- 27 ● The sulfate coating causes more compaction of soot aggregates than the organic  
28 coating upon cloud processing.
- 29 ● Changes in the microphysical properties of soot particles driven by one cloud event  
30 increase their absorption enhancement by 1.29 times.

31 **Abstract**

32 The radiative forcing of soot is dependent on the morphology, mixing state and structure.  
33 Cloud processing has been predicted to affect their mixing properties but little is known  
34 about the resulting light absorption properties. We collected ambient particles in the  
35 pre-cloud period, the cloud residues and interstitials in the in-cloud period at Mt.  
36 Tianjing (southern China). The morphology parameters of soot aggregates with varying  
37 mixing materials (sulfate (S) and organics (OM)) and mixing structures were  
38 investigated by a transmission electron microscope, and their absorption cross section  
39 were calculated based on discrete dipole approximation. We found that the number  
40 contribution of soot-S decreased from 45% in the pre-cloud period to 32% in the in-  
41 cloud period, and that of soot-OM increased from 44% to 60%. Moreover, the number  
42 proportion of soot-OM with fully embedded structure increased remarkably in the in-  
43 cloud period (29%), compared with that in the pre-cloud period (3%). In addition, the  
44 soot-S aggregates became denser after in-cloud aqueous process. However, for soot-  
45 OM aggregates, the morphology remained relatively constant. The distinctly different  
46 change of soot-S and soot-OM in morphology highlights the chemically resolved  
47 reconstruction of soot morphology. Theoretical calculation further shows that the  
48 changes of soot particles in the mixing state and morphological characteristics by the  
49 cloud process resulted in the light absorption enhancement increase from 1.57 to 2.01.  
50 This study highlights that the evolution of microphysical properties upon cloud  
51 processing should also be considered in climate models to more accurately evaluate the  
52 impacts of soot particles.

53 **Plain Language Summary**

54 While soot particles are the main component of aerosols that produce positive radiative  
55 forcing in clouds, the impact of in-cloud processes on key factors (mixing state and  
56 microphysical property) for evaluating the optical properties of soot particles is unclear.  
57 Here, we report the detailed information on several parameters required to calculate the  
58 optical properties of soot particles, throughout the cloud events. And we found that  
59 changes in the mixing state, mixing structure, and morphology of soot particles under  
60 the influence of in-cloud processes lead to a significant light absorption enhancement.  
61 The results also highlight the role of in-cloud aqueous formation of secondary  
62 compositions in reshaping the soot particles, which has substantial implications for the  
63 climate impact of soot particles. Given that ~70% of the Earth's surface is covered by  
64 clouds, taking the mixing state and microphysical properties of soot particles into  
65 account in climate models may help constrain the contribution of soot particles to global  
66 radiative forcing.

## 67 **1 Introduction**

68 Soot, also known as black carbon (BC) or elemental carbon (EC) [Buseck et al.,  
69 2014], is deeply concerning because it has a multitude of effects on the global  
70 atmosphere and Earth's surface. It is the second largest contribution to warm the  
71 atmosphere after carbon dioxide [Jacobson, 2001]. Soot particles in the atmosphere  
72 exhibit complexity in size, composition, morphology, and mixing structure, which  
73 results in the high uncertainties in climate models when predicting their optical  
74 properties [Lack and Cappa, 2010; Liu et al., 2017a; Zhang and Mao, 2020] and CCN  
75 activity [Li et al., 2018; Liu et al., 2013].

76 The chemical and physical properties of soot particles are dynamic and vary with  
77 aging processes during atmospheric transport. They are sensitive to the formation of  
78 secondary components and atmospheric conditions, leading to complex and ever-  
79 changing mixing state and mixing structure of soot particles [Liu et al., 2017b;  
80 Zelenyuk et al., 2010; Zhang et al., 2018]. Fresh soot particles emitted from biomass  
81 burning and fossil fuel combustion are hydrophobic with bare-like structure, which  
82 generally shows the chain-like structure composed of several nanospheres with onion-  
83 like and graphene-like layers of carbon [Buseck et al., 2014; Li and Shao, 2009]. As  
84 the degree of aging increases, the soot aggregates are wrapped thicker and become more  
85 compact, resulting in more partly coated and/or fully embedded soot particles [China  
86 et al., 2014; China et al., 2015b; Wang et al., 2017; Xu et al., 2020], and higher  
87 absorption enhancement [Wu et al., 2018]. The mixing structure and morphology  
88 parameters of soot particles are key factors to calculate their optical properties [Adachi

89 et al., 2010; Cappa et al., 2012; Yu et al., 2019]. From the bare-like structure to being  
90 gradually coated by other components, the light absorption of soot particles will be  
91 significantly enhanced by factors ranging from 1 to 3.5 [Wu et al., 2018]. And, the  
92 optical properties of the bare-like soot aggregates with diverse fractal degrees are also  
93 different [Radney et al., 2014; Wang et al., 2021b]. Additionally, soot aggregates coated  
94 by various materials may redistribute after liquid-liquid phase separation (LLPS) under  
95 high relative humidity, which could lead to an inverted core-shell structure (soot  
96 aggregates are distributed in the outer layer of the two-phase particle) [Brunamonti et  
97 al., 2015]. Consequently, the core-shell model (concentric spheres composed of soot  
98 nuclei and coating) assumed in current climate models results in higher uncertainty of  
99 light absorption compared to observations, either overestimation [Cappa et al., 2012;  
100 Wu et al., 2018] or underestimation [Kelesidis et al., 2022]. Therefore, an accurate  
101 understanding of the mixing structure and morphology parameters of soot particles is  
102 crucial for the climate model to evaluate their radiative forcing.

103 Previous studies have shown that cloud could modify the mixing state and  
104 morphology of atmospheric particles [China et al., 2015a; Liu et al., 2018], through  
105 collision/coalescence and Brownian capture as well as aqueous reactions. The  
106 reconstruction of soot aggregates has been found in high humidity environments such  
107 as fog and cloud events [Bhandari et al., 2019; China et al., 2015a], which is affected  
108 by the mixing state of soot particles [Khalizov et al., 2009; Qiu et al., 2012; Zhang et  
109 al., 2008]. High humidity condition during cloud process can promote the formation of  
110 various species including inorganic (such as nitrate, sulfate, and ammonium) and

111 organic substances (such as oxalic acid, organosulfate) [Kim et al., 2019; Schneider et  
112 al., 2017; Zhang et al., 2017b]. While the physical and chemical properties of organic  
113 matter have been observed to bring about varying mixing structure and structural  
114 reorganization of soot particles [Chen et al., 2020; Chen et al., 2016; Sharma et al.,  
115 2018], how the cloud processing driven changes in the mixing state of soot particles  
116 affects their microphysical properties is poorly understood yet.

117 In order to study the effects of cloud processes on mixing state and microphysical  
118 properties of soot particles, samples during pre-cloud and in-cloud periods at a remote  
119 high-altitude site in southern China were collected and analyzed. We investigated the  
120 changes of morphology parameters (i.e., fractal dimension, aspect ratio, roundness, and  
121 size) of soot aggregates with different mixing states and mixing structures from pre-  
122 cloud to in-cloud periods. The possible influencing factors such as different coating  
123 materials and mixing structures were discussed. Finally, the effect of cloud processing  
124 on the light absorption of soot particles was evaluated.

## 125 **2 Methods**

### 126 **2.1 Sampling and Instrumentation**

127 A field observation focused on the cloud events was conducted from May 8 to June  
128 3, 2018 at an atmospheric observation station (112°53'56" E, 24°41'56" N; 1690 m  
129 above sea level) on the top of Mt. Tianjing. There is a pristine forest covering 273 km<sup>2</sup>  
130 and few anthropogenic emissions around the sampling site. Cloud events are prevalent  
131 due to the influence of warm and humid air masses from the South China Sea [Lin et  
132 al., 2019]. The cloud events investigated in this study were mainly affected by local

133 biogenic sources and long-range transport from the southwest air masses (**Fig. S1**). Soot  
134 particles mainly come from combustion activities in Southeast Asia and industrial  
135 emissions in the urban agglomeration of southern China [Zhang et al., 2017a].

136 A ground-based counterflow virtual impactor (GCVI, model 1205, Brechtel Mfg.  
137 Inc., USA) was used to introduce cloud droplets larger than 7.5  $\mu\text{m}$  into the subsequent  
138 system [Shingler et al., 2012]. During cloud processes, cloud droplets and unactivated  
139 particles are introduced into the inlet of the GCVI, a wind tunnel, which can accelerate  
140 droplets or particles to 100  $\text{m s}^{-1}$ . The flow rate of the filtered and heated add-flow can  
141 be adjusted in the range of 16-23  $\text{L min}^{-1}$  to ensure the cut size of 7.5  $\mu\text{m}$  with the  
142 transmission efficiency of 50%. The flow rate of sample-flow is kept at a constant 15  $\text{L}$   
143  $\text{min}^{-1}$  for multiple instruments downstream. The cloud droplets went through an  
144 evaporation chamber (40  $^{\circ}\text{C}$ ) to remove the associated water, leaving the cloud residual  
145 (RES) particles for the follow-up particle sampling or on-line analysis. The add-flow  
146 with high temperature indirectly controls the sample-flow temperature of evaporation  
147 chamber. Some semi-volatile organic compounds will be lost under heating, and the  
148 loss at 40  $^{\circ}\text{C}$  may be less than 1% [Cao et al., 2018]. In addition, a  $\text{PM}_{2.5}$  cyclone inlet  
149 followed by a silica gel diffusion dryer was used for the collection of inactivated  
150 interstitial (INT) particles during the cloud events and the ambient particles during free-  
151 cloud periods.

152 A DKL-2 sampler (Genstar Electronic Technology Co., Ltd., China) employing a  
153 single-stage cascade impactor with a 0.5 mm diameter jet nozzle was used to collect  
154 individual particles on copper grids coated by carbon film (300 mesh, Beijing



155 Zhongjingkeyi Technology Co., Ltd., China) with the airflow of  $1 \text{ L min}^{-1}$ . The effective  
156 size  $d_{50}$  is  $0.5 \mu\text{m}$  for particles with a density of  $2 \text{ g cm}^{-3}$ . The DKL-2 sampler was  
157 described in detail by Li et al. [2011]. The sampling duration varied from 5 to 10 min  
158 based on the number concentration of particles. After collecting, samples were stored  
159 in a sealed plastic box and placed in a desiccator (RH (relative humidity)  $< 30\%$ ) at  $25 \text{ }^\circ\text{C}$   
160 for later analysis.

161 A transmission electron microscope (TEM, FEI Talos F200S, USA) operated at  
162  $200 \text{ kV}$  was used to observe the morphology of particles on the copper grids, which  
163 was coupled with an energy-dispersive X-ray spectrometer (EDS) to obtain the  
164 elemental composition including carbon and heavier elements ( $Z \geq 6$ ). The distribution  
165 of particles on the copper grids is not uniform. The diameter of particles distributed in  
166 the center is larger than that in the edge. Therefore, in order to ensure that particles  
167 analyzed are representative within the entire particle size range, 3-4 areas were selected  
168 from the center to the edge on each copper grid. The electron beam of TEM was focused  
169 on the particles when the soot aggregates were coated. The components with weak  
170 thermal stability were lost under the electron beam, but the refractory components  
171 remain unchanged, so the morphology of soot aggregates could be clearly captured (**Fig.**  
172 **S2**). An image analysis software, ImageJ (<https://imagej.nih.gov/>), was used to obtain  
173 the morphology parameters of all particles by analyzing TEM images.

174 During the observation period, the mass concentrations of  $\text{PM}_{2.5}$ ,  $\text{O}_3$ ,  $\text{SO}_2$  and  $\text{NO}_x$   
175 were monitored in real time by tapered element oscillating microbalance (TEOM 1405,  
176 Thermo Fisher Scientific Inc., MA, US),  $\text{O}_3$  analyzer (Model 49i, Thermo Fischer

177 Scientific Inc., USA), pulsed UV fluorescence (Model 43i-TLE, Thermo Fischer  
178 Scientific Inc., USA), and NO-NO<sub>2</sub>-NO<sub>x</sub> analyzer (Model 42i-TL, Thermo Fischer  
179 Scientific Inc., USA), with details described elsewhere [Gong et al., 2018].

## 180 **2.2 Division of cloud periods and meteorological characteristics**

181 This field observation spanned two complete non-precipitation cloud processes,  
182 which were named as cloud events #1 and #2. Both cloud events formed at night and  
183 dissipated at noon of the next day. During the cloud events, prevailing wind direction  
184 was from the southwest, and wind speed was in the range of 4.1-9.4 m s<sup>-1</sup>. The mass  
185 concentration of PM<sub>2.5</sub> was 2-23 μg m<sup>-3</sup> during cloud event #1 and 4-47 μg m<sup>-3</sup> during  
186 cloud event #2. Two periods were defined according to the visibility (*Vis*) and RH [Lin  
187 et al., 2019]: pre-cloud period (*Vis* > 3 km or RH < 95%) and in-cloud period (*Vis* ≤ 3  
188 km and RH ≥ 95%) (**Fig. S3**). Due to rapid development of formation and dissipation  
189 of cloud, INT particles were only collected in the stability period with low visibility  
190 (*Vis* < 0.1 km) and the mass concentration of PM<sub>2.5</sub> (< 5 μg m<sup>-3</sup>) [Yang et al., 2021].  
191 The average visibility and RH during the stability period of cloud processes were 38 ±  
192 13 m and 100 ± 0.2 %, which were 600 ± 890 m and 99 ± 0.7 % in the non-stability  
193 (formation and dissipation) period respectively. In the following text, particles in the  
194 in-cloud period refer to all the RES particles during cloud processes, except as  
195 specifically noted. Overall, 4569 particles were analyzed by TEM/EDS including 1188  
196 particles in the pre-cloud period and 3381 particles in the in-cloud period. Standard  
197 errors for the number fraction of particles were estimated assuming Poisson distribution  
198 [Pratt et al., 2010].

199 During two periods of cloud event #1, there was little change for the  
200 concentrations of SO<sub>2</sub> and O<sub>3</sub> in the atmosphere. Differently, they changed dramatically  
201 during cloud event #2. The average concentrations of SO<sub>2</sub> and O<sub>3</sub> reduced from 3.1 µg  
202 m<sup>-3</sup> and 188.2 µg m<sup>-3</sup> in the pre-cloud period to 1.5 µg m<sup>-3</sup> and 82.1 µg m<sup>-3</sup> in the in-  
203 cloud period, respectively. Due to the cloud scavenging of particulate matter, the  
204 average concentration of PM<sub>2.5</sub> was significantly reduced from 20.4 µg m<sup>-3</sup> in the pre-  
205 cloud period and to 5.1 µg m<sup>-3</sup> during the in-cloud period for cloud event #1 and from  
206 32.7 µg m<sup>-3</sup> to 12.3 µg m<sup>-3</sup> for cloud event #2. The average concentration of NO<sub>x</sub> was  
207 higher during cloud event #2 (10.5 µg m<sup>-3</sup>) than cloud event #1 (3.8 µg m<sup>-3</sup>).

### 208 **2.3 The calculation of morphology parameters of soot aggregates**

209 Fractal dimension ( $D_f$ ) is a morphology parameter used to characterize the mass-  
210 based fractal properties, which can quantify the fractal degree of soot aggregates  
211 [Samson et al., 1987]. Brasil et al. [1999] linked the structural properties between  
212 computer-simulated three-dimensional aggregates and their two-dimensional  
213 projection images, which can be used to infer morphology parameters (such as  $D_f$ ) of  
214 three-dimensional aggregates from analysis of two-dimensional electron microscopy  
215 images. This method has been frequently used in the calculation of soot aggregates  
216 morphology parameters [Wang et al., 2017; Wentzel et al., 2003]. The  $D_f$  follows the  
217 statistical scaling law:

$$N = k_g \left( \frac{2R_g}{d_p} \right)^{D_f} \quad (1)$$

218 where  $N$  and  $d_p$  represent the number and average diameter of monomers within a  
 219 certain soot aggregate, respectively.  $R_g$  refers to the gyration radius of the soot aggregate,  
 220 and  $k_g$  is the fractal pre-factor. The calculation method of  $D_f$  is described in detail in the  
 221 supporting information (Text S1). Aspect ratio (AR) and roundness (RN) are also  
 222 calculated to quantify the morphology of soot aggregates, which can be obtained by  
 223 equations (2) and (3):

$$\text{AR} = \frac{L_{max}}{W_{max}} \quad (2)$$

$$\text{RN} = \frac{4A_a}{(\pi L_{max}^2)} \quad (3)$$

224 where  $L_{max}$  and  $W_{max}$  are the maximum length and width of a bounding box.  $A_a$  is the  
 225 projected area of the soot aggregate. For a spherical particle:  $D_f = 3$ , and  $\text{AR} = \text{RN} = 1$ .  
 226 For a nonspherical particle:  $1$  (linear particle)  $\leq D_f < 3$ ,  $\text{AR} > 1$ , and  $\text{RN} < 1$ . In addition,  
 227 the equivalent area diameter ( $D_{eq}$ ) represents the size of particle, which can be obtained  
 228 by equation (4).

$$D_{eq} = \sqrt{\frac{4A_a}{\pi}} \quad (4)$$

## 229 **2.4 The calculation of optical properties of soot particles**

230 The absorption, scattering and extinction cross sections of soot particles at 532 nm  
 231 were calculated by discrete dipole approximation (DDA) (<http://ddscat.wikidot.com/>),  
 232 which has been widely applied to calculate the optical properties of soot particles with  
 233 different coating materials and mixing structures [Draine and Flatau, 1994]. The input  
 234 parameters are refractive index of material, the shape and effective radius of particle  
 235 ( $D_f$ ,  $N$ ,  $d_p$ ). The refractive indices are  $1.76+0.63i$ ,  $1.53+0i$  and  $1.48+0i$  for soot

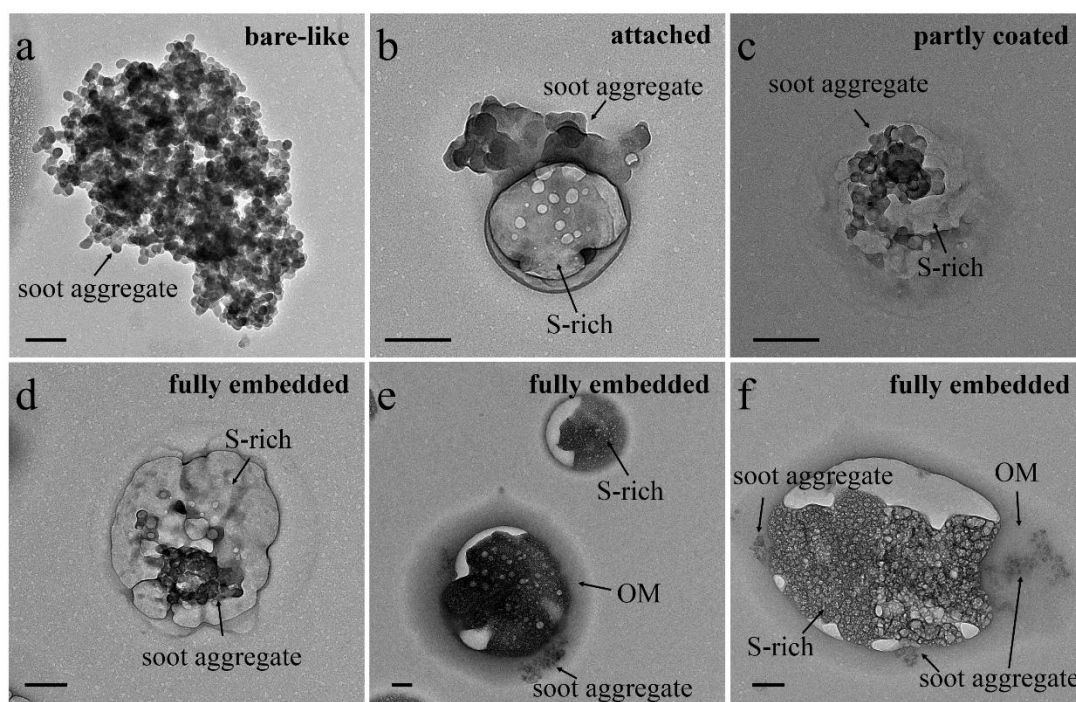
236 aggregate, sulfate coating and organic coating, respectively [Luo et al., 2019;  
237 Nakayama et al., 2010; Worringen et al., 2008]. Shape and effective radius can be  
238 obtained based on the images from TEM. Absorption enhancement ( $E_{abs}$ ) is the ratio of  
239  $C_{abs}$  (absorption cross section) of soot particles to soot aggregates, which is described  
240 in detail in the supporting information. In this study, the Electron-Microscope-to-BC-  
241 Simulation (EMBS) tool, recently developed by Wang et al. [2021b], was applied to  
242 build realistic soot shape models with diverse morphology and mixing structures. The  
243 average number and diameter of monomers of soot aggregates in the pre-cloud and in-  
244 cloud periods are 96, 102 and 45 nm, 44 nm, respectively. So soot aggregate is defined  
245 as 100 monomers with the diameter of 44 nm in the theoretical calculation. The  
246 variation of  $E_{abs}$  with the number of monomers is less than 3% (**Fig. S4**).

### 247 **3 Results and discussions**

#### 248 **3.1 The impact of cloud process on the mixing state and structure of soot particles**

249 Three components were identified within soot particles based on element  
250 composition and morphology, including soot, sulfate and organic matter [Li et al., 2016].  
251 Soot was identified by chain-like structure in morphology and abundant carbon in  
252 elemental composition. Organic matter exhibited spherical or irregular shape with  
253 abundant C and minor O. Sulfate was dominated by S and O, and some were  
254 accompanied by a small amount of N, K and Na. It is often bubbly appearance due to  
255 the loss of volatile materials (such as ammonium nitrate) by high-energy electron beams  
256 during TEM/EDS analysis. Therefore, the inorganic component is identified according  
257 to the bubble-shape and S-peak, so sulfate represent inorganic components here. Soot

258 particles contributed to 12-26% of all analyzed particles by number, which were divided  
259 into three classes based on the mixing materials directly contacting with soot aggregates  
260 (soot fraction in the whole particle), including bare-like soot aggregate with negligible  
261 coating (**Fig. 1a**), soot aggregate coated by sulfate (soot-S) (**Fig. 1b, c, d**), or by organic  
262 matter (soot-OM) (**Fig. 1e, f**). Such a classification stresses chemical components in  
263 direct contact with soot aggregates, which is slightly different from the classification  
264 based on the mixing state of the whole particle [Yuan et al., 2019]. According to various  
265 mixing structures between soot aggregates and other materials, soot particles were  
266 divided into four classes (**Fig. 1**), including bare-like (**Fig. 1a**), attached (**Fig. 1b**),  
267 partly coated (**Fig. 1c**), and fully embedded (**Fig. 1d-f**) [China et al., 2013; Liu et al.,  
268 2015]. Attached soot particle was soot aggregate adhering to the surface of the host  
269 particle. Partly coated and fully embedded soot particles referred to part and full of soot  
270 aggregate wrapped by other materials. Given the limited number of soot particles and  
271 similar air mass sources of two cloud events, the following results and discussions are  
272 based on a merged dataset from two cloud events.

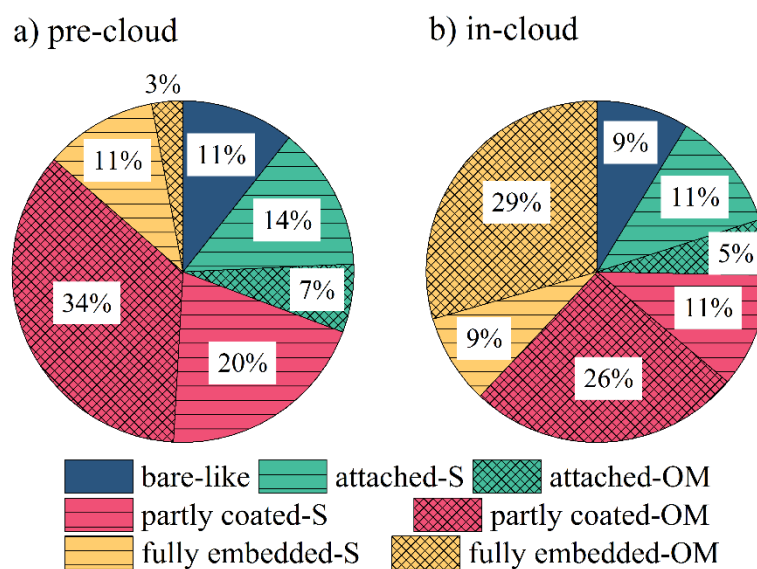


273

274 **Figure 1.** TEM images of soot particles with different mixing structures including bare-  
 275 like (a), attached (b), partly coated (c) and fully embedded (d-f). The scale in the images  
 276 is 200 nm.

277 It can be seen in **Fig. 2** that mixing states and structures of soot particles exhibit  
 278 distinctly different characteristics between the pre-cloud and in-cloud periods. Soot-  
 279 OM particles have a higher fraction in the in-cloud period ( $60 \pm 2.6\%$ ) than that in the  
 280 pre-cloud period ( $44 \pm 5.1\%$ ), but soot-S particles account for lower proportion in the  
 281 in-cloud period ( $31 \pm 3.7\%$ ) than pre-cloud period ( $45 \pm 5.1\%$ ). Compared with pre-  
 282 cloud period, the proportion of soot particles with fully embedded structure in the in-  
 283 cloud period shows a considerable increase (from  $14 \pm 2.8\%$  to  $38 \pm 2.9\%$ ). Specifically,  
 284 the proportion of soot particles fully embedded in OM, which increase from  $3 \pm 1.3\%$   
 285 in the pre-cloud period to  $29 \pm 2.6\%$  in the in-cloud period. Soot particles with partly  
 286 coated structure reduce from  $54 \pm 5.7\%$  in the pre-cloud period to  $37 \pm 2.9\%$  in the in-

287 cloud period. The contribution of bare-like and attached soot particles is lower, ranging  
 288 from  $9 \pm 1.4\%$  to  $21 \pm 3.4\%$ . The proportion of soot particles with LLPS was  $2 \pm 1.0\%$   
 289 and  $11 \pm 1.6\%$  in the pre-cloud and in-cloud period respectively, which were mainly  
 290 completely coated by OM (**Fig. 1e-f**).



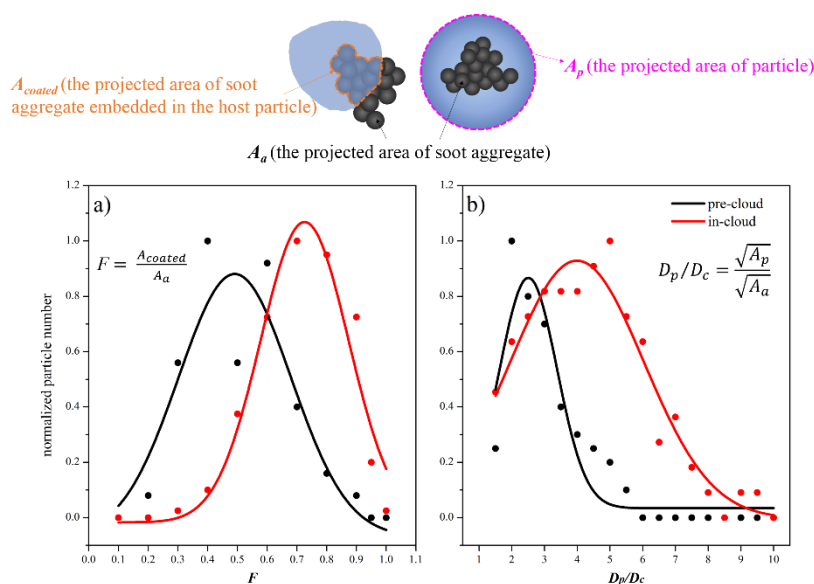
291

292 **Figure 2.** The number fraction of soot particles with different mixing structures coated  
 293 by sulfate (soot-S) or organics (soot-OM) in the pre-cloud and in-cloud periods.

294 The increase of soot-OM fraction in the in-cloud period is most likely attributed  
 295 to the cloud processing involving abundance of ozone and volatile organic compounds  
 296 (VOCs) [Lv et al., 2019; Wang et al., 2021c], which may also be attributed to the  
 297 preferential activation of soot-OM particles, but it is unlikely important. As observed  
 298 by Zhang et al. [2017a] at the same site, soot-OM particles are less activated than soot-  
 299 S particles. More soot-OM particles than soot-S particles during the in-cloud period  
 300 may be related to LLPS, in which the soot aggregates are distributed in the organic  
 301 phase with similar polarity rather than inorganic phase [Brunamonti et al., 2015].  
 302 Furthermore, as shown in **Fig. 3**, from the pre-cloud to in-cloud period, the average

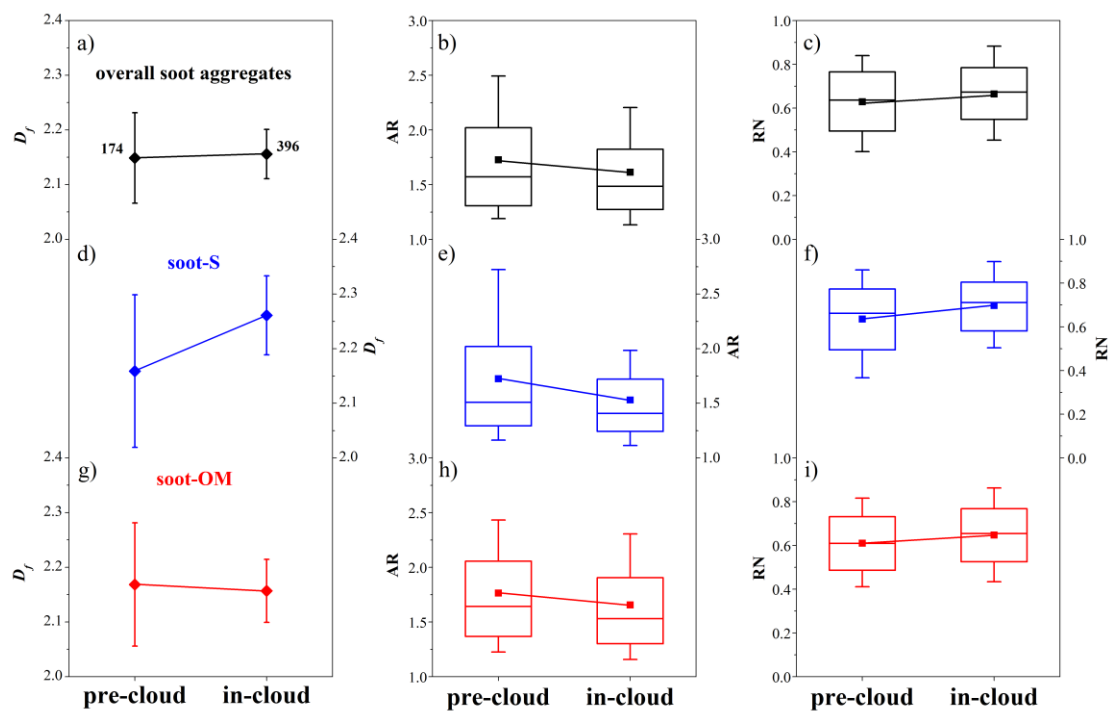


303 coated fraction ( $F$ , the area ratio of soot aggregate embedded in host particle to soot  
 304 aggregate) increased from  $46 \pm 15\%$  to  $67 \pm 14\%$  for soot particles with partly coated  
 305 structure, and average relative coating thickness ( $D_p/D_c$ , the diameter ratio between  
 306 particle and soot aggregate) increased from  $2.3 \pm 1.2$  to  $4.4 \pm 3.6$  for soot particles with  
 307 fully embedded structure. Consistently, the average  $D_{eq}$  of soot particles increased from  
 308  $845 \pm 417$  nm to  $1214 \pm 898$  nm after cloud processing. The increase of  $F$ ,  $D_p/D_c$  and  
 309  $D_{eq}$  indicates that soot aggregates obtain thicker coating due to cloud processing.



310

311 **Figure 3.** The distribution of  $F$  ( $0 < F < 1$ ) for partly coated soot particles (a) and  $D_p/D_c$   
 312 ( $> 1$ ) for fully embedded soot particles (b) in the pre-cloud and in-cloud periods. With  
 313 the increase of  $F$  and  $D_p/D_c$ , the volume ratio of the coated soot aggregates is higher,  
 314 and the coating layer is thicker.

315 **3.2 The impact of cloud process on the morphology of soot aggregates**

316

317 **Figure 4.** The  $D_f$  (a, d, g), AR (b, e, h) and RN (c, f, i) of soot aggregates including  
 318 overall soot aggregates, soot-S and soot-OM in the pre-cloud and in-cloud period. The  
 319 number (a) is the amount of soot aggregates used to calculate the morphology  
 320 parameters.  $D_f$  error bars are the error in the fitting of  $N$  and  $2R_g/d_p$  when calculating  $D_f$ .  
 321 Box and whisker plot shows lower, median, and upper lines, denoting the 25th, 50th,  
 322 and 75th percentiles, respectively; the lower and upper edges denote the 10th and 90th  
 323 percentiles, respectively.

324 The morphology parameters of overall soot aggregates, including  $D_f$ , AR and RN,  
 325 show a limited change from the pre-cloud to in-cloud period (**Fig. 4**). Consistently, the  
 326 RES and INT soot aggregates during the stability period of cloud process also exhibit  
 327 similar morphology with similar  $D_f$ , AR, and RN (**Fig. S5**). The above result indicates  
 328 limited restructuring of soot aggregates upon cloud processing, which is inconsistent

329 with previous studies reporting that high humidity condition facilitates the restructuring  
330 of soot aggregates [Qiu et al., 2012; Zhang et al., 2008], leading to larger  $D_f$  and RN,  
331 and lower AR.

332 Such inconsistency might be attributed to the chemically resolved reconstruction  
333 of soot aggregates due to cloud processing. As shown in **Fig. 4**, the  $D_f$  of soot-S  
334 aggregates was  $2.16 \pm 0.14$  in the pre-cloud period, and increased to  $2.26 \pm 0.07$  in the  
335 in-cloud period. Differently, the  $D_f$  of soot-OM aggregates between the pre-cloud and  
336 in-cloud periods were quite similar ( $2.17 \pm 0.11$  and  $2.16 \pm 0.06$ ). Similar trend was  
337 also observed for the AR and RN of soot aggregates (**Fig. 4**), with larger changes of AR  
338 and RN for soot-S aggregates from the pre-cloud to in-cloud periods. Such difference  
339 indicates that organics lead to less compaction of soot aggregates, rather than sulfate.  
340 The restructure of soot aggregates depends on the uptake of water by the coating  
341 materials [Guo et al., 2016], and the following evaporation of water [Enekwizu et al.,  
342 2021; Ma et al., 2013]. Complete encapsulation of organic coating produces lower  
343 surface tension than partly coated structure and sulfate coating during water evaporation  
344 [Schnitzler et al., 2017], resulting in less degree of reconstruction between monomers  
345 of soot aggregates fully embedded in organics. Such phenomenon is similar to that  
346 reported by Cross et al. [2010], who found that sulfuric acid coating caused a more  
347 compact shape of soot aggregates than DOS (dioctyl sebacate) coating in laboratory  
348 experiments. Therefore, with the increase contribution of soot-OM in the in-cloud  
349 period, in particular the fully embedded structure ( $> 50\%$ ) (**Fig. 2**), the overall soot  
350 aggregates show similar morphology parameters between pre-cloud and in-cloud

351 period.

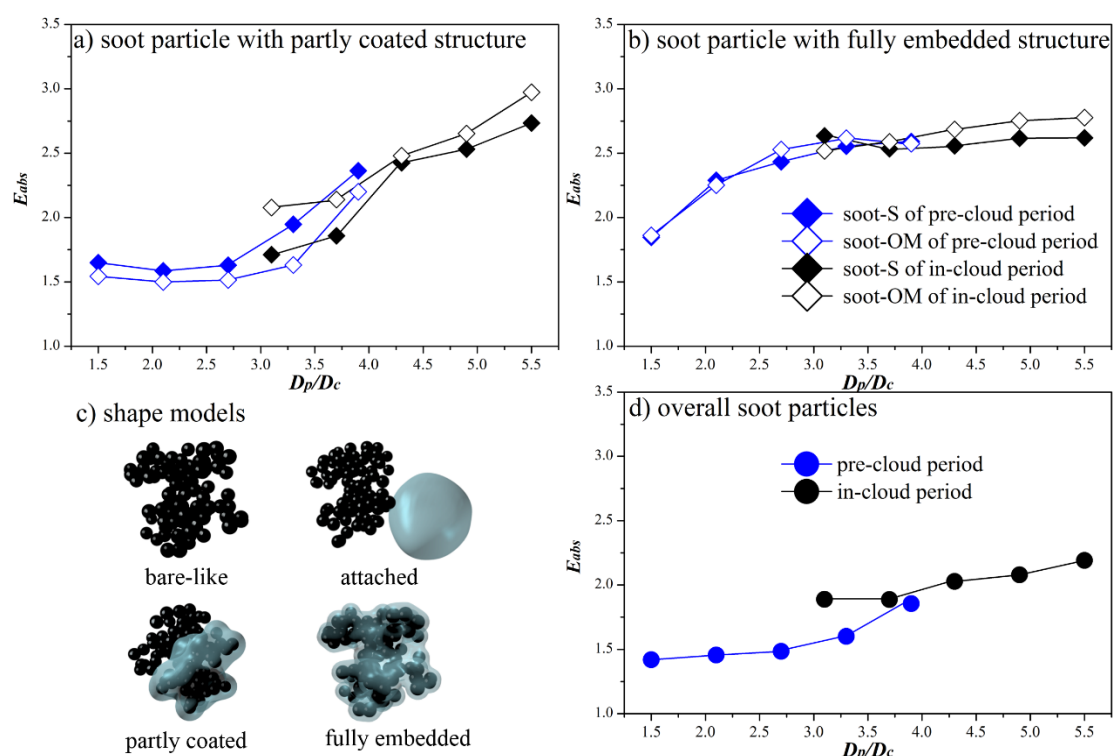
352 The differences in morphology, mixing state and mixing structure from the  
353 inactivated INT to activated RES particles during the stability period of cloud events  
354 can also support the above hypothesis. The proportion of soot-OM in the RES particles  
355 ( $68 \pm 4.6\%$ ) was higher than that in the INT particles ( $48 \pm 7.1\%$ ), and the proportion  
356 of soot-OM particles with fully embedded structure in the RES particles ( $37 \pm 3.4\%$ )  
357 was about 4 times that in the INT particles ( $9 \pm 3.5\%$ ) (**Fig. S6**). Furthermore, the  $D_f$  of  
358 soot-OM aggregates in the in-cloud period was smaller than that of the soot-S  
359 aggregates for the RES particles. Higher proportions and lower  $D_f$  value of soot-OM  
360 aggregates lead to the limited change in the morphology of overall soot aggregates  
361 between the RES and INT particles. Our previous study also found a phenomenon  
362 inconsistent with the general concept that lower  $D_f$  of soot aggregates in the RES  
363 particles than that in the INT particles at the same site [Fu et al., 2020], which is most  
364 likely due to the higher fraction of soot-OM in the RES particles ( $58 \pm 6.2\%$ ) than that  
365 in the INT particles ( $28 \pm 5.0\%$ ).

### 366 **3.3 Effect of cloud processing in light absorption of soot particles**

367 **Fig. 5** shows the  $E_{abs}$  of soot particles in the pre-cloud and in-cloud periods, based  
368 on the observed morphology, mixing state, and mixing structure. In order to illustrate  
369 the changes in the light absorption of soot particles within a representative size range  
370 upon cloud processing, five groups of  $D_p/D_c$  were selected, ranging from 1.5 to 3.9 in  
371 the pre-cloud period and from 3.1 to 5.5 in the in-cloud period (**Table S1**). The selection  
372 of the  $D_p/D_c$  range is based on the observed particles, as detailed in the supporting

373 information (Text S3). For bare-like soot particles, the  $D_f$  of soot-S aggregates increases  
374 from  $2.16 \pm 0.14$  in the pre-cloud period to  $2.26 \pm 0.07$  in the in-cloud period, resulting  
375 in a limited absorption enhancement with the  $E_{abs}$  of 1.02. As reported elsewhere, the  
376 enhancement of light absorption of soot particles with attached structures is limited  
377 ( $E_{abs} < 1.1$ ) [Adachi and Buseck, 2013; Schnaiter et al., 2003; Wang et al., 2021b].  
378 Differently, the light absorption increased significantly for soot particles with partly  
379 coated and fully embedded structures (**Fig. 5a, b**), associated with the increase of  
380 coating degree including coated fraction and thickness [Adachi et al., 2010; Guo et al.,  
381 2016]. With the increase of  $F$  and  $D_p/D_c$  after cloud processing,  $E_{abs}$  are  $1.76 \pm 0.29$  in  
382 the pre-cloud period and  $2.36 \pm 0.38$  in the in-cloud period for soot particles with partly  
383 coated structure. For soot particles with fully embedded structure,  $E_{abs}$  are  $2.35 \pm 0.28$   
384 in the pre-cloud period and  $2.63 \pm 0.08$  in the in-cloud period. The  $E_{abs}$  of soot particles  
385 with LLPS was  $1.78 \pm 0.14$  and  $1.93 \pm 0.07$  in the pre-cloud and in-cloud period  
386 respectively, which was smaller than that of soot particles without LLPS, with an  
387 average of 26% over different  $D_p/D_c$  (**Fig. S8**). The  $E_{abs}$  of soot particles with different  
388 mixing structures vary greatly, which may be one of the reasons for the lower  $E_{abs}$  of  
389 soot particles in some field investigation cases [Cappa et al., 2012; Fu et al., 2021], that  
390 is, the attached and partly coated structures soot particles may account for the main  
391 proportion. In the pre-cloud period, the average  $E_{abs}$  of partly coated soot particles ( $1.76$   
392  $\pm 0.29$ ) and the average  $E_{abs}$  of fully embedded soot particles ( $2.35 \pm 0.28$ ) were both  
393 within the range reported by current laboratory studies (1-1.9 for partly coated structure  
394 and 2.2-2.5 for fully embedded structure) [Chakrabarty and Heinson, 2018; Scarnato et

395 al., 2013], which were lower than the average  $E_{abs}$  of soot particles with partly coated  
 396 ( $2.36 \pm 0.38$ ) and fully embedded ( $2.63 \pm 0.08$ ) structure in the in-cloud period.  
 397 Additionally, the  $E_{abs}$  of soot-S and soot-OM particles with different mixing structures,  
 398  $F$  and  $D_p/D_c$  is in the range of 1.0-2.7 and 1.0-3.0 respectively, which is consistent with  
 399 previous studies [Peng et al., 2016; Schnaiter et al., 2003; Schnaiter et al., 2005].



400

401 **Figure 5.** Shape models of soot particles with different mixing structures, and  $E_{abs}$  of  
 402 soot particles in the pre-cloud and in-cloud periods.

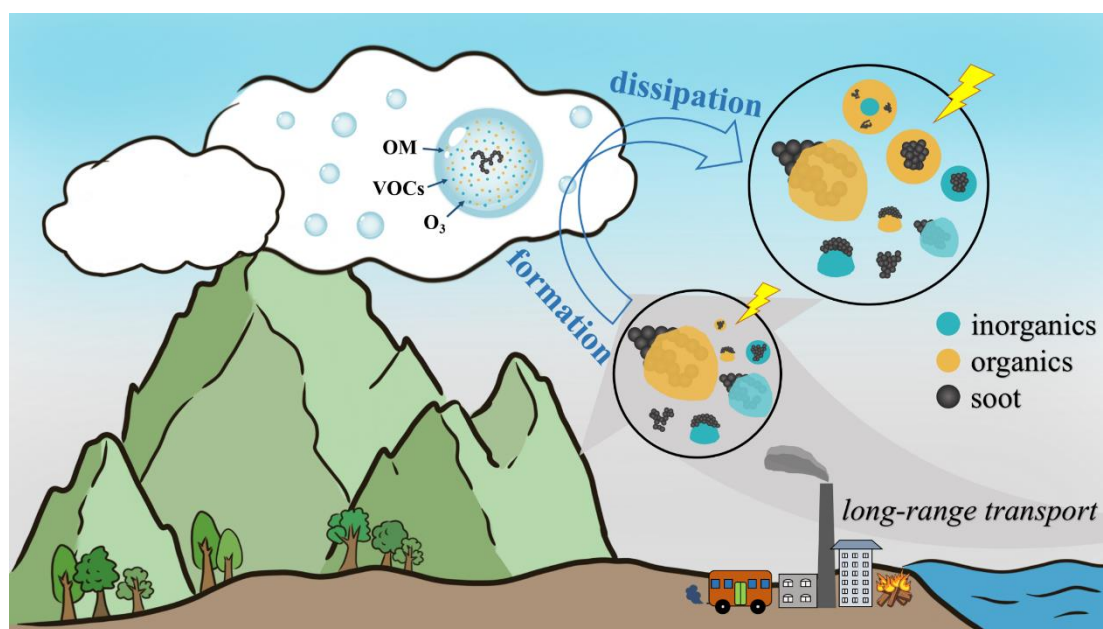
403 For the overall soot particles, the cloud process likely induces a pronounced  
 404 increase of light absorption based on theoretical calculations. Corresponding to the  
 405 changes in the proportion of soot particles with various mixing states and structures  
 406 upon cloud processing, the  $E_{abs}$  of all soot particles increased from  $1.57 \pm 0.16$  in the  
 407 pre-cloud period to  $2.01 \pm 0.12$  in the in-cloud period (**Fig. 5d**), and the cloud process

408 resulted in a factor of  $1.29 \pm 0.06$  increase in  $E_{abs}$ . Our calculated  $E_{abs}$  of soot particles  
409 in the pre-cloud period is higher than that in urban areas [Healy et al., 2015; Lan et al.,  
410 2013; Nakayama et al., 2014], which may be due to the higher degree of aging of the  
411 soot particles at this remote site. Higher  $E_{abs}$  of soot particles (2.1) has also been  
412 observed at the cloudy Mt. Tai in northern China [Wang et al., 2021a]. As indicated by  
413 Peng et al. [2016], the  $E_{abs}$  of soot particles could increase by a factor of 2.4 after several  
414 hour's aging in polluted urban air. Our study indicates that cloud processing during one  
415 event could increase the  $E_{abs}$  of soot particles by 1.29 times, which underlines the  
416 potential effect of cloud on rapid aging of soot particles.

### 417 **3.4 A conceptual model for the impact on the mixing state and microphysical** 418 **properties of soot particles of cloud process**

419 We came up with a conceptual model, as illustrated in **Fig. 6**, to reflect the changes  
420 of mixing state and microphysical properties of soot particles from the pre-cloud to in-  
421 cloud period. In the pre-cloud period, soot particles come from long-range transport  
422 and are mainly partly coated. Aged soot particles would be activated and grow into  
423 cloud droplets during cloud events. The mixing structure of soot particles is expected  
424 to change since the water-soluble components mixed with the soot aggregates will be  
425 dissolved in the cloud droplets. In addition, secondary organic compositions could also  
426 be formed through uptake and aqueous processing of VOCs [Fu et al., 2020; Zhang et  
427 al., 2017a]. After the evaporation of cloud droplets, LLPS occurs and would lead to  
428 two-phase solid particles composed of outer organic phase and inner inorganic phase  
429 [Shiraiwa et al., 2013], and soot aggregates will redistribute in organic phase with

430 similar polarity. Although not emphasized, the soot aggregates have been found to be  
431 dominantly embedded in the organic phase in the remote region [Yu et al., 2019], with  
432 a large fraction (61%) of soot-OM aggregates distributed in outer organic phase likely  
433 due to cloud processing. The production of organic matter in cloud and the occurrence  
434 of LLPS correspond to the significant increase of soot-OM particles with fully  
435 embedded structure, which would significantly enhance the light absorption of soot  
436 particles.



437  
438 **Figure 6.** Schematic diagram showing the change in morphology, mixing state, and  
439 mixing structure of soot particles throughout cloud processing.

#### 440 **4 Atmospheric implications and conclusions**

441 This study emphasizes that it is necessary to consider the encapsulating materials  
442 and mixing structures for better understanding of the impact of cloud on soot particles.  
443 Attributed to cloud processing, the soot-S aggregates become more compact, while the  
444 soot-OM aggregates do not. Meanwhile, the changes in morphology of soot aggregates,



445 mixing state and mixing structure of soot particles driven by cloud processing would  
446 have substantial implications for the optical properties of soot particles. Theoretical  
447 calculation shows that the  $E_{abs}$  of soot-particles processed by cloud is as high as  $2.01 \pm$   
448  $0.12$ , which is  $1.29 \pm 0.06$  times that of the pre-cloud period. Given that clouds cover  
449 nearly 70% of the Earth [King et al., 2013], such a significant increase in  $E_{abs}$  indicates  
450 that cloud processing may represent a critical factor, which has not been revealed in the  
451 evolution of light absorption of soot. The results also support the significance of  
452 morphology, mixing state and mixing structure of soot particles in model calculations  
453 to evaluate the climate impact of soot particles [Penner et al., 2006; Zhuang et al., 2010].  
454 While such information is accounted for in particle-resolved aerosol model [Fierce et  
455 al., 2020], it is still not considered in most climate models [Cooke and Wilson, 1996;  
456 Park et al., 2011; Zhang et al., 2019]. In addition, field observations found high  
457 uncertainty in the abundance of soot particles in ice clouds [Cozic et al., 2008;  
458 Kupiszewski et al., 2016]. Our observations indicate that the morphological evolution  
459 of soot aggregates due to warm cloud processing was mainly driven by the coating of  
460 organic compounds, which may have potential influence on the ice nucleation activities  
461 [Nichman et al., 2019]. Laboratory simulation experiments have shown a significant  
462 enhancement of ice nucleation activity of soot particles processed by cloud, most likely  
463 attributed to the modification of morphology by cloud processing [Mahrt et al., 2020].  
464 Our results at the remote site surrounded by a large area of forest may represent the  
465 mixing state and morphological characteristics of soot particles in the area with  
466 significant biogenic VOCs precursors, and thus can be used as a reference for the

467 climate model to access the climatic impact of soot particles in similar regions.

#### 468 **Acknowledgements**

469 This work was funded by the Natural Science Foundation of Guangdong Province  
470 (2019B151502022), National Natural Science Foundation of China (42192514,  
471 42077322, and 41877307), Youth Innovation Promotion Association CAS (2021354),  
472 and Guangdong Foundation for Program of Science and Technology Research  
473 (2020B1212060053).

#### 474 **Conflict of Interest**

475 There are no conflicts of interest in the article.

#### 476 **Data Availability Statement**

477 Data supporting the results are available in the supporting information. The  
478 processed data presented in this paper are available on Zenodo  
479 (<https://doi.org/10.5281/zenodo.7073072>) [Fu et al., 2022], including the number  
480 fraction of soot particles with different mixing states and structures, the distribution of  
481  $F$  and  $D_p/D_c$ , morphology parameters of soot aggregates, and  $E_{abs}$  of soot particles in  
482 the pre-cloud and in-cloud periods. The back trajectories data are calculated with hybrid  
483 single-particle Lagrangian integrated trajectory model  
484 ([https://www.ready.noaa.gov/HYSPLIT\\_hytrial.php](https://www.ready.noaa.gov/HYSPLIT_hytrial.php)) [NOAA, 2022]. The source code  
485 of EMBS is from Wang et al. [2021b]. The source code of DDA developed by Bruce T.  
486 Draine and Piotr J. Flatau, and the version used in this study is DDSCAT 7.3.3, which  
487 can be obtained online (<http://ddscat.wikidot.com/>) [Draine and Flatau, 2019].

488 **References**

489 Adachi, K., and P. R. Buseck (2013), Changes of ns-soot mixing states and shapes in  
490 an urban area during CalNex, *Journal of Geophysical Research-Atmospheres*, *118*(9),  
491 3723-3730, doi:10.1002/jgrd.50321.

492 Adachi, K., S. H. Chung, and P. R. Buseck (2010), Shapes of soot aerosol particles and  
493 implications for their effects on climate, *Journal of Geophysical Research-Atmospheres*,  
494 *115*, D15206, doi:10.1029/2009jd012868.

495 Bhandari, J., et al. (2019), Extensive Soot Compaction by Cloud Processing from  
496 Laboratory and Field Observations, *Scientific reports*, *9*(1), 11824-11824,  
497 doi:10.1038/s41598-019-48143-y.

498 Brasil, A. M., T. L. Farias, and M. G. Carvalho (1999), A recipe for image  
499 characterization of fractal-like aggregates, *Journal of Aerosol Science*, *30*(10), 1379-  
500 1389, doi:10.1016/s0021-8502(99)00026-9.

501 Brunamonti, S., U. K. Krieger, C. Marcolli, and T. Peter (2015), Redistribution of black  
502 carbon in aerosol particles undergoing liquid-liquid phase separation, *Geophysical*  
503 *Research Letters*, *42*(7), 2532-2539, doi:10.1002/2014gl062908.

504 Buseck, P. R., K. Adachi, G. Andras, E. Tompa, and P. Mihaly (2014), Ns-Soot: A  
505 Material-Based Term for Strongly Light-Absorbing Carbonaceous Particles, *Aerosol*  
506 *Science and Technology*, *48*(7), 777-788, doi:10.1080/02786826.2014.919374.

507 Cao, L.-M., X.-F. Huang, Y.-Y. Li, M. Hu, and L.-Y. He (2018), Volatility measurement  
508 of atmospheric submicron aerosols in an urban atmosphere in southern China,  
509 *Atmospheric Chemistry and Physics*, *18*(3), 1729-1743, doi:10.5194/acp-18-1729-2018.

510 Cappa, C. D., et al. (2012), Radiative Absorption Enhancements Due to the Mixing  
511 State of Atmospheric Black Carbon, *Science*, 337(6098), 1078-1081,  
512 doi:10.1126/science.1223447.

513 Chakrabarty, R. K., and W. R. Heinson (2018), Scaling Laws for Light Absorption  
514 Enhancement Due to Nonrefractory Coating of Atmospheric Black Carbon Aerosol,  
515 *Physical Review Letters*, 121(21), doi:10.1103/PhysRevLett.121.218701.

516 Chen, C., O. Y. Enekwizu, X. Ma, Y. Jiang, A. F. Khalizov, J. Zheng, and Y. Ma (2020),  
517 Effect of organic coatings derived from the OH-initiated oxidation of amines on soot  
518 morphology and cloud activation, *Atmospheric Research*, 239, 104905,  
519 doi:10.1016/j.atmosres.2020.104905.

520 Chen, C., X. Fan, T. Shaltout, C. Qiu, Y. Ma, A. Goldman, and A. F. Khalizov (2016),  
521 An unexpected restructuring of combustion soot aggregates by subnanometer coatings  
522 of polycyclic aromatic hydrocarbons, *Geophysical Research Letters*, 43(20), 11080-  
523 11088, doi:10.1002/2016gl070877.

524 China, S., et al. (2015a), Morphology of diesel soot residuals from supercooled water  
525 droplets and ice crystals: implications for optical properties, *Environmental Research*  
526 *Letters*, 10(11), 114010, doi:10.1088/1748-9326/10/11/114010.

527 China, S., C. Mazzoleni, K. Gorkowski, A. C. Aiken, and M. K. Dubey (2013),  
528 Morphology and mixing state of individual freshly emitted wildfire carbonaceous  
529 particles, *Nature Communications*, 4, 2122, doi:10.1038/ncomms3122.

530 China, S., N. Salvadori, and C. Mazzoleni (2014), Effect of Traffic and Driving  
531 Characteristics on Morphology of Atmospheric Soot Particles at Freeway On-Ramps,

- 532 *Environmental Science & Technology*, 48(6), 3128-3135, doi:10.1021/es405178n.
- 533 China, S., et al. (2015b), Morphology and mixing state of aged soot particles at a remote  
534 marine free troposphere site: Implications for optical properties, *Geophysical Research*  
535 *Letters*, 42(4), 1243-1250, doi:10.1002/2014gl062404.
- 536 Cooke, W. F., and J. J. N. Wilson (1996), A global black carbon aerosol model, *Journal*  
537 *of Geophysical Research-Atmospheres*, 101(D14), 19395-19409,  
538 doi:10.1029/96jd00671.
- 539 Cozic, J., S. Mertes, B. Verheggen, D. J. Cziczo, S. J. Gallavardin, S. Walter, U.  
540 Baltensperger, and E. Weingartner (2008), Black carbon enrichment in atmospheric ice  
541 particle residuals observed in lower tropospheric mixed phase clouds, *Journal of*  
542 *Geophysical Research-Atmospheres*, 113(D15), doi:10.1029/2007jd009266.
- 543 Cross, E. S., et al. (2010), Soot Particle Studies-Instrument Inter-Comparison-Project  
544 Overview, *Aerosol Science and Technology*, 44(8), 592-611,  
545 doi:10.1080/02786826.2010.482113.
- 546 Draine, B. T., and P. J. Flatau (1994), DISCRETE-DIPOLE APPROXIMATION FOR  
547 SCATTERING CALCULATIONS, *Journal of the Optical Society of America a-Optics*  
548 *Image Science and Vision*, 11(4), 1491-1499, doi:10.1364/josaa.11.001491.
- 549 Draine, B. T., and P. J. Flatau (2019), Discrete Dipole Scattering (DDSCAT) [Software].  
550 <http://ddscat.wikidot.com/>.
- 551 Enekwizu, O. Y., A. Hasani, and A. F. Khalizov (2021), Vapor Condensation and  
552 Coating Evaporation Are Both Responsible for Soot Aggregate Restructuring,  
553 *Environmental Science & Technology*, 55(13), 8622-8630, doi:10.1021/acs.est.1c02391.

554 Fierce, L., et al. (2020), Radiative absorption enhancements by black carbon controlled  
555 by particle-to-particle heterogeneity in composition, *Proceedings of the National*  
556 *Academy of Sciences of the United States of America*, *117*(10), 5196-5203,  
557 doi:10.1073/pnas.1919723117.

558 Fu, Y., et al. (2020), Impact of in-cloud aqueous processes on the chemical  
559 compositions and morphology of individual atmospheric aerosols, *Atmospheric*  
560 *Chemistry and Physics*, *20*(22), 14063-14075, doi:10.5194/acp-20-14063-2020.

561 Fu, Y., et al. (2021), Filter-based absorption enhancement measurement for internally  
562 mixed black carbon particles over southern China, *Science of The Total Environment*,  
563 *762*, 144194, doi:<https://doi.org/10.1016/j.scitotenv.2020.144194>.

564 Fu, Y., et al. (2022), Impact of Cloud Process in the Mixing State and Microphysical  
565 Properties of Soot Particles: Implications in Light Absorption Enhancement [Dataset].  
566 Zenodo, <https://doi.org/10.5281/zenodo.4265639>.

567 Gong, D., et al. (2018), Low-level summertime isoprene observed at a forested  
568 mountaintop site in southern China: implications for strong regional atmospheric  
569 oxidative capacity, *Atmospheric Chemistry and Physics*, *18*(19), 14417-14432,  
570 doi:10.5194/acp-18-14417-2018.

571 Guo, S., M. Hu, Y. Lin, M. Gomez-Hernandez, M. L. Zamora, J. Peng, D. R. Collins,  
572 and R. Zhang (2016), OH-Initiated Oxidation of m-Xylene on Black Carbon Aging,  
573 *Environmental Science & Technology*, *50*(16), 8605-8612,  
574 doi:10.1021/acs.est.6b01272.

575 Healy, R. M., et al. (2015), Light-absorbing properties of ambient black carbon and

576 brown carbon from fossil fuel and biomass burning sources, *Journal of Geophysical*  
577 *Research-Atmospheres*, 120(13), 6619-6633, doi:10.1002/2015jd023382.

578 Jacobson, M. Z. (2001), Strong radiative heating due to the mixing state of black carbon  
579 in atmospheric aerosols, *Nature*, 409(6821), 695-697, doi:10.1038/35055518.

580 Kelesidis, G. A., D. Neubauer, L.-S. Fan, U. Lohmann, and S. E. Pratsinis (2022),  
581 Enhanced Light Absorption and Radiative Forcing by Black Carbon Agglomerates,  
582 *Environmental Science & Technology*, 56(12), 8610-8618, doi:10.1021/acs.est.2c00428.

583 Khalizov, A. F., R. Zhang, D. Zhang, H. Xue, J. Pagels, and P. H. McMurry (2009),  
584 Formation of highly hygroscopic soot aerosols upon internal mixing with sulfuric acid  
585 vapor, *Journal of Geophysical Research-Atmospheres*, 114, D05208,  
586 doi:10.1029/2008jd010595.

587 Kim, H., S. Collier, X. Ge, J. Xu, Y. Sun, W. Jiang, Y. Wang, P. Herckes, and Q. Zhang  
588 (2019), Chemical processing of water-soluble species and formation of secondary  
589 organic aerosol in fogs, *Atmospheric Environment*, 200, 158-166,  
590 doi:10.1016/j.atmosenv.2018.11.062.

591 King, M. D., S. Platnick, W. P. Menzel, S. A. Ackerman, and P. A. Hubanks (2013),  
592 Spatial and Temporal Distribution of Clouds Observed by MODIS Onboard the Terra  
593 and Aqua Satellites, *Ieee Transactions on Geoscience and Remote Sensing*, 51(7),  
594 3826-3852, doi:10.1109/tgrs.2012.2227333.

595 Kupiszewski, P., et al. (2016), Ice residual properties in mixed-phase clouds at the high-  
596 alpine Jungfrauoch site, *Journal of Geophysical Research-Atmospheres*, 121(20),  
597 12343-12362, doi:10.1002/2016jd024894.

598 Lack, D. A., and C. D. Cappa (2010), Impact of brown and clear carbon on light  
599 absorption enhancement, single scatter albedo and absorption wavelength dependence  
600 of black carbon, *Atmospheric Chemistry and Physics*, *10*(9), 4207-4220,  
601 doi:10.5194/acp-10-4207-2010.

602 Lan, Z.-J., X.-F. Huang, K.-Y. Yu, T.-L. Sun, L.-W. Zeng, and M. Hu (2013), Light  
603 absorption of black carbon aerosol and its enhancement by mixing state in an urban  
604 atmosphere in South China, *Atmospheric Environment*, *69*, 118-123,  
605 doi:10.1016/j.atmosenv.2012.12.009.

606 Li, K., X. Ye, H. Pang, X. Lu, H. Chen, X. Wang, X. Yang, J. Chen, and Y. Chen (2018),  
607 Temporal variations in the hygroscopicity and mixing state of black carbon aerosols in  
608 a polluted megacity area, *Atmospheric Chemistry and Physics*, *18*(20), 15201-15218,  
609 doi:10.5194/acp-18-15201-2018.

610 Li, W., P. Li, G. Sun, S. Zhou, Q. Yuan, and W. Wang (2011), Cloud residues and  
611 interstitial aerosols from non-precipitating clouds over an industrial and urban area in  
612 northern China, *Atmospheric Environment*, *45*(15), 2488-2495,  
613 doi:10.1016/j.atmosenv.2011.02.044.

614 Li, W., and L. Shao (2009), Transmission electron microscopy study of aerosol particles  
615 from the brown hazes in northern China, *Journal of Geophysical Research-*  
616 *Atmospheres*, *114*, D09302, doi:10.1029/2008jd011285.

617 Li, W., L. Shao, D. Zhang, C.-U. Ro, M. Hu, X. Bi, H. Geng, A. Matsuki, H. Niu, and  
618 J. Chen (2016), A review of single aerosol particle studies in the atmosphere of East  
619 Asia: morphology, mixing state, source, and heterogeneous reactions, *Journal of*



- 620 *Cleaner Production*, 112, 1330-1349, doi:10.1016/j.jclepro.2015.04.050.
- 621 Lin, Q., et al. (2019), Enrichment of submicron sea-salt-containing particles in small  
622 cloud droplets based on single-particle mass spectrometry, *Atmospheric Chemistry and*  
623 *Physics*, 19(16), 10469-10479, doi:10.5194/acp-19-10469-2019.
- 624 Liu, D., J. Allan, J. Whitehead, D. Young, M. Flynn, H. Coe, G. McFiggans, Z. L.  
625 Fleming, and B. Bandy (2013), Ambient black carbon particle hygroscopic properties  
626 controlled by mixing state and composition, *Atmospheric Chemistry and Physics*, 13(4),  
627 2015-2029, doi:10.5194/acp-13-2015-2013.
- 628 Liu, D., et al. (2017a), Black-carbon absorption enhancement in the atmosphere  
629 determined by particle mixing state, *Nature Geoscience*, 10(3), 184-U132,  
630 doi:10.1038/ngeo2901.
- 631 Liu, L., et al. (2017b), Morphology, composition, and mixing state of primary particles  
632 from combustion sources - crop residue, wood, and solid waste, *Scientific Reports*, 7,  
633 5047, doi:10.1038/s41598-017-05357-2.
- 634 Liu, L., et al. (2018), Cloud scavenging of anthropogenic refractory particles at a  
635 mountain site in North China, *Atmospheric Chemistry and Physics*, 18(19), 14681-  
636 14693, doi:10.5194/acp-18-14681-2018.
- 637 Liu, S., et al. (2015), Enhanced light absorption by mixed source black and brown  
638 carbon particles in UK winter, *Nature Communications*, 6, 8435,  
639 doi:10.1038/ncomms9435.
- 640 Luo, J., Q. Zhang, J. Luo, J. Liu, Y. Huo, and Y. Zhang (2019), Optical Modeling of  
641 Black Carbon With Different Coating Materials: The Effect of Coating Configurations,

642 *Journal of Geophysical Research-Atmospheres*, 124(23), 13230-13253,

643 doi:10.1029/2019jd031701.

644 Lv, S., et al. (2019), Elevated levels of glyoxal and methylglyoxal at a remote mountain

645 site in southern China: Prompt in-situ formation combined with strong regional

646 transport, *The Science of the total environment*, 672, 869-882,

647 doi:10.1016/j.scitotenv.2019.04.020.

648 Ma, X., C. D. Zangmeister, J. Gigault, G. W. Mulholland, and M. R. Zachariah (2013),

649 Soot aggregate restructuring during water processing, *Journal of Aerosol Science*, 66,

650 209-219, doi:10.1016/j.jaerosci.2013.08.001.

651 Mahrt, F., K. Kilchhofer, C. Marcolli, P. Gronquist, R. O. David, M. Rosch, U.

652 Lohmann, and Z. A. Kanji (2020), The Impact of Cloud Processing on the Ice

653 Nucleation Abilities of Soot Particles at Cirrus Temperatures, *Journal of Geophysical*

654 *Research-Atmospheres*, 125(3), e2019JD030922, doi:10.1029/2019jd030922.

655 Nakayama, T., Y. Ikeda, Y. Sawada, Y. Setoguchi, S. Ogawa, K. Kawana, M. Mochida,

656 F. Ikemori, K. Matsumoto, and Y. Matsumi (2014), Properties of light-absorbing

657 aerosols in the Nagoya urban area, Japan, in August 2011 and January 2012:

658 Contributions of brown carbon and lensing effect, *Journal of Geophysical Research-*

659 *Atmospheres*, 119(22), 12721-12739, doi:10.1002/2014jd021744.

660 Nakayama, T., Y. Matsumi, K. Sato, T. Imamura, A. Yamazaki, and A. Uchiyama

661 (2010), Laboratory studies on optical properties of secondary organic aerosols

662 generated during the photooxidation of toluene and the ozonolysis of alpha-pinene,

663 *Journal of Geophysical Research-Atmospheres*, 115, doi:10.1029/2010jd014387.

664 Nichman, L., M. Wolf, P. Davidovits, T. B. Onasch, Y. Zhang, D. R. Worsnop, J.  
665 Bhandari, C. Mazzoleni, and D. J. Cziczo (2019), Laboratory study of the  
666 heterogeneous ice nucleation on black-carbon-containing aerosol, *Atmospheric*  
667 *Chemistry and Physics*, *19*(19), 12175-12194, doi:10.5194/acp-19-12175-2019.

668 NOAA. (2022). Hybrid single-particle Lagrangian integrated trajectory model  
669 [Software]. NOAA. Retrieved from <https://www.ready.noaa.gov/HYSPLIT.php>.

670 Park, S. H., S. L. Gong, V. S. Bouchet, W. Gong, P. A. Makar, M. D. Moran, C. A.  
671 Stroud, and J. Zhang (2011), Effects of black carbon aging on air quality predictions  
672 and direct radiative forcing estimation, *Tellus Series B-Chemical and Physical*  
673 *Meteorology*, *63*(5), 1026-1039, doi:10.1111/j.1600-0889.2011.00558.x.

674 Peng, J., et al. (2016), Markedly enhanced absorption and direct radiative forcing of  
675 black carbon under polluted urban environments, *Proceedings of the National Academy*  
676 *of Sciences of the United States of America*, *113*(16), 4266-4271,  
677 doi:10.1073/pnas.1602310113.

678 Penner, J. E., J. Quaas, T. Storelvmo, T. Takemura, O. Boucher, H. Guo, A. Kirkevag,  
679 J. E. Kristjansson, and O. Seland (2006), Model intercomparison of indirect aerosol  
680 effects, *Atmospheric Chemistry and Physics*, *6*, 3391-3405, doi:10.5194/acp-6-3391-  
681 2006.

682 Pratt, K. A., A. J. Heymsfield, C. H. Twohy, S. M. Murphy, P. J. DeMott, J. G. Hudson,  
683 R. Subramanian, Z. Wang, J. H. Seinfeld, and K. A. Prather (2010), In Situ Chemical  
684 Characterization of Aged Biomass-Burning Aerosols Impacting Cold Wave Clouds,  
685 *Journal of the Atmospheric Sciences*, *67*(8), 2451-2468, doi:10.1175/2010jas3330.1.

686 Qiu, C., A. F. Khalizov, and R. Zhang (2012), Soot Aging from OH-Initiated Oxidation  
687 of Toluene, *Environmental Science & Technology*, *46*(17), 9464-9472,  
688 doi:10.1021/es301883y.

689 Radney, J. G., R. You, X. Ma, J. M. Conny, M. R. Zachariah, J. T. Hodges, and C. D.  
690 Zangmeister (2014), Dependence of Soot Optical Properties on Particle Morphology:  
691 Measurements and Model Comparisons, *Environmental Science & Technology*, *48*(6),  
692 3169-3176, doi:10.1021/es4041804.

693 Samson, R. J., G. W. Mulholland, and J. W. Gentry (1987), STRUCTURAL-  
694 ANALYSIS OF SOOT AGGLOMERATES, *Langmuir*, *3*(2), 272-281,  
695 doi:10.1021/la00074a022.

696 Scarnato, B. V., S. Vahidinia, D. T. Richard, and T. W. Kirchstetter (2013), Effects of  
697 internal mixing and aggregate morphology on optical properties of black carbon using  
698 a discrete dipole approximation model, *Atmospheric Chemistry and Physics*, *13*(10),  
699 5089-5101, doi:10.5194/acp-13-5089-2013.

700 Schnaiter, M., H. Horvath, O. Mohler, K. H. Naumann, H. Saathoff, and O. W. Schock  
701 (2003), UV-VIS-NIR spectral optical properties of soot and soot-containing aerosols,  
702 *Journal of Aerosol Science*, *34*(10), 1421-1444, doi:10.1016/s0021-8502(03)00361-6.

703 Schnaiter, M., C. Linke, O. Mohler, K. H. Naumann, H. Saathoff, R. Wagner, U.  
704 Schurath, and B. Wehner (2005), Absorption amplification of black carbon internally  
705 mixed with secondary organic aerosol, *Journal of Geophysical Research-Atmospheres*,  
706 *110*(D19), D19204, doi:10.1029/2005jd006046.

707 Schneider, J., S. Mertes, D. van Pinxteren, H. Herrmann, and S. Borrmann (2017),

708 Uptake of nitric acid, ammonia, and organics in orographic clouds: mass spectrometric  
709 analyses of droplet residual and interstitial aerosol particles, *Atmospheric Chemistry  
710 and Physics*, 17(2), 1571-1593, doi:10.5194/acp-17-1571-2017.

711 Schnitzler, E. G., J. M. Gac, and W. Jaeger (2017), Coating surface tension dependence  
712 of soot aggregate restructuring, *Journal of Aerosol Science*, 106, 43-55,  
713 doi:10.1016/j.jaerosci.2017.01.005.

714 Sharma, N., S. China, J. Bhandari, K. Gorkowski, M. Dubey, R. A. Zaveri, and C.  
715 Mazzoleni (2018), Physical Properties of Aerosol Internally Mixed With Soot Particles  
716 in a Biogenically Dominated Environment in California, *Geophysical Research Letters*,  
717 45(20), 11473-11482, doi:10.1029/2018gl079404.

718 Shingler, T., et al. (2012), Characterisation and airborne deployment of a new  
719 counterflow virtual impactor inlet, *Atmospheric Measurement Techniques*, 5(6), 1259-  
720 1269, doi:10.5194/amt-5-1259-2012.

721 Shiraiwa, M., A. Zuend, A. K. Bertram, and J. H. Seinfeld (2013), Gas-particle  
722 partitioning of atmospheric aerosols: interplay of physical state, non-ideal mixing and  
723 morphology, *Physical Chemistry Chemical Physics*, 15(27), 11441-11453,  
724 doi:10.1039/c3cp51595h.

725 Wang, Y., et al. (2021a), Nonlinear Enhancement of Radiative Absorption by Black  
726 Carbon in Response to Particle Mixing Structure, *Geophysical Research Letters*, 48(24),  
727 doi:10.1029/2021gl096437.

728 Wang, Y., F. Liu, C. He, L. Bi, T. Cheng, Z. Wang, H. Zhang, X. Zhang, Z. Shi, and W.  
729 Li (2017), Fractal Dimensions and Mixing Structures of Soot Particles during

730 Atmospheric Processing, *Environmental Science & Technology Letters*, 4(11), 487-493,

731 doi:10.1021/acs.estlett.7b00418.

732 Wang, Y., Y. Pang, J. Huang, L. Bi, H. Che, X. Zhang, and W. Li (2021b), Constructing

733 Shapes and Mixing Structures of Black Carbon Particles With Applications to Optical

734 Calculations, *Journal of Geophysical Research-Atmospheres*, 126(10),

735 e2021JD034620, doi:10.1029/2021jd034620. [Software].

736 Wang, Y., et al. (2021c), Unexpected seasonal variations and high levels of ozone

737 observed at the summit of Nanling Mountains: Impact of Asian monsoon on southern

738 China, *Atmospheric Environment*, 253, doi:10.1016/j.atmosenv.2021.118378.

739 Wentzel, M., H. Gorzawski, K. H. Naumann, H. Saathoff, and S. Weinbruch (2003),

740 Transmission electron microscopical and aerosol dynamical characterization of soot

741 aerosols, *Journal of Aerosol Science*, 34(10), 1347-1370, doi:10.1016/s0021-

742 8502(03)00360-4.

743 Worringen, A., M. Ebert, T. Trautmann, S. Weinbruch, and G. Helas (2008), Optical

744 properties of internally mixed ammonium sulfate and soot particles-a study of

745 individual aerosol particles and ambient aerosol populations, *Applied Optics*, 47(21),

746 3835-3845, doi:10.1364/ao.47.003835.

747 Wu, Y., T. Cheng, D. Liu, J. D. Allan, L. Zheng, and H. Chen (2018), Light Absorption

748 Enhancement of Black Carbon Aerosol Constrained by Particle Morphology,

749 *Environmental Science & Technology*, 52(12), 6912-6919,

750 doi:10.1021/acs.est.8b00636.

751 Xu, L., et al. (2020), Tracing the evolution of morphology and mixing state of soot

752 particles along with the movement of an Asian dust storm, *Atmospheric Chemistry and*  
753 *Physics*, 20(22), 14321-14332, doi:10.5194/acp-20-14321-2020.

754 Yang, Y., et al. (2021), Stage-resolved in-cloud scavenging of submicron and BC-  
755 containing particles: A case study, *Atmospheric Environment*, 244,  
756 doi:10.1016/j.atmosenv.2020.117883.

757 Yu, H., W. Li, Y. Zhang, P. Tunved, M. Dall'Osto, X. Shen, J. Sun, X. Zhang, J. Zhang,  
758 and Z. Shi (2019), Organic coating on sulfate and soot particles during late summer in  
759 the Svalbard Archipelago, *Atmospheric Chemistry and Physics*, 19(15), 10433-10446,  
760 doi:10.5194/acp-19-10433-2019.

761 Yuan, Q., J. Xu, Y. Wang, X. Zhang, Y. Pang, L. Liu, L. Bi, S. Kang, and W. Li (2019),  
762 Mixing State and Fractal Dimension of Soot Particles at a Remote Site in the  
763 Southeastern Tibetan Plateau, *Environmental Science & Technology*, 53(14), 8227-  
764 8234, doi:10.1021/acs.est.9b01917.

765 Zelenyuk, A., D. Imre, M. Earle, R. Easter, A. Korolev, R. Leaitch, P. Liu, A. M.  
766 Macdonald, M. Ovchinnikov, and W. Strapp (2010), In Situ Characterization of Cloud  
767 Condensation Nuclei, Interstitial, and Background Particles Using the Single Particle  
768 Mass Spectrometer, SPLAT II, *Analytical Chemistry*, 82(19), 7943-7951,  
769 doi:10.1021/ac1013892.

770 Zhang, G., et al. (2017a), The single-particle mixing state and cloud scavenging of  
771 black carbon: a case study at a high-altitude mountain site in southern China,  
772 *Atmospheric Chemistry and Physics*, 17(24), 14975-14985, doi:10.5194/acp-17-14975-  
773 2017.

774 Zhang, G., et al. (2017b), Insight into the in-cloud formation of oxalate based on in situ  
775 measurement by single particle mass spectrometry, *Atmospheric Chemistry and Physics*,  
776 *17*(22), 13891-13901, doi:10.5194/acp-17-13891-2017.

777 Zhang, H., H. Guo, J. Hu, Q. Ying, and M. J. Kleemani (2019), Modeling Atmospheric  
778 Age Distribution of Elemental Carbon Using a Regional Age-Resolved Particle  
779 Representation Framework, *Environmental Science & Technology*, *53*(1), 270-278,  
780 doi:10.1021/acs.est.8b05895.

781 Zhang, R., A. F. Khalizov, J. Pagels, D. Zhang, H. Xue, and P. H. McMurry (2008),  
782 Variability in morphology, hygroscopicity, and optical properties of soot aerosols  
783 during atmospheric processing, *Proceedings of the National Academy of Sciences of the*  
784 *United States of America*, *105*(30), 10291-10296, doi:10.1073/pnas.0804860105.

785 Zhang, X., and M. Mao (2020), Radiative properties of coated black carbon aerosols  
786 impacted by their microphysics, *Journal of Quantitative Spectroscopy & Radiative*  
787 *Transfer*, *241*, 106718, doi:10.1016/j.jqsrt.2019.106718.

788 Zhang, Y., et al. (2018), Direct Observations of Fine Primary Particles From Residential  
789 Coal Burning: Insights Into Their Morphology, Composition, and Hygroscopicity,  
790 *Journal of Geophysical Research-Atmospheres*, *123*(22), 12964-12979,  
791 doi:10.1029/2018jd028988.

792 Zhuang, B. L., L. Liu, F. H. Shen, T. J. Wang, and Y. Han (2010), Semidirect radiative  
793 forcing of internal mixed black carbon cloud droplet and its regional climatic effect  
794 over China, *Journal of Geophysical Research-Atmospheres*, *115*,  
795 doi:10.1029/2009jd013165.



796 **References From the Supporting Information**

797 Brasil, A. M., T. L. Farias, and M. G. Carvalho (1999), A recipe for image  
798 characterization of fractal-like aggregates, *Journal of Aerosol Science*, 30(10), 1379-  
799 1389, doi:10.1016/s0021-8502(99)00026-9.

800 Oh, C., and C. M. Sorensen (1997), The effect of overlap between monomers on the  
801 determination of fractal cluster morphology, *Journal of Colloid and Interface Science*,  
802 193(1), 17-25, doi:10.1006/jcis.1997.5046.

Figure1.

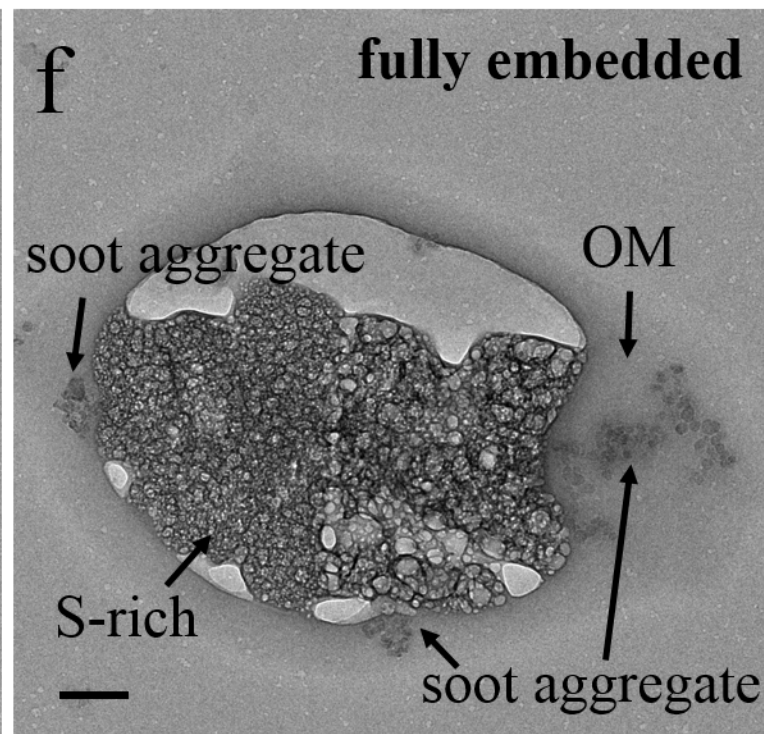
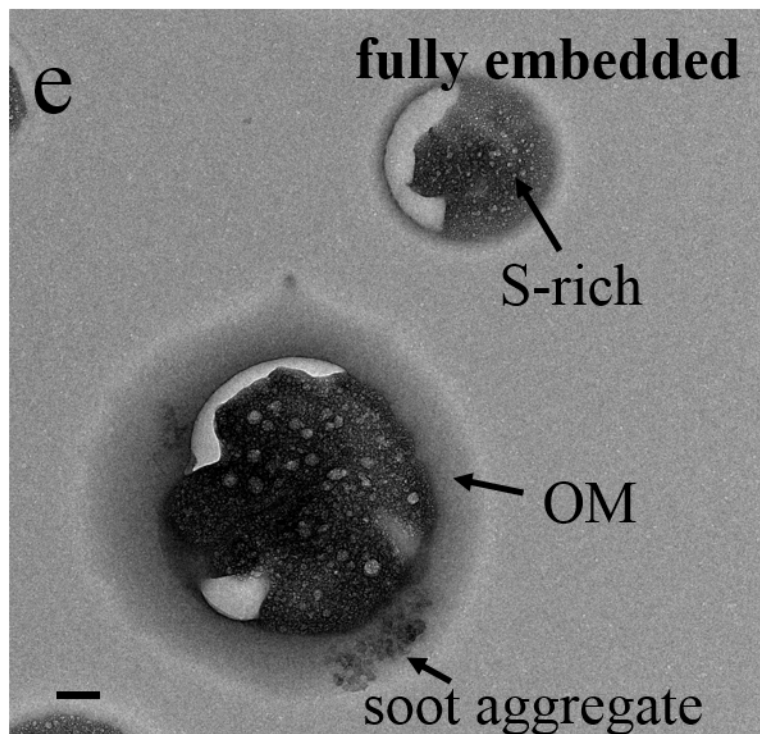
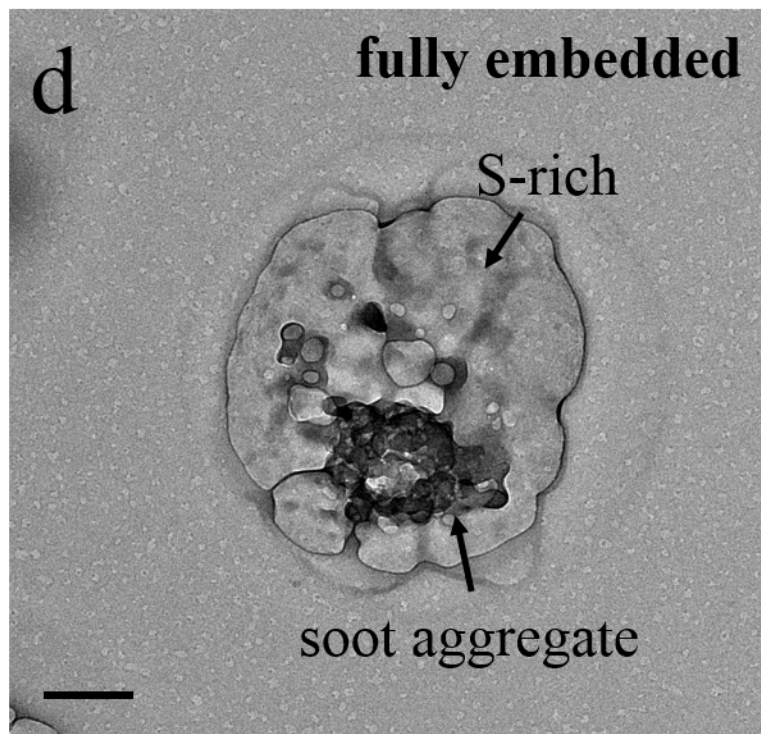
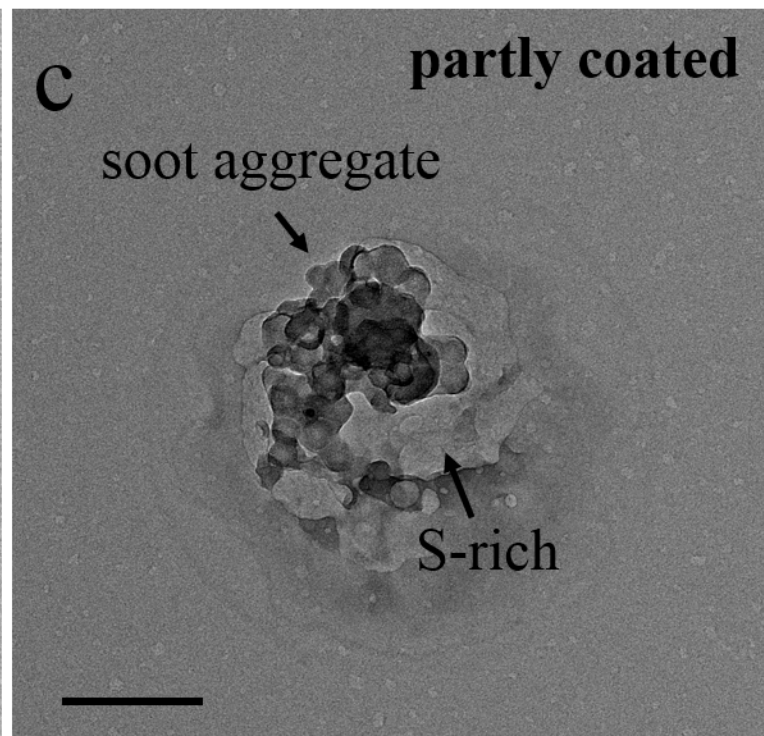
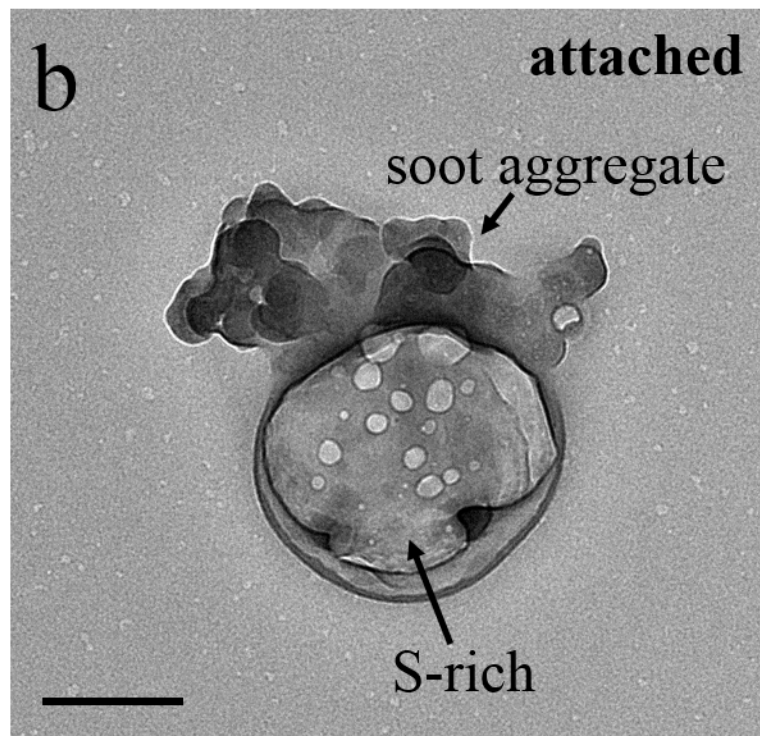
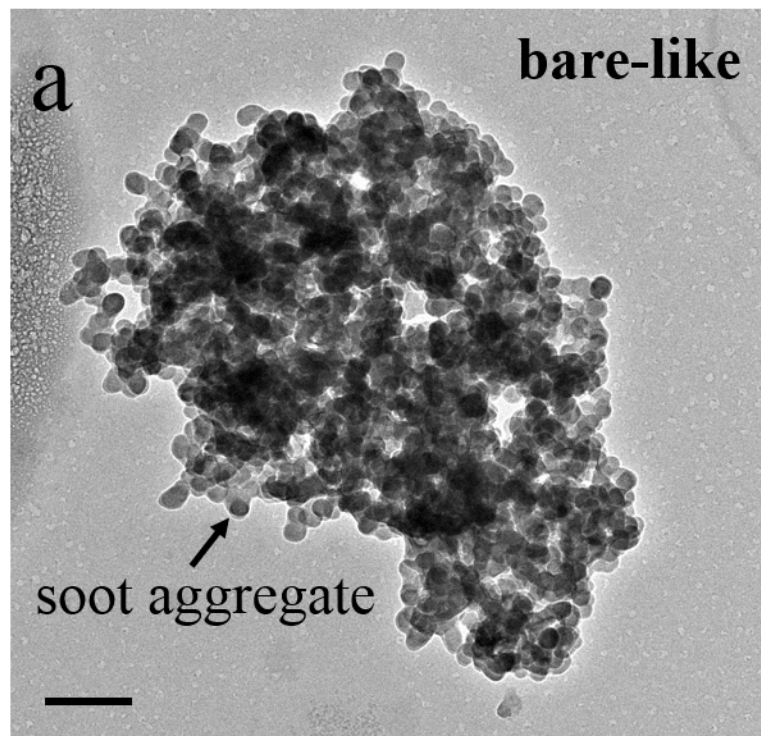
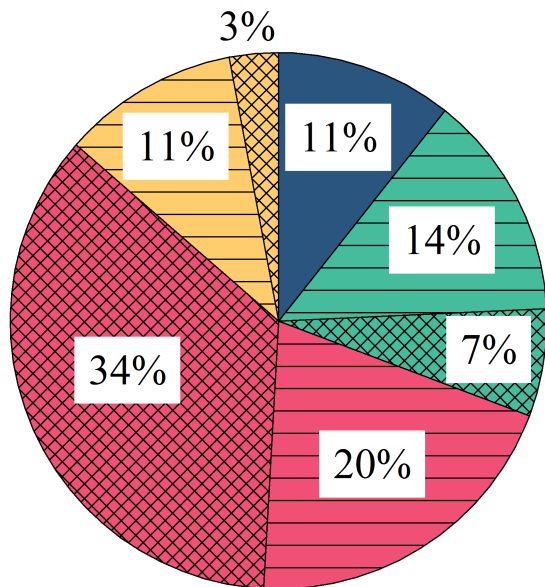


Figure2.

a) pre-cloud



b) in-cloud

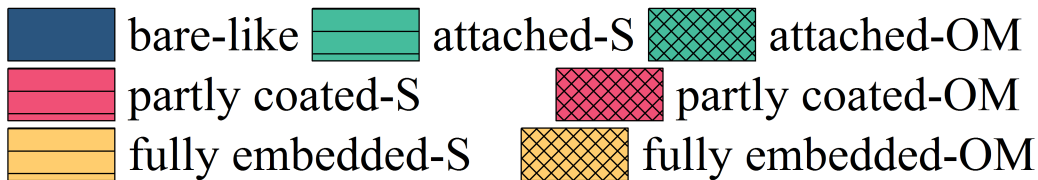
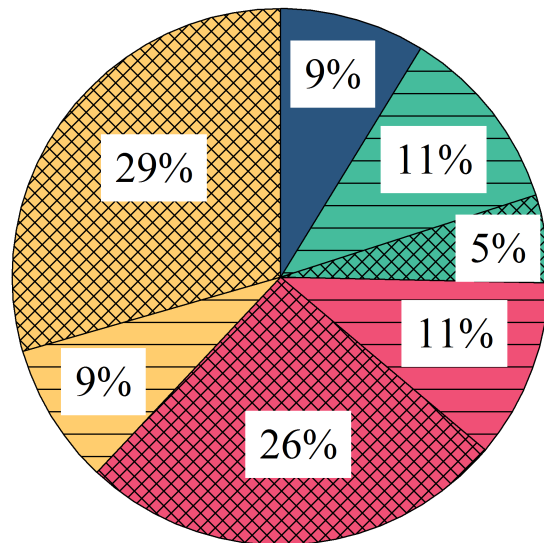
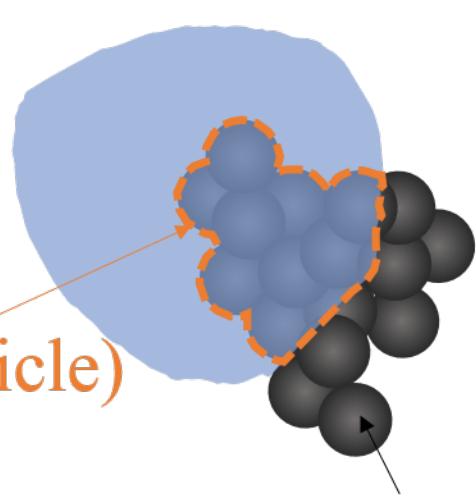
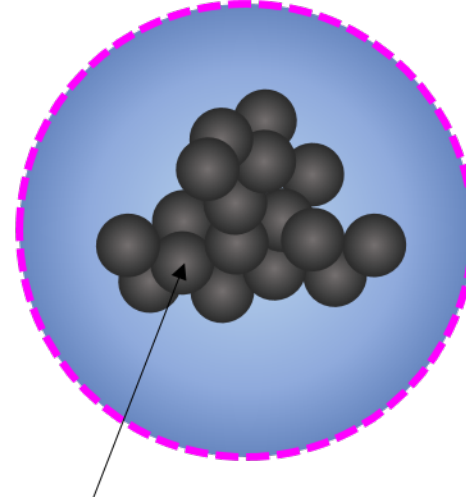


Figure3.

$A_{coated}$  (the projected area of soot aggregate embedded in the host particle)



$A_a$  (the projected area of soot aggregate)



$A_p$  (the projected area of particle)

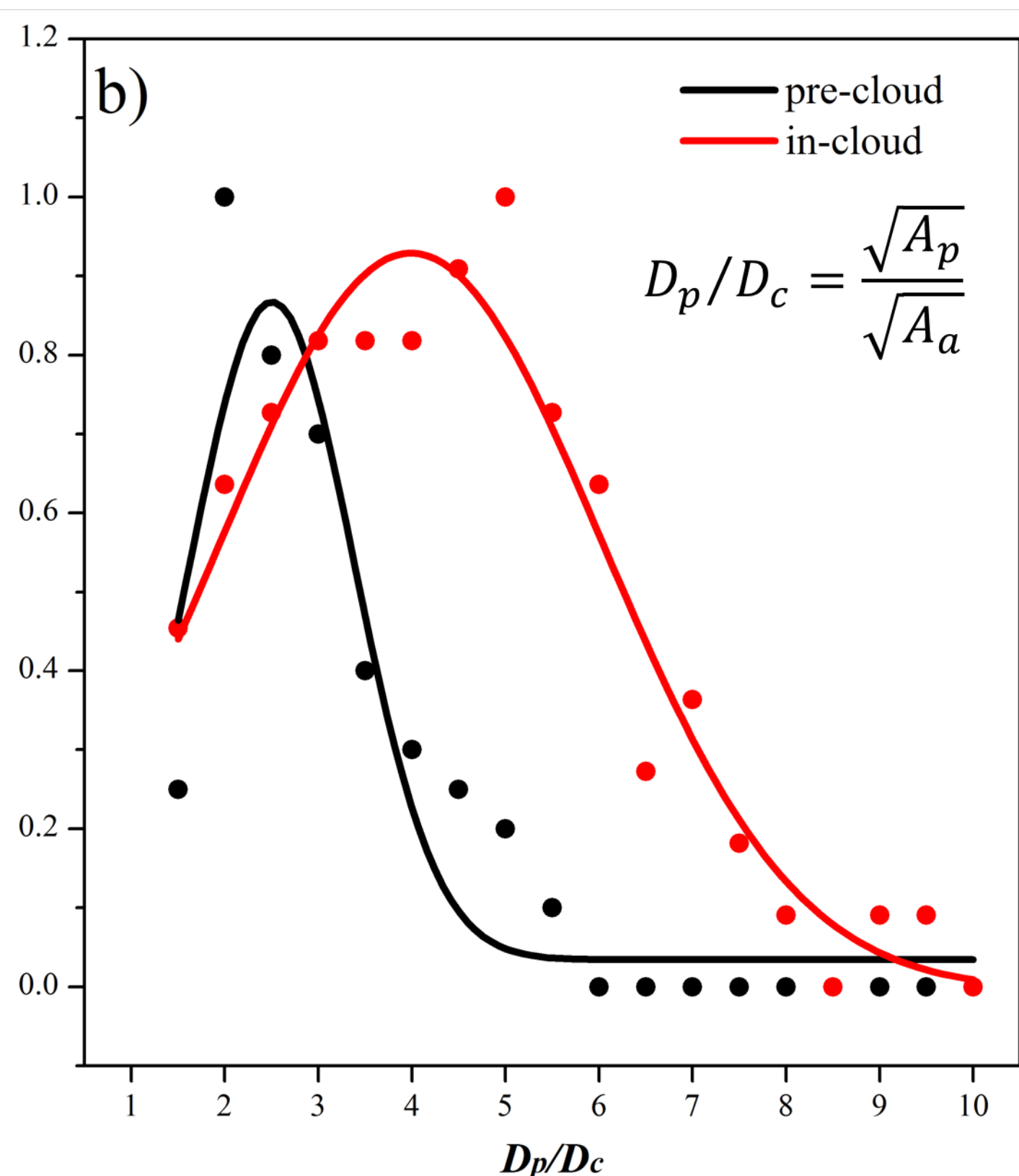
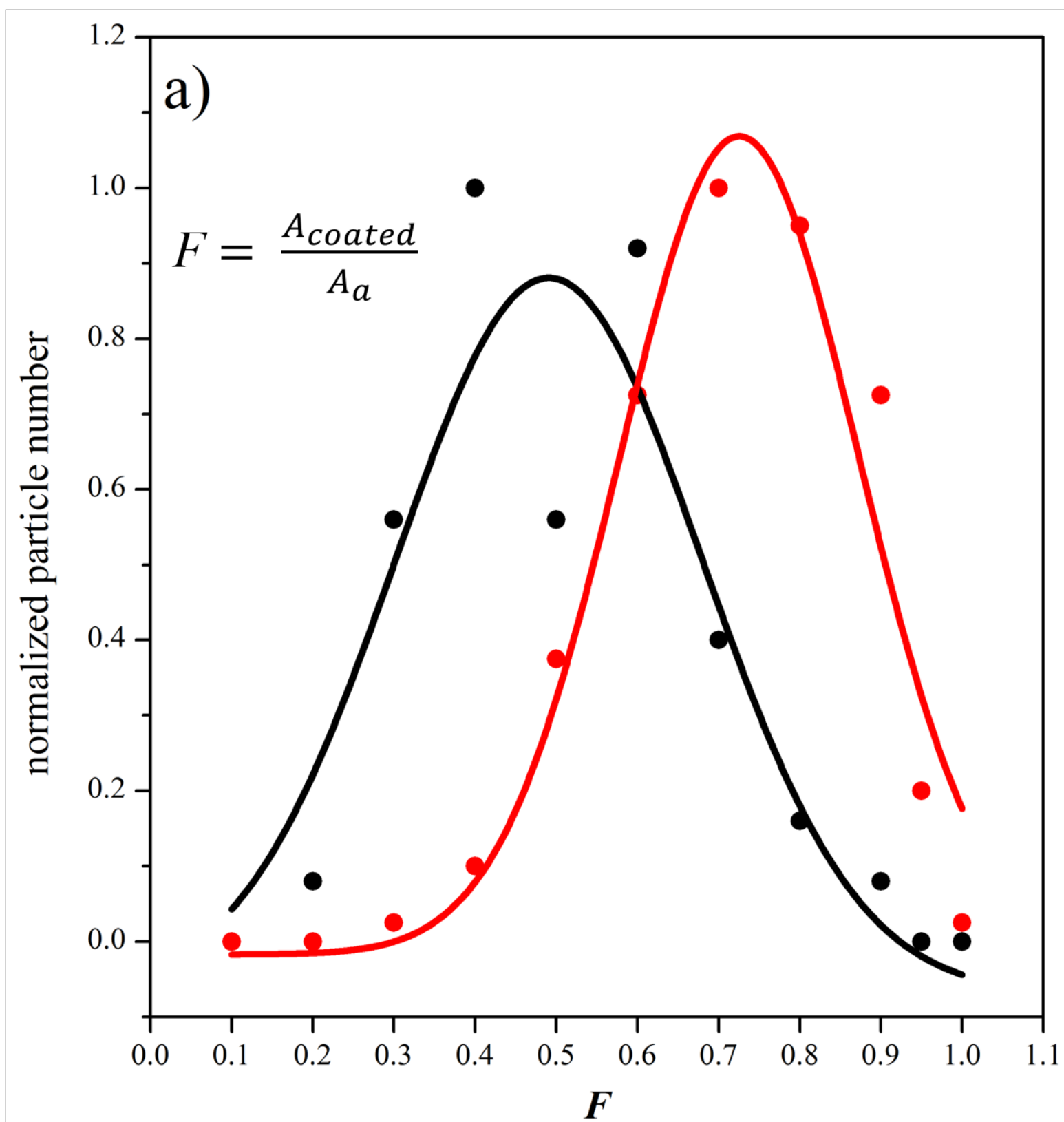


Figure4.



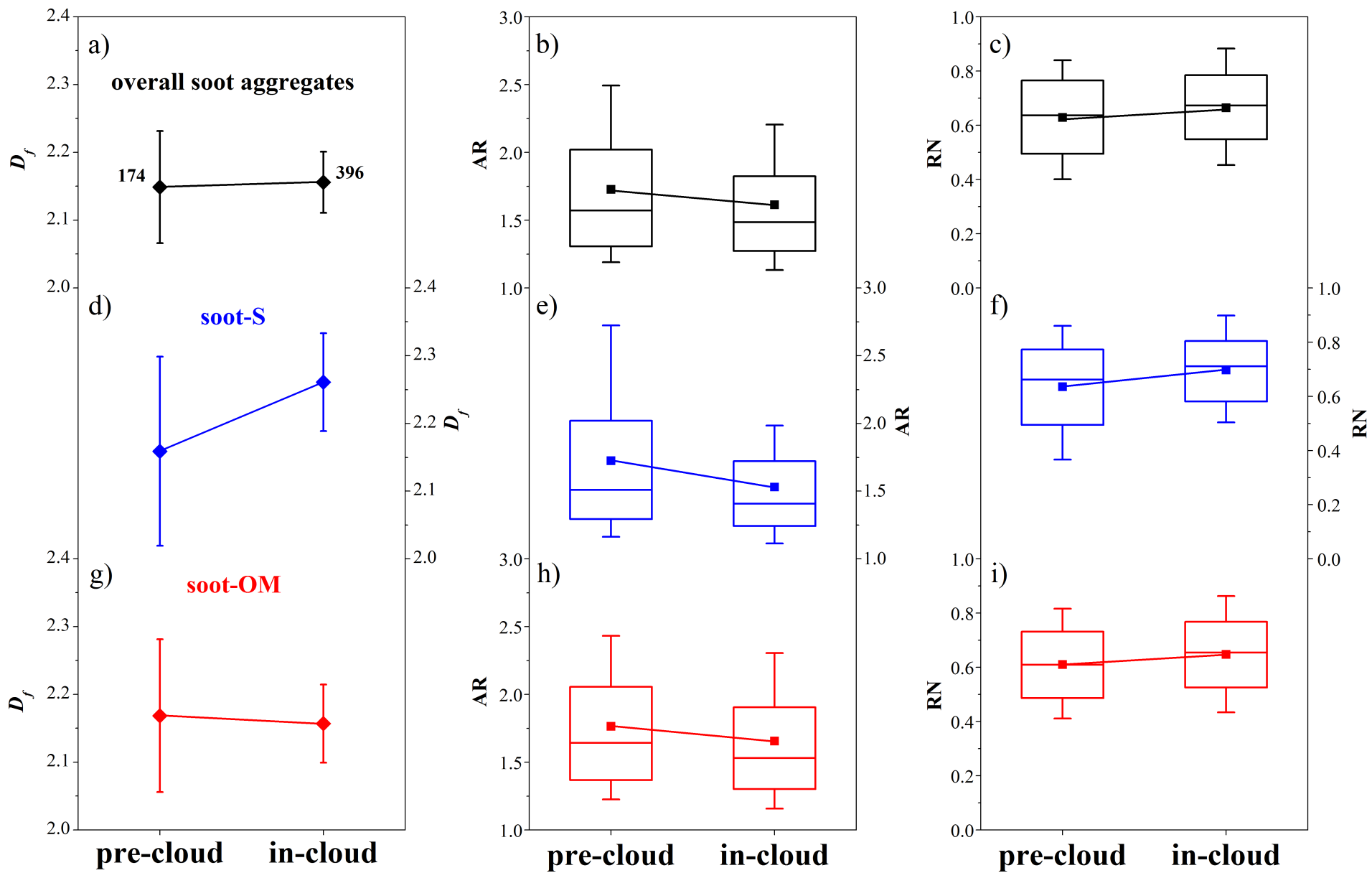
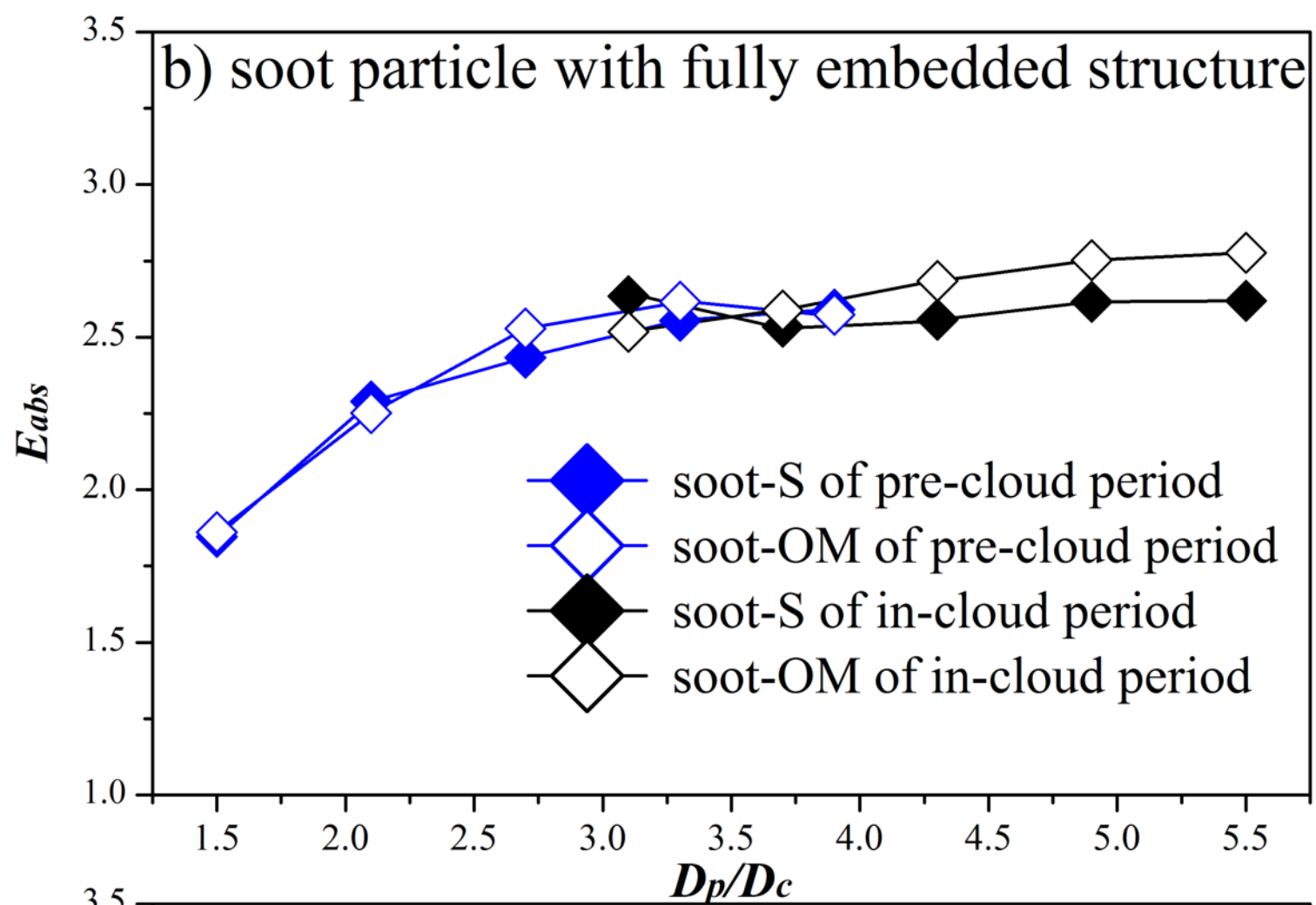
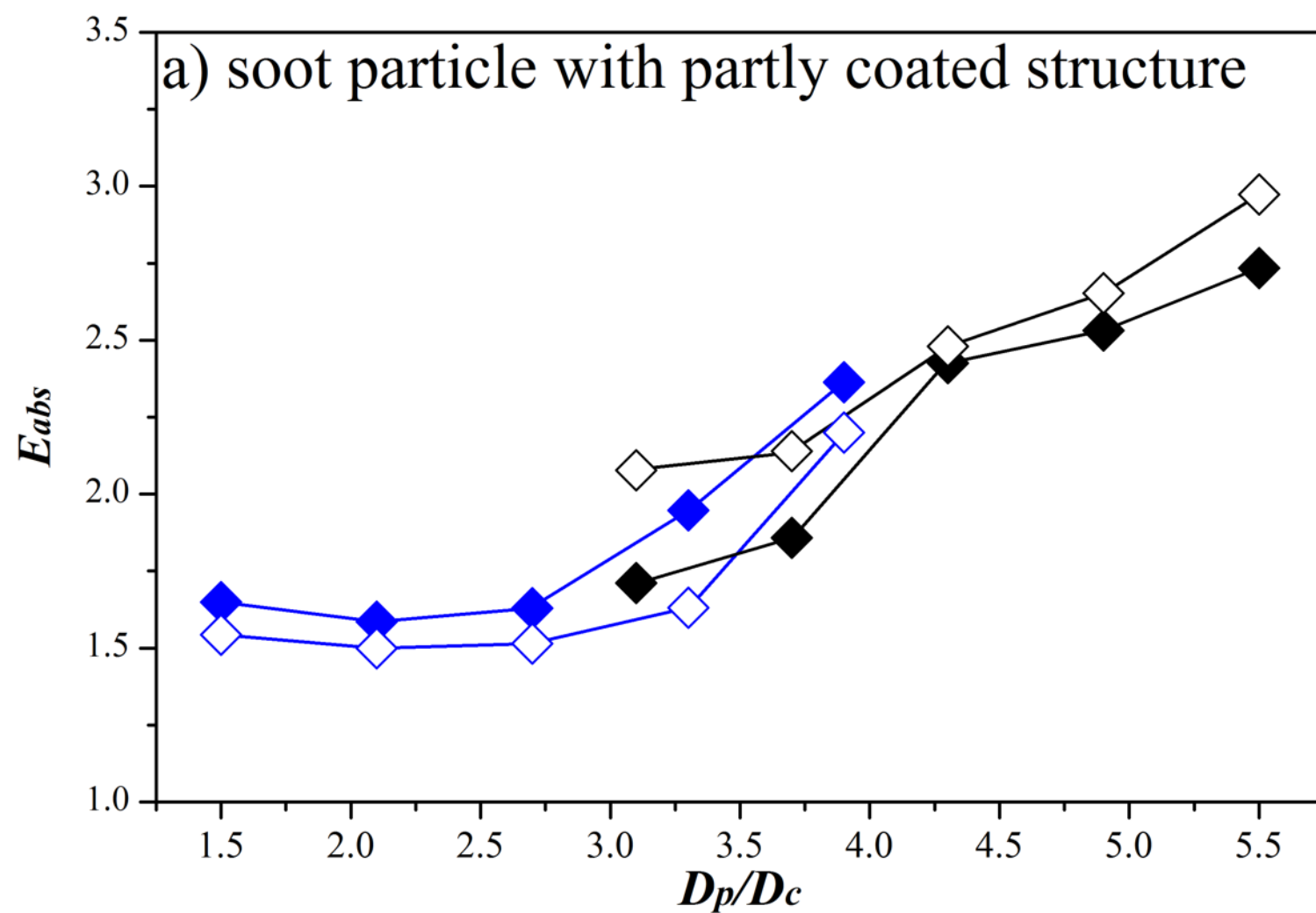
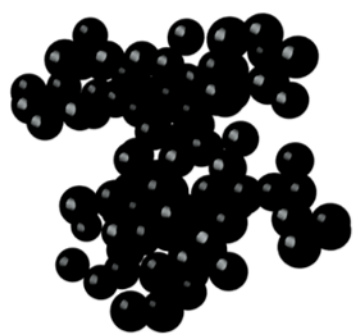


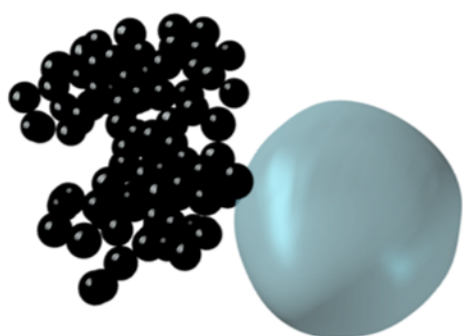
Figure5.



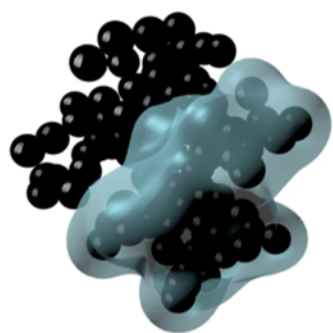
c) shape models



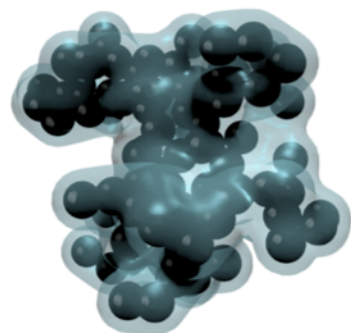
bare-like



attached



partly coated



fully embedded

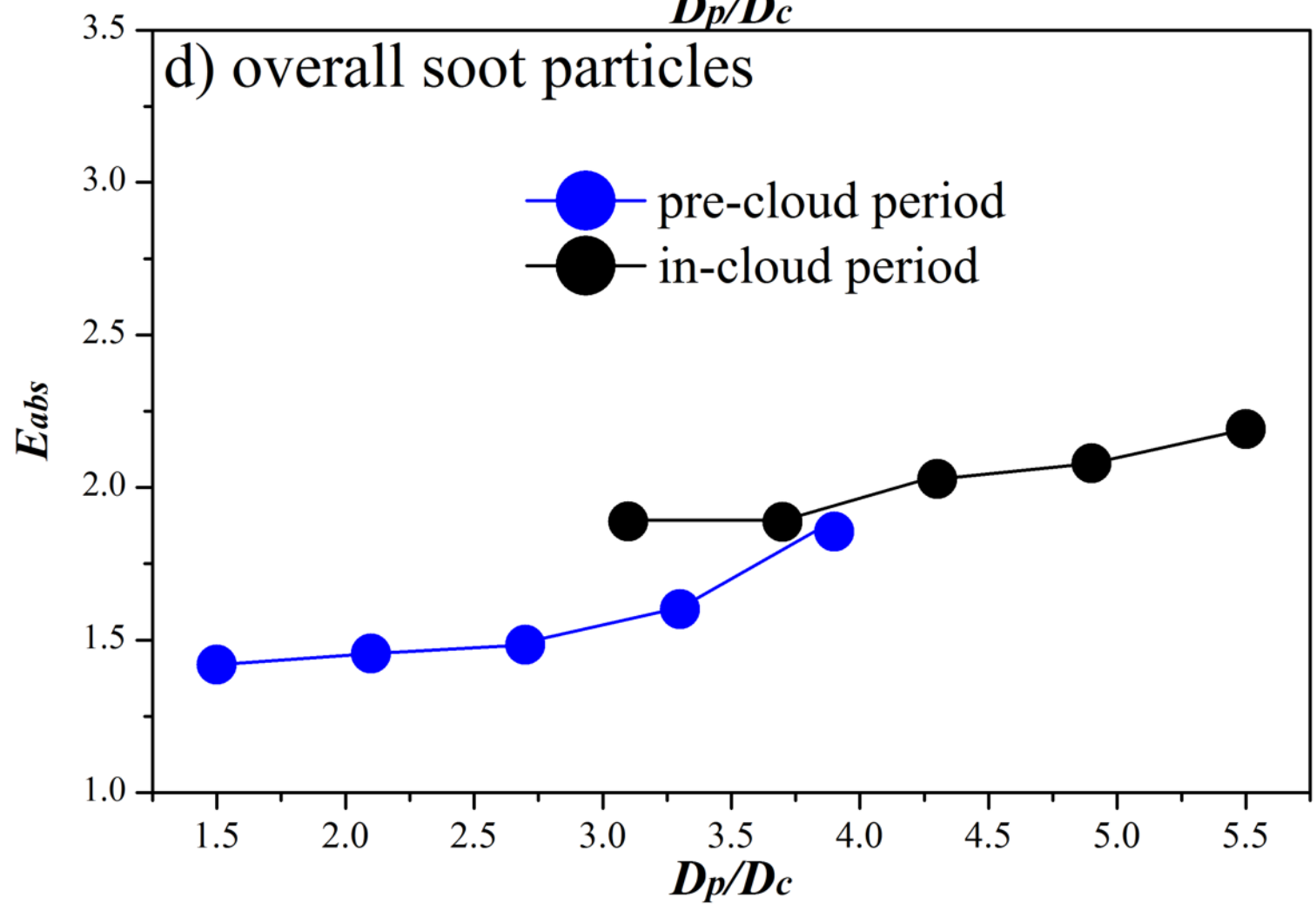


Figure6.

