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Growing rice (Oryza sativa) aerobically reduces phytotoxicity, uptake, and transformation of CeO₂ nanoparticles

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1 Growing rice (Oryza sativa) aerobically reduces

2 phytotoxicity, uptake and transformation of CeO₂

3 nanoparticles

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21 ABSTRACT

22 This study compared the impact and uptake of root administered CeO₂ NPs in rice growing under 23 flooded and aerobic soil conditions, which are two water regimes commonly used for rice cultivation. 24 CeO₂ NPs at 100 mg/kg improved photosynthesis and plant growth by reducing the oxidative 25 damage and enhancing plant tolerance to stress, while higher concentration (500 mg/kg) of CeO_2 26 NPs negatively affected plant growth. More significant effects were observed under the flooded 27 condition than the aerobic condition. CeO₂ NPs of 100 and 500 mg/kg resulted in 78% and 70% 28 higher accumulation of Ce in shoots under the flooded condition compared to the aerobic condition. 29 CeO₂ NPs partially transformed to Ce(III) species in soils and plants under both conditions. Higher 30 extent of transformation under the flooded condition, which was partly attributed to the lower soil 31 pH and redox potential under the flooded condition, leads to higher plant uptake of Ce. A higher 32 extent of transformation in rhizosphere soil was observed. Higher plant transpiration rate under 33 flooded conditions resulted in higher accumulation of CeO₂ species in shoots . This study for the first 34 time reported that water regimes influenced the biotransformation of CeO₂ NPs and their uptake 35 and impacts in rice plant.

36 **Keywords**: CeO₂ nanoparticles; plant; transformation; uptake; water regime; rhizosphere

- 37 Synposis: Growing rice aerobically reduces the transformation of CeO₂ nanoparticles and uptake of
- 38 Ce species in plants, leading to less phytotoxicity.

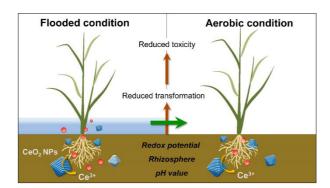


Table of content graphic

41 **INTRODUCTION**

42 Terrestrial ecosystems represent a major sink of engineered nanomaterials (ENMs) after 43 their entry into the environment, where plants represent the largest interface in the system. CeO₂ NPs may not only impair plant growth,¹ but also accumulate in plant organs² and fruit³ and transfer 44 to higher trophic levels of living organisms,⁴ posing risks to human health and environmental safety⁵. 45 46 Despite the adverse effects reported, the application of CeO₂ NPs for improving plant growth was 47 also explored recently. For example, CeO₂ NPs showed potential to improve plant growth under stress conditions such as heat⁶ and high salinity.⁷ A more recent study suggested that CeO₂ NPs can 48 49 capture reactive oxygen species (ROS) such as OH•⁻ in leaves and activate the K⁺ efflux channels in 50 mesophyll cells thereby improving the photosynthesis efficiency and carbon assimilation in 51 *Arabidopsis thaliana*.⁸ The beneficial effects of CeO₂ NPs are mainly derived from their well-known redox properties.⁹ Cerium has two oxidation states, Ce³⁺ and Ce⁴⁺, which alternate during redox 52 reactions. This property enables the reversible stoichiometric storing and oxygen release as cerium 53 54 changes oxidation states, enabling CeO₂ NPs to become a ROS scavenger. However, the biological 55 effects of CeO₂ NPs on plant can be biphasic and need to be understood further to ensure safety as well as beneficial application. 56

It has been known that the effects of CeO₂ NPs on plant growth are highly related to the physicochemical properties of the NPs^{10, 11} and plant species¹² as well as the growth environment¹³ of the plant. Smaller-sized¹² or negatively charged¹⁰ CeO₂ NPs may enter plant more easily thus causing higher toxicity. Different plant species (e.g., monocot and dicot) have different vascular structures and transpiration rates, which affects the uptake and translocation of CeO₂ NPs¹⁴ and subsequent biological effects in plants. Hydroponic, soil or sand cultivated plants may show great variance in their responses to CeO₂ NPs exposure.¹ Studies have shown that biotransformation of 64 CeO₂ NPs is one of the critical mechanisms responsible for the different uptake and effects of CeO₂
 65 NPs in plants.¹⁵

66 ENMs are highly dynamic and tend to transform in soil-plant systems, driven by various 67 physical, chemical or biological factors.¹⁶ Despite being highly stable, CeO₂ NPs can transform after interaction with plants, and the transformation mainly occur at the plant root interface,¹⁷ where 68 69 rhizosphere composition such as organic acids and reducing substances play vital roles in the transformation process.¹⁵ Small-sized CeO₂ NPs are prone to transform more easily and release 70 more Ce³⁺ than larger ones, which can cause phytotoxicity depending on the sensitivity of the plant 71 72 species to the NPs and their transformation species.¹² Rod shaped CeO₂ NPs containing a higher 73 amount of reactive facets on the crystal surface showed higher reactivity and thus a higher extent of transformation than octahedral or cubic CeO₂ NPs.¹¹ Changes in the composition of the culture 74 medium, e.g., presence of phosphate¹⁸ and organic acids,¹ can also affect the transformation 75 76 process. Previous studies also suggests that plant species can affect the transformation of CeO₂ NPs due to differences in the rhizosphere chemistry.¹⁴ Despite these studies, there is still a lack of 77 78 systematic understanding of CeO₂ NPs transformation in complex soil environment as well as in 79 different agricultural practice scenarios (e.g., under various water regimes) and the impact of these 80 on the subsequent uptake and effect in plants.

Rice is a semi-aquatic plant species. Irrigated rice cultures, i.e. culturing rice under flooded condition, account for about 75% of the global yield of rice grain.¹⁹ Due to the high demand for water resources of irrigated rice, aerobically cultured rice is an increasingly important strategy to reduce water use and enhance water use efficiency.²⁰ The different water regimes used for the two cultures may result in significantly different behaviours and impacts of contaminants in the rice plants due to the change of the chemical and biological environments around the roots,²¹ which have been reported for contaminants such as heavy metals (e.g., Zn,²² Cd,²³ Hg²⁴) but have not yet been studied for ENMs. The behaviour and impact of CeO₂ NPs in rice under different water regimes
 is currently unknown.

90 In this study, we hypothesized that changing water regime may affect the transformation, 91 uptake, translocation and subsequent impacts of CeO₂ NPs in rice plants. Specifically, we 92 investigated the impacts of increasing CeO₂ NPs concentrations on the growth of rice either in 93 flooded or aerobic soils by measuring the phenotypic and physiological parameters including 94 biomass, seedling elongation, photosynthesis, and organic and inorganic nutrients. Stress status and 95 plant tolerance to the stress were evaluated by measuring components of the antioxidant system, 96 DNA damage and proline levels in the plants. Uptake and transformation of CeO₂ NPs in soil and 97 plants were measured and correlated with the observed biological effects.

98

99 MATERIALS AND METHODS

100 Chemicals

101 CeO₂ NPs (< 25 nm) were purchased from Sigma Aldrich. The primary size of the CeO₂ NPs 102 characterized by transmission electron microscopy (TEM, JEOL, Japan) were 28 ± 13 nm (**Figure S1**). 103 The crystal structure was cubic fluorite as verified by X-ray diffraction (XRD, D8 Advance, Bruker, 104 USA) (**Figure S2**). The CeO₂ NPs have a hydrodynamic size (intensity) of 151 ± 23 nm and zeta 105 potential of 18 ± 6 mV in deionized water, as analysed using a Zetasizer Nano ZS90 (Malvern, UK). 106 The zeta potential was converted from electrophoric mobility using Smoluchowski equation. Other 107 chemicals were purchased from Sigma Aldrich.

108 Soil preparation, plant cultivation and treatments

109 Sandy loam soil was collected from a local area in Beijing. The soil was air-dried and sieved through

a 5 mm mesh. Soil characteristics were measured and are provided in Table S1. CeO₂ NPs powders

111 were mixed homogeneously into the soil to achieve final concentrations of 100 and 500 mg/kg CeO₂ 112 NPs which are considered as low and high concentrations in this study. The concentrations were comparable with many existing plant studies in soil.^{25, 26} In addition, rare earth elements are not 113 114 rare, the average abundance of Ce in crust is 60 mg/kg and up to 900 mk/kg in soils nearby the rare 115 earth element industry²⁷. Indirect input of Ce into soils due to application of phosphate fertilizers 116 which contain rare earth elements also leads to increasing soil Ce contents. The Ce contents reported in soils ranged from 15-350 mg/kg in United states.²⁸ A rhizobag pot system was used for 117 plant cultivation following a previously described procedure.²⁹ In order to study the rhizosphere soil, 118 a rhizobag made of 40 μ m nylon mesh with a size of 10 x 10 cm² (D X H) was used. 0.3 kg soil 119 120 amended with CeO₂ NPs was placed in each rhizobag and then transferred to a 2L light proof PVC 121 pot. Another 0.7 kg amended soil was used to fill the gap between the rhizobag and the pot, with 122 the total soil being 1 kg in each pot. Soil without CeO₂ NP amendment was used as control. Three 123 replicates were set up for each treatment. The schematic show of the rhizobag experiment was 124 provided in Figure S3.

125 Rice seeds (Oryza sativa) were purchased from the Chinese Academy of Agricultural Science. The 126 seeds were sterilized in 5% H₂O₂ for 20 min and rinsed thoroughly with deionized water. The seeds 127 were germinated in moist sand at 25 °C in the dark and grew for 10 days. Uniform seedlings were 128 then selected and carefully transferred into the rhizobags with three seedlings per bag. In the 129 flooded group, deionized water was supplemented to full saturation capacity of the soil and 130 replenished every day to a standing water level of ~ 2 cm. In the aerobic group, deionized water was 131 added to the soil to reach 70% of the water holding capacity (21%) of the soil. Soil Eh was measured 132 at 6 cm below the surface on the day of exposure and 7, 14, 21 and 28 days after the exposure.

133 Impact of CeO₂ NPs on photosynthetic system and transpiration

134 The relative chlorophyll contents were measured on days 0, 7, 14, 21 and 28 using a portable SPAS-135 502 Plus. The first fully expanded leaf of each plant with five points on each leaf was measured. The optimum quantum yield (F_v/Fm) and photochemical efficiency of PSII (ФPSII) was measured 136 following a previously described method.³⁰ Briefly, the plants after 28 days of exposure were placed 137 138 in the dark and the minimum fluorescence (F0) in the plant leaves was measured on a portable 139 photosynthesis measurement system (Li-6400, LI-COR, USA). A pulse of white light was applied and 140 the maximal fluorescence (Fm) was assessed. The plants were then exposed to saturating light (50 141 μ mol m⁻² s ⁻¹ for 3 min) to establish the maximal fluorescence (F_m') after light adaption. The minimal 142 fluorescence (F₀') was then measured after turning off the actinic light. The Fv/Fm of PSII was 143 calculated as $(F_m-F_0)/F_m$ and the Φ PSII was calculated as $(F_m' - F')/F_m'$. The transpiration rate of 144 plants were measured on the portable photosynthesis measurement system (Li-6400, LI-COR, USA).

145 **Protein, sugar and starch contents in plant samples**

The plants were harvested after 4 weeks of exposure. Roots and shoots were separated and rinsed in 0.01M HNO₃ and deionized water, and the fresh weight and root/shoot lengths were measured. Soluble protein contents in the samples were measured using a BCA assay kit (Thermo Scientific, USA) following the manufacturer's instructions.

Total soluble sugar content was measured following a previously described method.¹ Briefly, 0.5g samples were homogenized in 80% ethanol and centrifuged at 2000 rpm for 20 min. The supernatant was mixed with 5% phenol and 98% sulphuric acid followed by incubation in a water bath (30 °C) for 20 min. The absorbance of the solution was then measured at 490 nm using a UVvis spectrometer. Glucose solutions with known concentrations were used as standards to calculate the sugar concentration in the samples.

Reducing sugar content was quantified according to a previous method described by Miller et al.³¹
Briefly, 0.5g dry samples were homogenized in 80% ethanol and centrifuged at 2000 rpm for 20 min.

The supernatant was mixed with 3,5-dinitrosalicylic acid and boiled for 5 min. The absorbance was measured at 515 nm using a UV-vis spectrometer (LAMBDA 365, Perkin Elmer, USA). Glucose solutions with known concentrations were used as standards to calculate the sugar concentration in the samples.

The starch content was measured using a method described by McCready et al.³² The residues after centrifugation for the extraction of total soluble sugar were re-suspended in deionized water. Perchloric acid was added and the mixture was centrifuged at 2000 rpm for 20 min. The supernatant was diluted by 10-fold with deionized water and processed following the same protocol as for total soluble sugar. Starch contents were quantified using glucose as standard with a factor of 0.9 applied for conversion of glucose to starch.

168 Content of Ce, macro- and micro- nutrient elements in plant samples

169 To measure the elemental content, fresh roots and shoots were separated and washed with 0.01M HNO₃ and deionized water (three times) to remove contaminants³³. lyophilized. Dry samples were 170 171 digested with a mixture of HNO₃/H₂O₂ (v/v ratio, 3:1) on a heating plate (80 °C for 1 h, 120 °C for 3 h, and 160 °C for another 0.5 h). The residues were diluted with deionized water for measurement. 172 173 Ce contents were measured by inductively coupled plasma mass spectrometry (ICP-MS, Thermo X7, 174 USA). Ce standard solution (0.1~10 mg/L) was used for external calibration. K, Ca, Mg, P, Fe, Cu, Zn 175 and Mn were measured by inductively coupled plasma optical emission spectrometry (ICP-OES, 176 Perkin Elmer, USA). Multi-element standard solutions (0.5~50 mg/L) containing the selected 177 elements were used for external calibration. Blanks were analysed between every six samples. 178 Spiking recovery experiments and analysis of certified reference material (GBW 07602 Bush 179 Branches and Leaves) were performed for analytical method validation. The recoveries and limits of 180 detection for all the elements are reported in Table S2.

181 Antioxidant enzyme activities, proline content and 8-OHdG content

The activities of the antioxidant enzymes including superoxide dismutase (SOD) and catalase (CAT), and the content of malondialdehyde (MDA) in plant tissues were measured using the relevant assay kits according to the manufacturer's instructions (Nanjing Jiancheng Biotechnology Co., Ltd).

Proline content was measured following the method described by Bates et al.³⁴ Briefly, fresh roots and leaves were homogenized with cooling by an ice bath and diluted with 3% sulfosalicylic acid to 10 mL followed by centrifugation at 1500 g for 10 min. The supernatant was collected and mixed with acidic ninhydrin and glacial acetic acid (100%) and allowed to react in boiling water for 1 h followed by addition of methylbenzene. The absorbance of the extractives was measured at 520 nm and the concentration was calculated based on calibration curve established with proline standard solutions.

The content of 8-hydroxy-2 deoxyguanosine (8-OHdG) was measured to determine the DNA oxidative damage. Fresh roots and leaves were ground in liquid nitrogen into powders followed by homogenization in a buffer solution containing 2% CTAB, 100 mM Tris-HCl, 2% Polyvinylpyrrolidone (PVP), 1.5 mM NaCl and 0.2 mM EDTA and 1% mercaptoethanol. DNA in the mixtures was then extracted using the DNA kit (Qiagen, France), and the 8-OHdG content was quantified using a DNA Damage ELISA kit (Stress Marq Biosciences, Inc, Thermo, USA).

198 Transmission electronic microscopy (TEM) observation of root sections

Fresh root apexes were cut and prefixed overnight in 2.5% glutaradehyde solution in phosphate buffered saline (PBS, pH 7.4). The samples were then washed with PBS three times followed by dehydration in gradient acetone and embedding in Spurr's resin. Ultrathin sections (90 nm) were obtained using an UC6i ultramicrotome (Lecia, Austria). To avoid artifacts, the sections were not stained by uranyl acetate. The sections were collected on copper TEM grids and observed on a JEM-1230 (JEOL, Japan) transmission electron microscope.

205 X-ray absorption spectroscopy (XAS)

206 Samples in treatments with 500 mg/kg CeO₂ NMs were used for XAS analysis. Rhizobag soils and pot 207 soils were collected followed by lyophillization in a freeze-dryer. Fresh root and shoot samples were 208 washed with deionized water and lyophilized. The samples were motor homogenized and pressed 209 into thin pellets for XAS analysis. The XAS spectra were collected on beamline 1W1B at the Beijing 210 Synchrotron Radiation Facility. The energy of the storage ring during the data collection was 2.5 GeV 211 with current intensity of 50 mA. CeL_{III}-edge spectra of the soil and root samples were collected in 212 transmission mode while the shoot samples were collected in fluorescence mode using a 19-213 element germanium array solid detector. The XAS spectra of CePO₄ and Ce(CH₃COO)₃ as well as 214 CeO₂ NPs were collected as standards. ATHENA software was used to perform the data analysis 215 including normalization, energy calibration and linear combination fitting (LCF) analysis. 216 **Data analysis** 217 Data were expressed as mean ± standard deviation (SD) (n=8). Statistical analysis was performed on 218 IBM SPSS 19.0. One-way ANOVA and student t-test were used to evaluate the significance between

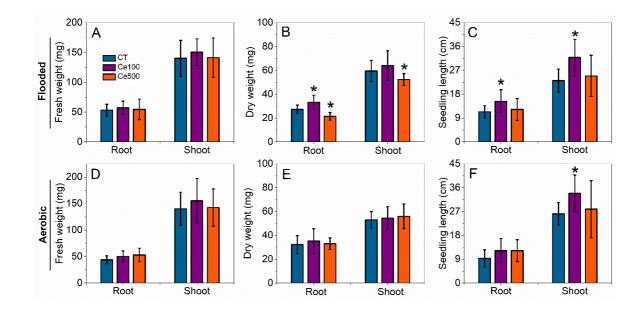
219 data. P < 0.05 was considered significantly different.

220

221 **RESULTS AND DISCUSSION**

Different impacts of CeO₂ NPs on rice phenotype and photosynthesis under aerobic and flooded conditions

224 Under flooded conditions, CeO₂ NPs treatment did not significantly affect the fresh weight of rice 225 seedlings (**Figure 1A**) compared to the untreated controls, while the dry weight of the roots 226 increased by 22% at the concentration of 100 mg/kg and reduced the dry weight of roots and shoots 227 at 500 mg/kg by 14% and 34%. CeO₂ NPs at 100 mg/kg enhanced the root and shoot elongation by 228 35% and 38%, respectively. While under aerobic conditions, CeO₂ NPs showed no significant impacts on the fresh and dry biomass and the seedling length, with the exception of shoot length which was



230 increased by 29% following 100 mg/kg CeO₂ NPs treatment.

Figure 1. Fresh weight (A and D), dry weight (B and E) and seedling length (C and F) of rice under
 flooded (A-C) and aerobic (D-F) conditions. Asterisk (*) indicates significant difference compared
 to untreated control at P < 0.05.

235

231

236 The impacts of CeO_2 NPs on the photosynthesis of the rice plants were evaluated by measuring the 237 relative chlorophyll content, PSII and Φ PSII. The effects of CeO₂ NPs on the relative chlorophyll 238 content in leaves were different under the two conditions (Figure 2A). The relative chlorophyll 239 content was increased by 40% following 100 mg/kg CeO₂ NPs treatment while decreased by 20% at 240 500 mg/kg under flooded conditions. However, CeO₂ NPs showed no effects on the relative 241 chlorophyll content under aerobic conditions. The Fv/Fm is a commonly used indicator of stress in 242 leaves since stress leads to damaged PSII with lowing of Fv/Fm.³⁵ The Fv/Fm was reduced by 17% following treatment with 500 mg/kg CeO₂ NPs (Figure 2B), suggesting that the plant was in stress 243 244 and the PSII was impaired. Although CeO₂ NPs of 100 mg/kg did not affect the Fv/Fm, they enhanced 245 the ΦPSII by 11% and 19% under flooded and aerobic conditions, respectively (Figure 2C). CeO₂ NPs at 500 mg/kg reduced the ΦPSII by 18% under flooded conditions while had no effects under aerobic
conditions. These results suggest that CeO₂ NPs improved plant photosynthesis at 100 mg/kg while
impairing the photosynthetic system if the dose was high (500 mg/kg) under flooded conditions.
The results correlated with the results of biomass and seedling elongation, suggesting that the
difference in the impact of CeO₂ NPs on the photosynthesis contributed to the different effects on
plant growth under flooded and aerobic soil conditions.

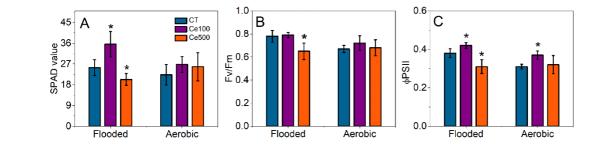


Figure 2. Effects of CeO₂ NPs on the relative chlorophyll content (A), Fv/Fm (B) and Φ PSII (C) in leaves under flooded and aerobic (B) conditions, Asterisk (*) indicates significant difference compared with untreated control at *P* < 0.05.

256

252

257 CeO₂ NPs caused more disturbance of the organic and mineral nutritional contents under

258 flooded condition

259 To further explore the difference in the physiological impact of CeO₂ NPs on rice, we measured the 260 effect on organic (Table 1) and inorganic (Table 2) nutrient contents in plants. Under flooded 261 conditions, CeO₂ at 100 mg/kg significantly increased the organic nutrient contents. Although 100 262 mg/kg CeO₂ NPs did not affect the protein content in roots, it increased the protein content in 263 shoots by 48%. Under flooded conditions, 100 mg/kg CeO₂ NPs also increased the contents of 264 soluble sugar, reducing sugar and starch in both roots and shoots. At high concentration (500 mg/kg), 265 CeO₂ NPs reduced the protein and soluble sugar content in root by 20% and 38%, respectively, but 266 showed no effects on these organic nutrients in shoots (flooded conditions).

267 CeO₂ NPs showed similar patterns of effects on the organic nutrients under aerobic conditions with less significant differences compared with that under flooded conditions. Under aerobic 268 269 conditions, 100 mg/kg CeO₂ NPs increased the starch content in roots by 81% but showed no effects 270 on other organic nutrients in roots compared with control. However, 100 mg/kg CeO₂ NPs increased 271 the contents of all the organic nutrients in shoots. CeO₂ at 500 mg/kg did not affect the organic 272 nutrients in either roots or shoots (aerobic conditions). These results suggest that CeO₂ NPs affected 273 the carbon metabolism of rice seedlings and the effects were different depending on the water 274 irrigation regime. Starch and sugar are two key components involved in carbon flux within most 275 plants. They are produced through fixation of atmospheric carbon via photosynthetic process. 276 Increase in sugar and starch contents provide sufficient energy source for the cellular respiration in 277 order to cope with ambient stress such as excess light, heat and high salinity. The increase of the 278 sugar and starch contents by CeO₂ NPs agree with the results observed for photosynthesis that more 279 profound impacts were observed under flooded conditions than under aerobic condition and that 280 only positive effects on plant growth were oserved under aerobic condition.

281

Table 1. Total soluble protein, total sugar, reducing sugar and starch content in plant roots and shoots (unit: mg/g FW). * indicates significant difference at p < 0.05 compared with the corresponding control. "F" indicates flooded. "A" indicates aerobic.

Treatments		Protein (mg/g FW)	Soluble sugar (mg/g FW)	Reducing sugar (mg/g FW)	Starch (mg/g FW)	
	CT-F	14.2 ± 1.3	18.5 ± 2.9	7.9 ± 2.4	1.9 ± 0.3	
Root	Ce100-F	15.7 ± 2.2	27.5 ± 4.6*	11.9 ± 1.6*	$3.1 \pm 0.4^{*}$	
	Ce500-F	11.4 ± 2.5*	11.5 ± 3.3*	7.1 ± 3.0	2.3 ± 0.2	
	CT-A	12.7 ± 2.5	19.7 ± 2.2	6.1 ± 1.4	1.6 ± 0.5	
	Ce100-A	13.1 ± 3.0	23.3 ± 1.8	8.7 ± 2.9	2.9 ± 0.3*	
	Ce500-A	11.7 ± 2.8	16.5 ± 3.6	8.3 ± 3.7	1.8 ± 0.4	
Shoot	CT-F	8.3 ± 1.6	14.7 ± 2.4	8.3 ± 3.1	1.2 ± 0.2	
	Ce100-F	12.3 ± 3.1*	18.5 ± 1.6*	13.9 ± 3.7*	$2.1 \pm 0.1^{*}$	
	Ce500-F	7.3 ± 1.0	15.1 ± 3.0	11.0 ± 2.4	1.9 ± 0.3	

CT-A	7.5 ± 1.6	12.3 ± 2.2	9.4 ± 1.8	1.4 ± 0.5
Ce100-A	$9.4 \pm 1.9^*$	15.7 ± 1.8*	12.9 ± 2.4*	$2.1 \pm 0.3^{*}$
Ce500-A	6.3 ± 2.5	10.7 ± 3.3	11.8 ± 3.2	1.7 ± 0.4

285

286 CeO₂ NPs did not affect the contents of macronutrients (K, Ca, Mg, P) in shoots, while 100 mg/kg 287 CeO₂ NPs increased the micronutrients including Fe, Zn and Mn in shoots under both flooded and 288 aerobic conditions. Specifically, the Fe, Zn and Mn contents were increased by 29%, 73% and 42%, 289 respectively, under flooded conditions, and were increased by 23%, 36% and 29%, respectively 290 under aerobic conditions. The positive effects of 100 mg/kg CeO₂ on plant growth might be related 291 to the increased Fe and Zn levels because both Fe and Zn are essential nutrients playing significant 292 roles in photosynthetic process. In contrast, 500 mg/kg CeO₂ NPs showed no effects on the content 293 of inorganic nutrients in shoots. Taken together, these results suggest that CeO₂ NPs showed more 294 significant positive impacts at 100 mg/kg or more severe negative impacts at 500 mg/kg on the 295 macro- and micro- nutrient elements uptake and accumulation in rice seedlings under flooded 296 conditions than under aerobic conditions.

297

Table 2. Contents of macro- and micro- element nutrients in plant shoots affected by CeO_2 NPs treatment under flooded and aerobic conditions. *indicates significant difference at p < 0.05 compared with untreated control.

Treatments		Macronutrient (g/kg DW)			Micronutrient (mg/kg DW)				
		К	Са	Mg	Р	Fe	Cu	Zn	Mn
Flooded	СТ	35.6 ± 1.3	1.6 ± 0.1	3.4 ± 0.2	4.4 ± 0.7	95 ± 10	5.2 ± 1.5	11 ± 3	43 ± 9
	Ce100	33.4 ± 2.0	1.7 ± 0.3	3.5 ± 0.3	4.0 ± 0.5	123 ± 14*	4.4 ± 0.9	19 ± 4 *	61 ± 8*
	Ce500	34.1 ± 1.8	1.6 ± 0.3	3.5 ± 0.4	4.2 ± 0.3	101 ± 21	4.1 ± 1.2	15 ± 4	48 ± 7
Aerobic	СТ	38.1 ± 1.8	1.5 ± 0.2	3.7 ± 0.3	4.1 ± 0.5	87 ± 15	4.7 ± 1.3	14 ± 2	45 ± 5
	Ce100	36.7 ± 2.3	1.6 ± 0.2	3.4 ± 0.3	4.3 ± 0.7	107 ± 13*	4.0 ± 1.2	19 ± 3*	58 ± 7*
	Ce500	36.9 ± 3.1	1.6 ± 0.3	3.5 ± 0.2	4.5 ± 0.4	95 ± 19	4.3 ± 2.0	16 ± 3	47 ± 6

301

302 Distinct stress responses in plants induced by CeO₂ NPs under aerobic and flooded conditions

303 To further evaluate the stress induced by CeO₂ NPs exposure, we measured the antioxidant enzyme 304 activity, proline content and the oxidative DNA damage marker (8-OHdG). The response patterns of 305 the antioxidant enzymes and proline in plants to CeO₂ NPs were similar under flooded and aerobic 306 conditions (Figure 3). CeO₂ NPs at 100 mg/kg enhanced the SOD and CAT activities in roots under 307 both flooded and aerobic conditions, with a higher enhancement of CAT under flooded (145%) than 308 aerobic (63%) conditions (Figure 3A-3B and 3E-3F). The SOD and CAT activities in shoots were also 309 enhanced by 100 mg/kg CeO₂ NPs treatment with the exception of SOD under aerobic conditions 310 (Figure 3E). The MDA contents in roots were reduced by 100 mg/kg CeO₂ NPs under both flooded 311 and aerobic conditions (Figure 3C and 3G), while it was increased by 40% following 500 mg/kg CeO₂ 312 NPs treatment. The MDA contents in shoots were not affected under aerobic condition. However, under flooded condition, CeO₂ NPs at 100 mg/kg reduced the MDA contents in shoots by 41%. 313

314 Plants have developed their own defense system against ambient stresses including 315 enzymatic and non-enzymatic antioxidant system. There was no significant difference in the MDA 316 contents between control groups under the two conditions, suggesting that water regime didn't 317 cause stress to plants. The enhancement of SOD and CAT levels and reduced MDA levels in plant 318 treated with 100 mg/kg CeO₂ NPs were observed under both conditions, suggesting that the CeO₂ 319 NPs reduced the stress level in plants by improving the antioxidative capacity of plants. While plants 320 failed to defense the stress in roots caused by high concentration (500 mg/kg) of CeO₂ NPs exposure 321 under flooded condition but succeed under aerobic condition, as shown by the MDA content in 322 plant.

We further examined the proline content in plants, which is a marker for plant stress (**Figure** 324 **3D** and **3H**). CeO₂ NPs at 100 mg/kg reduced the proline content in roots by 33% under both flooded 325 and aerobic conditions. The proline content in shoots was also reduced in the 100 mg/kg CeO₂ NPs 326 treatment under flooded conditions while no change was observed under aerobic conditions. High 328 both flooded and aerobic conditions while it did not affect the proline content in shoots.

concentration (500 mg/kg) of CeO₂ NPs caused enhanced accumulation of proline in roots under

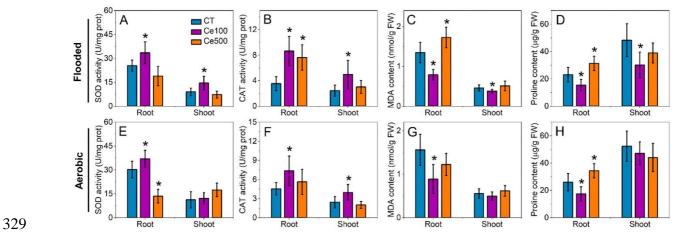
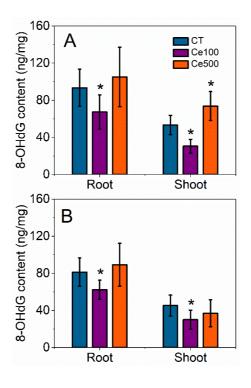


Figure 3. Stress response of rice plants to CeO_2 NPs exposure. (A-D) and (E-H) indicate the SOD activity, CAT activity, MDA content and proline content under flooded and aerobic conditions, respectively. * indicates significant difference at p < 0.05 compared with the corresponding control.



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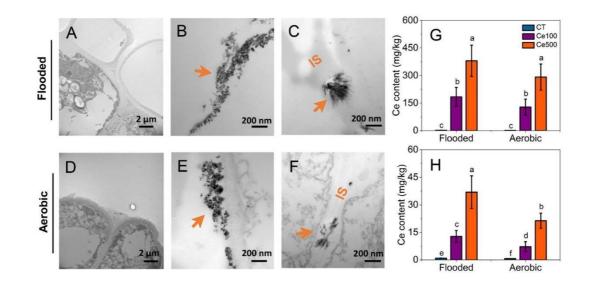
334 **Figure 4.** The content of 8-OHdG in plants affected by CeO₂ NPs under flooded (A) and aerobic (B)

335 conditions. * indicates significant difference at p < 0.05 compared with the corresponding control.

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The biomarker for oxidative DNA damage (8-OHdG) was measured to evaluate whether oxidative stress caused any genetic damage (Figure 4). The 8-OHdG contents in roots and shoots were reduced in the 100 mg/kg treatment under both flooded and aerobic conditions, while CeO₂ NPs of 500 mg/kg increased the 8-OHdG content in shoots under flooded conditions. Overall, our results suggest that 100 mg/kg CeO₂ NPs reduced the plant stress while 500 mg/kg CeO₂ NPs caused stress to plants, and either the positive effects at 100 mg/kg or the negative effects at 500 mg/kg seemed stronger under flooded conditions, compared to aerobic conditions.

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345 Lower uptake of Ce in plants under aerobic than flooded conditions

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Figure 5. TEM images of root sections (A-F) of plants treated with 500 mg/kg CeO₂ NPs and Ce uptake in plants (G and H) treated with 100 and 500 mg/kg CeO₂ NPs. (A) and (D) are TEM images of roots in control group under flooded and aerobic conditions, respectively. (B) and (E) show the root surface. (C) and (E) show the intercellular space (IS). Arrows indicate the particles observed on root surface and in IS. (G) and (H) are the Ce contents in roots and shoots, respectively. Different lowercase letters indicate significant difference between treatments at P < 0.05.

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Uptake of CeO₂ NPs in plants was first examined by TEM observation of root sections (**Figure 5A**-**5F**). There were no particles observed in control groups (**Figure 5A** and **5D**). CeO₂ NPs aggregates were observed along the root surface under both flooded and aerobic conditions (**Figure 5B** and **5E**). Needle-like clusters were observed in the intercellular space (**Figure 5C** and **5F**). Similar clusters were previously found in CeO₂ NPs treated cucumber, which were demonstrated to be CePO₄.¹⁵ No difference in the CeO₂ NP uptake between flooded and aerobic conditions can be distinguished from the TEM images.

Ce uptake in plants was quantified by ICP-MS and dose-dependent Ce uptake in plants was observed (Figure 5G and 5H). The Ce contents in roots were not significantly different between the two conditions at the same exposure concentration (Figure 5G). However, the Ce contents in shoots treated with 100 mg/kg and 500 mg/kg CeO₂ NPs under flooded condition were 1.78 and 1.70 fold higher than those under aerobic conditions.

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367 Less transformation of CeO₂ NPs in soil and plants under aerobic than flooded conditions

368 Transformation of CeO₂ NPs in soil and plants was analysed by synchrotron based XANES. As seen 369 from Figure 7, the low-energy feature (a) and high-energy features (b and c) are respectively 370 attributed to the Ce(III) and Ce(IV) compounds. This spectral difference is an important criterion for 371 distinguishing Ce compounds of the two different oxidation states. The Ce L_{III}-edge XANES spectra 372 in all the samples presented mainly the feature (b and c) of Ce(IV) while also showing a small peak 373 (a) which is the feature of Ce(III), suggesting that transformation occurred. The fractions of CeO₂ 374 NPs and its transformed products were obtained by LCF analysis using the XANES spectra of CeO₂ 375 NPs, Ce acetate (Ce(Ac)₃) and CePO₄ as standard references (Figure S4).

Figure 6A showed the results in rhizobag soils. A larger bump at the position of peak a in
 rhizobag soils under flooded condition than under aerobic condition can be clearly seen, suggesting

378 more transformation of CeO₂ NPs occurred under flooded condition. LCF analysis revealed 11% and 379 17% of Ce(III) species in rhizobag soils under aerobic and flooded conditions, respectively (Figure 380 **6A**). In contrast, the Ce presented mainly as CeO_2 in pot soils, with only 3% and 4% of Ce(III) species 381 obtained by LCF (Figure 6B). A previous study by Ma et al. demonstrated that XANES and LCF analysis 382 are very sensitive for distinguishing Ce species in plant matrix with error between 2% - 6%.³⁶ 383 Therefore, the 3% or 4% reported here are within the error that can be negligible. The significantly higher Ce(III) species found in rhizobag soils than in pot soils suggest that rhizosphere chemistry 384 385 played a significant role in the transformation of CeO₂ NPs. The rhizosphere, which contains large 386 amounts of root excreted organic acids and reducing substances as well as abundant 387 microorganisms, contributes significantly to the reduction and dissolution of CeO₂ NPs and their 388 further transformation such as phosphorylation. The rhizobag soils are close to the root thus 389 containing larger amount of root exudates than pot soils, therefore, more transformation of CeO₂ 390 NPs can occur.

391 Higher fractions of Ce(III) (15% under aerobic and 28% under flooded conditions) were 392 observed in root samples (Figure 6C) than in soil samples at the same condition, suggesting more transformation occurred in roots. The root surface has been reported to be the main site for 393 394 transformation of CeO₂ NPs in plants.¹⁷ A study by Ma et al. found that CeO₂ NPs that entered into 395 plants remained as CeO₂ and did not transform further inside plant. Rico et al. found more 396 transformation of CeO₂ NPs in hotspots on the root surface by synchrotron XRF/XANES than in soils. 397 Our results are in accordance with the previous reports that roots played a critical role in CeO₂ NPs transformation. 398

The higher fraction of CePO₄ than that of Ce(AC)₃ in rhizobag soil can be attributed to the stronger affinity of Ce(III) for PO₄³⁻ (Ksp=1 × 10⁻²³) than for the organic ligand. Reducing substance and organic acids in the root exudates are the dertemining factors that stimulate the reduction and release of Ce³⁺. The Ce³⁺ can further bind with phosphates on the root surface or in the intercellular 403 space after entering the root¹⁵. Phosphates alone cannot lead to obvious transformation as 404 comparable with that caused by root exudates, which has been demonstrated in a previous study, although they can bind with Ce(III) on the particle surface³⁷. Therefore, more fractions of Ce(III) 405 406 including CePO₄ and Ce carboxylates formed in rhizobag soils than in pot soils. Since CePO₄ 407 accounted for the major part of the transformed species and the CePO₄ are needle-like clusters with 408 low mobility, the upward translocation of Ce(III) was therefore restricted. As shown in Figure 6D, 409 the majority of the Ce presented as CeO₂ in shoots under both aerobic and flooded conditions, with 410 only 5% and 9% of Ce(Ac)₃ being detected, respectively, and no evidence of CePO₄.

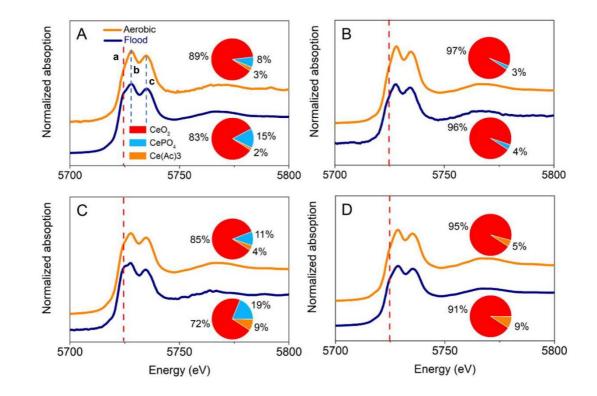


Figure 6. Ce L_{III} -edge XANES spectra and the fraction of Ce species (shown as pie charts) in rhizobag soil (A), pot soil (B), root (C) and shoot (D) samples collected from the 500 mg/kg CeO₂ NPs treatment groups. The dotted lines indicate the location of the characteristic peak of Ce(III) (a) and Ce(IV) (b and c) species.

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417 The absolute contents of Ce(IV) and Ce(III) species were calculated by multiplying the total Ce 418 content by the percentages of Ce(IV) and Ce(III). As shown in Figure 7A, the Ce(IV) content in roots 419 under flooded (274 mg/kg) and aerobic (249 mg/kg) conditions were close, while the Ce(III) content 420 in roots was higher under the flooded condition (106 mg/kg) than under the aerobic condition (43 421 mg/kg). The contents of Ce(IV) and Ce(III) in shoots were both higher under flooded than aerobic 422 conditions (Figure 7B). The different transformation and translocation of CeO₂ NPs in soils and 423 plants under different conditions can be explained by the following reasons. Firstly, the redox 424 potential (Eh) was much lower in flooded soil (-188 mV) than in aerobic soil (-90 mV), and the Eh 425 remained almost constant during the experiment (Figure 7C). Low Eh is favourable for the reduction 426 of cerium from Ce(IV) to Ce(III). Moreover, flooding can cause variation of the soil pH, i.e. decrease 427 of the pH of alkaline soil and increase of pH of acidic soil. As expected, the pH of flooded soil (5.1 ± 428 0.7) was lower than that of aerobic soil (7.7 \pm 0.6). A lower pH is favourable for the dissolution of 429 CeO₂ NPs and the release of Ce³⁺. The lower Eh and pH caused higher transformation of CeO₂ NPs 430 and subsequent translocation of Ce(III) species. The change of pH affects the surface charge of CeO₂ 431 NPs, which can affect the uptake in plant. Previous studies suggest that the positive CeO₂ NPs adhere more to plant roots and translocate less into aboveground shoot than negative or neutral CeO2 432 433 NPs.^{38, 39} However, the surface charge of the CeO₂ NPs used here was nerutral at pH 7.8 (pH at 434 aerobic condition) while was positive (21 mV) at pH 5 (pH at flooded condition) (Figure S5), which is 435 not in accordance with the previous reports, suggesting that surface charge is not the main driver 436 leading to the difference between flooded and aerobic condition. Secondly, inorganic ions and NPs 437 share the same vascular system with the water and nutrients and translocate upwards from root to 438 shoot following the water flow.² Therefore, the upward translocation of CeO₂ NPs or Ce³⁺ is highly 439 dependent on the transpiration rate. Plants can adjust the opening or closure of the stomata on the 440 leaves in response to different water irrigation regimes.⁴⁰ In flooded conditions, plants usually have 441 a high transpiration rate. However, in aerobic conditions, plant reduce their transpiration rate by 442 restricting the opening of stomata.⁴⁰ As shown in **Figure 8**, the transpiration rate of rice was significantly higher in the flooded condition than in the aerobic condition for all groups. Under 443 444 flooded condition, 100 mg/kg CeO2 NPs enhanced the transpiration rate (TR) by 38% at day 28, 445 while 500 mg/kg CeO2 NPs reduced the TR by 30%. Under aerobic condition, 100 mg/kg CeO2 NPs 446 enhanced the TR at day 14 and day 28 by 54% and 30%, respectively. The higher mobility of both particles or ions with water flow under flooded condition than aerobic condition led to high uptake 447 and translocation of Ce in plants. It should be noted that in Fe²⁺ rich paddy soil, iron plaque can 448 449 easily form on rice root surface. Previous studies suggest that iron plaque reduces the uptake of CeO₂ NPs in hydroponic conditions^{41, 42}, which is not in accordance with our results that Ce uptake 450 451 was enhanced in flooded condition under which the iron plaque may form. The inconsistence may 452 be related to that the difference in the actual amount of iron plaque on the root can cause 453 contrasting effects. For example, it was reported that medium amount iron plaque increased Zn 454 uptake while high amount of iron plaque reduced the Zn uptake in plant⁴³. In these hydroponic 455 studies, iron plaque was intentionally created by adding additional FeSO₄ as Fe source. The iron 456 plaque formed in our study might be not significant enough thus its impacts on the Ce uptake are 457 not comparable with the root exudates which contribute significantly to the release of Ce³⁺ and 458 subsequent transformation. Moreover, while iron plaque can act as a physical barrier for root 459 absorption of foreign substances, positive effects on the absorption and translocation of metals such as Zn, Pb, and Hg in plants have been reported⁴⁴. More studies are required to gain further 460 insights into the effects of iron plaque on the transformation and uptake of CeO₂ NPs in plants 461 growing in realistic fields under different conditions (e.g., water regime, soil iron content, pH etc.). 462

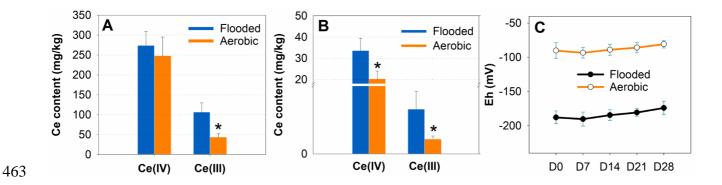
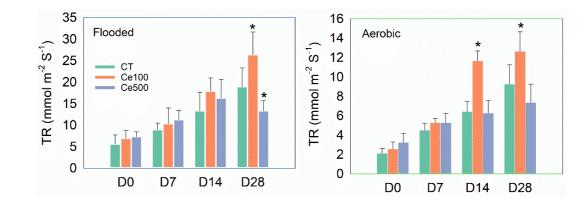


Figure 7. Ce(IV) and Ce(III) contents in roots (A) and shoots (B) calculated by multiplying the total Ce content by the percentages of Ce(IV) and Ce(III). (C) Eh of the soil in flooded and aerobic conditions at 0, 7, 14, 21 and 28 days. * indicates significant difference compared with untreated control at P < 0.05.

468



470 Figure 8. Transpiration rate of rice plant growing under flooded and aerobic ocnditions with CeO2
471 NPs treatments. * indicates significant difference compared with untreated control at *P* < 0.05.

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473 Previous studies have suggested that the phytotoxicity of metal-based NPs such as CeO_2 NPs is 474 highly linked with their transformation. For example, the phytotoxicity of CeO_2 NPs to *Lactuca* plants 475 at germination stage was attributed mainly to the dissolution of CeO_2 NPs; a small amount of Ce^{3+} 476 released at the root interface can significantly inhibit the seedling growth. Another study suggested 477 particle-specific effects of CeO_2 NPs to *Arabidopsis thaliana*.⁴⁵ More recently, a life cycle study 478 showed that CeO_2 NPs and Ce^{3+} showed different and even opposing effects on the soil grown 479 common bean (*Phaseolus vulgaris*); CeO₂ NPs impaired the photosynthesis and reduced the yield while Ce³⁺ enhanced the photosynthesis and improved the nutrition quality of pods, suggesting that 480 481 the effects of CeO₂ NPs on plants cannot be solely explained by the ionic effect.³ In agreement with this study, our results also suggest that both CeO₂ NPs and Ce³⁺ played distinct roles in the effects 482 483 on rice plants. CeO₂ NPs caused more significant impacts on the growth of rice seedlings under 484 flooded conditions than under aerobic conditions, which correlated well with the higher uptake and 485 translocation of Ce in the plants. At 100 mg/kg, CeO₂ NPs showed stronger positive effects under 486 flooded conditions than under aerobic conditions. As discussed previously, CeO₂ NPs are capable of 487 capturing ROS thereby reducing oxidative stress in plants and improving photosynthesis and plant growth.⁸ In addition to the CeO₂ NPs, the positive effects of low dose Ce³⁺ on plant growth have 488 489 been long known, although the mechanisms are poorly understood. One possible mechanism that 490 has been proposed is that Ce³⁺ can act as a substitute for Mg²⁺ to improve photosynthesis.⁴⁶ Indeed, 491 the photosynthesis was enhanced by 100 mg/kg CeO₂ NPs treatment and higher improvement was 492 observed under flooded conditions corresponding to the higher Ce(III) content in the shoots. Therefore, both CeO₂ and Ce³⁺ contributed to the positive effects at 100 mg/kg. In contrast, the high 493 494 concentration of CeO₂ (500 mg/kg) caused negative impacts on rice growth, and the impacts 495 correlated again with the total Ce and Ce(IV)/Ce(III) contents in the plants. More significant negative 496 impacts were observed under flooded conditions than under aerobic conditions, which 497 corresponded to the higher Ce(IV) and Ce(III) contents in the rice plants.

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499 Environmental Implications and future perspectives

Taken together, these results suggest that the water irrigation regime influences the impact of CeO₂ NPs on photosynthesis, antioxidant system, plant stress, DNA damage and eventually the plant growth, by changing the local environmental in soil thus affecting the CeO₂ transformation and Ce 503 uptake and translocation in rice plants. It should be noted that CeO₂ NPs are a type of redox sensitive 504 NM, and thus whether changing water irrigation regimes can also influence the effects of other 505 ENMs on plant growth needs to be explored in the future. Even for ENMs that are not sensitive to 506 redox potential, the shift of redox potential can affect not only the soil chemistry, but also the 507 microbial activity and root rhizosphere chemistry that may significantly change the metal 508 bioavailability.

The positive effects on plant growth of CeO_2 NPs at 100 mg/kg imply the potential of CeO_2 NPs for application in agriculture. Indeed, CeO_2 NPs have shown potential to improve plant growth under stress caused by excess heat and light⁶, high salinity⁷, nitrogen deficiency or excess⁴⁷ maily owing to its ROS scavenge property, while high dose of CeO_2 may also pose risk to plant growth.

513 Under realistic field condition, ENMs may co-exist with other conventional pollutants such as As, Hg 514 and Cd. It has been known that whereas aerobic condition may be beneficial for reducing 515 bioavailability of As²⁴, this may also lead to increase of the bioavailability of Cd⁴⁸, known to be 516 present in high quantities in many paddy soils with potential risks for food safety and human health. 517 The mutual effects of CeO₂ NPs and Cd or As has been reported^{49, 50}. Although no effects on the Cd 518 or As accumulation was exerted by CeO₂ NPs, the uptake of Ce in plant was enhanced by the 519 presence of Cd or As. Note that these studies were performed in hydrponic and dry soils. More 520 studies are needed to gain insights into whether and how variation of redox potential due to the 521 change of water regime affects the mutual effects between these chemicals and ENMs.

522

523 SUPPORTING INFORMATION

TEM image of CeO₂ NPs (Figure S1); XRD patterns of CeO₂ NPs (Figure S2); Schematic show of the rhizobag system (Figure S3); Ce L-_{III} edge XANES spectra of standard reference samples including CeO₂, CePO₄ and Ce carboxyaltes (Figure S4); Zeta potential of CeO₂ NPs as a function of pH

- 527 (Figure S5); Properties of the soil used in this study (Table S1); Limit of detection, precision and
- 528 recovery data of ICP-OES for the selected elements (Table S2).

529 AUTHOR INFORMATION

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537 CONFLICT OF INTEREST

538 The authors declare no conflict of interest.

539 **REFERENCE**

540 1. Zhang, P.; Ma, Y.; Liu, S.; Wang, G.; Zhang, J.; He, X.; Zhang, J.; Rui, Y.; Zhang, Z.,

541 Phytotoxicity, uptake and transformation of nano-CeO₂ in sand cultured romaine lettuce. *Environ.*

542 *Pollut.* **2017,** *220,* 1400-1408.

543 2. Zhang, Z.; He, X.; Zhang, H.; Ma, Y.; Zhang, P.; Ding, Y.; Zhao, Y., Uptake and distribution of
544 ceria nanoparticles in cucumber plants. *Metallomics* **2011**, *3*, (8), 816-822.

545 3. Ma, Y.; Xie, C.; He, X.; Zhang, B.; Yang, J.; Sun, M.; Luo, W.; Feng, S.; Zhang, J.; Wang, G.,

546 Effects of Ceria Nanoparticles and CeCl3 on Plant Growth, Biological and Physiological Parameters,

- 547 and Nutritional Value of Soil Grown Common Bean (Phaseolus vulgaris). *Small* **2020**, 1907435.
- 548 4. Ma, Y.; Yao, Y.; Yang, J.; He, X.; Ding, Y.; Zhang, P.; Zhang, J.; Wang, G.; Xie, C.; Luo, W.,
- 549 Trophic transfer and transformation of CeO2 nanoparticles along a terrestrial food chain: Influence
- 550 of exposure routes. *Environ. Sci. Technol.* **2018**, *52*, (14), 7921-7927.

551 5. Gardea-Torresdey, J. L.; Rico, C. M.; White, J. C., Trophic transfer, transformation, and

impact of engineered nanomaterials in terrestrial environments. *Environ. Sci. Technol.* 2014, 48,
(5), 2526-2540.

6. Wu, H.; Tito, N.; Giraldo, J. P., Anionic cerium oxide nanoparticles protect plant

555 photosynthesis from abiotic stress by scavenging reactive oxygen species. ACS Nano 2017, 11,

556 (11), 11283-11297.

7. Rossi, L.; Zhang, W.; Ma, X., Cerium oxide nanoparticles alter the salt stress tolerance of
Brassica napus L. by modifying the formation of root apoplastic barriers. *Environ. Pollut.* 2017, 229,
132-138.

560 8. Wu, H.; Shabala, L.; Shabala, S.; Giraldo, J. P., Hydroxyl radical scavenging by cerium oxide

561 nanoparticles improves *Arabidopsis* salinity tolerance by enhancing leaf mesophyll potassium

562 retention. *Environ. Sci.: Nano* **2018,** *5*, (7), 1567-1583.

563 9. Dhall, A.; Self, W., Cerium oxide nanoparticles: a brief review of their synthesis methods
564 and biomedical applications. *Antioxidants* **2018**, *7*, (8), 97.

565 10. Liu, M.; Feng, S.; Ma, Y.; Xie, C.; He, X.; Ding, Y.; Zhang, J.; Luo, W.; Zheng, L.; Chen, D.,

566 Influence of Surface Charge on the Phytotoxicity, Transformation, and Translocation of CeO2

567 Nanoparticles in Cucumber Plants. ACS Appl. Mat. Interfaces **2019**, *11*, (18), 16905-16913.

568 11. Zhang, P.; Xie, C.; Ma, Y.; He, X.; Zhang, Z.; Ding, Y.; Zheng, L.; Zhang, J., Shape-dependent
569 transformation and translocation of ceria nanoparticles in cucumber plants. *Environ. Sci. Technol.*570 *Lett.* 2017, 4, (9), 380-385.

571 12. Zhang, P.; Ma, Y.; Zhang, Z.; He, X.; Li, Y.; Zhang, J.; Zheng, L.; Zhao, Y., Species-specific

572 toxicity of ceria nanoparticles to Lactuca plants. *Nanotoxicology* **2015**, *9*, (1), 1-8.

573 13. Cui, D.; Zhang, P.; Ma, Y.; He, X.; Li, Y.; Zhang, J.; Zhao, Y.; Zhang, Z., Effect of cerium oxide
574 nanoparticles on asparagus lettuce cultured in an agar medium. *Environ. Sci.: Nano* 2014, 1, (5),
575 459-465.

576 14. Zhang, P.; Ma, Y.; Xie, C.; Guo, Z.; He, X.; Valsami-Jones, E.; Lynch, I.; Luo, W.; Zheng, L.;

577 Zhang, Z., Plant species-dependent transformation and translocation of ceria nanoparticles.

578 Environ. Sci.: Nano **2019**, *6*, (1), 60-67.

579 15. Zhang, P.; Ma, Y.; Zhang, Z.; He, X.; Zhang, J.; Guo, Z.; Tai, R.; Zhao, Y.; Chai, Z.,

580 Biotransformation of ceria nanoparticles in cucumber plants. ACS Nano **2012**, *6*, (11), 9943-9950.

581 16. Zhang, P.; Guo, Z.; Zhang, Z.; Fu, H.; White, J. C.; Lynch, I., Nanomaterial Transformation in

the Soil–Plant System: Implications for Food Safety and Application in Agriculture. *Small* **2020**, *16*,

583 (21), 2000705.

584 17. Ma, Y.; Zhang, P.; Zhang, Z.; He, X.; Zhang, J.; Ding, Y.; Zhang, J.; Zheng, L.; Guo, Z.; Zhang,

585 L., Where does the transformation of precipitated ceria nanoparticles in hydroponic plants take 586 place? *Environ. Sci. Technol.* **2015**, *49*, (17), 10667-10674.

587 18. Rui, Y.; Zhang, P.; Zhang, Y.; Ma, Y.; He, X.; Gui, X.; Li, Y.; Zhang, J.; Zheng, L.; Chu, S.,

588 Transformation of ceria nanoparticles in cucumber plants is influenced by phosphate. *Environ.*

589 *Pollut.* **2015,** *198,* 8-14.

590 19. Chauhan, B. S.; Jabran, K.; Mahajan, G., *Rice production worldwide*. Springer: 2017; Vol.
591 247.

592 20. Belder, P.; Bouman, B.; Spiertz, J.; Peng, S.; Castaneda, A.; Visperas, R., Crop performance, 593 nitrogen and water use in flooded and aerobic rice. *Plant Soil* **2005**, *273*, (1-2), 167-182.

Zhang, Y.; Lin, X.; Werner, W., Effects of aerobic conditions in the rhizosphere of rice on the
dynamics and availability of phosphorus in a flooded soil—a model experiment. *J. Plant Nutr. Soil Sci.* 2004, *167*, (1), 66-71.

597 22. Gao, X.; Hoffland, E.; Stomph, T.; Grant, C. A.; Zou, C.; Zhang, F., Improving zinc

bioavailability in transition from flooded to aerobic rice. A review. *Agron. Sustain. Dev.* 2012, *32*,
(2), 465-478.

600 23. Wan, Y.; Camara, A. Y.; Yu, Y.; Wang, Q.; Guo, T.; Zhu, L.; Li, H., Cadmium dynamics in soil

601 pore water and uptake by rice: Influences of soil-applied selenite with different water

602 managements. *Environ. Pollut.* **2018**, *240*, 523-533.

603 24. Wang, X.; Ye, Z.; Li, B.; Huang, L.; Meng, M.; Shi, J.; Jiang, G., Growing rice aerobically

604 markedly decreases mercury accumulation by reducing both Hg bioavailability and the production
605 of MeHg. *Environ. Sci. Technol.* 2014, 48, (3), 1878-1885.

606 25. Majumdar, S.; Trujillo-Reyes, J.; Hernandez-Viezcas, J. A.; White, J. C.; Peralta-Videa, J. R.;

607 Gardea-Torresdey, J. Cerium biomagnification in a terrestrial food chain: influence of particle size

608 and growth stage. *Environ. Sci. Technol.* **2016**, *50*, (13), 6782-6792.

609 26. Zhao, L.; Sun, Y.; Hernandez-Viezcas, J. A.; Hong, J.; Majumdar, S.; Niu, G.; Duarte-Gardea,

610 M.; Peralta-Videa, J. R.; Gardea-Torresdey, J. L. Monitoring the environmental effects of CeO₂ and

511 ZnO nanoparticles through the life cycle of corn (Zea mays) plants and in situ μ-XRF mapping of

612 nutrients in kernels. *Environ. Sci. Technol.* **2015**, *49*, (5), 2921-2928.

613 27. Carpenter, D.; Boutin, C.; Allison, J. E.; Parsons, J. L.; Ellis, D. M. Uptake and effects of six

614 rare earth elements (REEs) on selected native and crop species growing in contaminated soils.

615 *PLoS One* **2015,** *10,* (6), e0129936.

- 616 28. Ramos, S. J.; Dinali, G. S.; Oliveira, C.; Martins, G. C.; Moreira, C. G.; Siqueira, J. O.;
- 617 Guilherme, L. R. Rare earth elements in the soil environment. *Curr. Pollut. Rep.* **2016**, *2*, (1), 28-50.
- 618 29. Yang, L.; Liu, B.; Lu, Y.; Lu, F.; Wu, X.; You, W.; Huang, B. Bioavailability of cadmium to
- 619 celery (Apium graveolens L.) grown in acidic and Cd-contaminated greenhouse soil as affected by
- 620 the application of hydroxyapatite with different particle sizes. *Chemosphere* **2020**, *240*, 124916.
- 621 30. Dias, M. C.; Santos, C.; Pinto, G.; Silva, A. M.; Silva, S., Titanium dioxide nanoparticles
- 622 impaired both photochemical and non-photochemical phases of photosynthesis in wheat.
- 623 *Protoplasma* **2019**, *256*, (1), 69-78.
- 624 31. Das, D.; Das, P.; Biswas, A. K., Regulation of growth and carbohydrate metabolism in rice
 625 (Oryza sativa L.) seedlings by selenium and sulphate. *J. Plant Stud.* 2018, 7, 61.
- 626 32. McCready, R.; Guggolz, J.; Silviera, V.; Owens, H., Determination of starch and amylose in
- 627 vegetables. Anal. Chem. **1950**, 22, (9), 1156-1158.
- 628 33. Hernandez-Viezcas, J. A.; Castillo-Michel, H.; Peralta-Videa, J. R.; Gardea-Torresdey, J. L.,
- 629 Interactions between CeO2 Nanoparticles and the Desert Plant Mesquite: A Spectroscopy
- 630 Approach. ACS Sustain. Chem. Eng. **2016**, *4*, (3), 1187-1192.
- 631 34. Bates, L. S.; Waldren, R. P.; Teare, I., Rapid determination of free proline for water-stress
 632 studies. *Plant Soil* **1973**, *39*, (1), 205-207.
- 633 35. Maxwell, K.; Johnson, G. N., Chlorophyll fluorescence—a practical guide. *J. Exp. Bot.* 2000,
 634 51, (345), 659-668.
- 635 36. Ma, Y.; Zhang, P.; Zhang, Z.; He, X.; Li, Y.; Zhang, J.; Zheng, L.; Chu, S.; Yang, K.; Zhao, Y.
- 636 Origin of the different phytotoxicity and biotransformation of cerium and lanthanum oxide
- 637 nanoparticles in cucumber. Nanotoxicology **2015**, *9*, (2), 262-270.
- 638 37. Singh, S.; Dosani, T.; Karakoti, A. S.; Kumar, A.; Seal, S.; Self, W. T., A phosphate-dependent
- 639 shift in redox state of cerium oxide nanoparticles and its effects on catalytic properties.
- 640 *Biomaterials* **2011**, *32*, (28), 6745-6753.
- 641 38. Spielman-Sun, E.; Avellan, A.; Bland, G. D.; Tappero, R. V.; Acerbo, A. S.; Unrine, J. M.;
- 642 Giraldo, J. P.; Lowry, G. V. Nanoparticle surface charge influences translocation and leaf
- distribution in vascular plants with contrasting anatomy. *Environ. Sci.: Nano* **2019**, *6*, (8), 2508-
- 644 2519.
- 645 39. Spielman-Sun, E.; Lombi, E.; Donner, E.; Howard, D.; Unrine, J. M.; Lowry, G. V. Impact of
- 646 surface charge on cerium oxide nanoparticle uptake and translocation by wheat (Triticum
- 647 aestivum). Environ. Sci. Technol. **2017**, *51*, (13), 7361-7368.

- Kato, Y.; Okami, M., Root growth dynamics and stomatal behaviour of rice (Oryza sativa L.)
 grown under aerobic and flooded conditions. *Field Crops Res.* 2010, *117*, (1), 9-17.
- 650 41. Bao, Y.; Pan, C.; Liu, W.; Li, Y.; Ma, C.; Xing, B., Iron plaque reduces cerium uptake and
- translocation in rice seedlings (Oryza sativa L.) exposed to CeO2 nanoparticles with different sizes.

652 Sci. Total Environ. **2019,** 661, 767-777.

42. Bao, Y.; Ma, J.; Pan, C.; Guo, A.; Li, Y.; Xing, B., Citric acid enhances Ce uptake and

accumulation in rice seedlings exposed to CeO₂ nanoparticles and iron plaque attenuates the
enhancement. *Chemosphere* **2020**, *240*, 124897.

656 43. Otte, M.; Rozema, J.; Koster, L.; Haarsma, M.; Broekman, R. Iron plaque on roots of Aster
657 tripolium L.: interaction with zinc uptake. *New Phytol.* **1989**, *111*, (2), 309-317.

44. Khan, N.; Seshadri, B.; Bolan, N.; Saint, C. P.; Kirkham, M. B.; Chowdhury, S.; Yamaguchi, N.;

Lee, D. Y.; Li, G.; Kunhikrishnan, A.; Qi, F.; Karunanithi, R.; Qiu, R.; Zhu, Y. G.; Syu, C. H., Chapter

660 One - Root Iron Plaque on Wetland Plants as a Dynamic Pool of Nutrients and Contaminants. In

661 Adv. Agron. Sparks, D. L., Ed. Academic Press: 2016; Vol. 138, pp 1-96.

- 45. Yang, X.; Pan, H.; Wang, P.; Zhao, F.-J., Particle-specific toxicity and bioavailability of cerium
 oxide (CeO2) nanoparticles to Arabidopsis thaliana. *J. Hazard. Mater.* 2017, *322*, 292-300.
- 46. Zhou, M.; Gong, X.; Wang, Y.; Liu, C.; Hong, M.; Wang, L.; Hong, F., Improvement of cerium
 of photosynthesis functions of maize under magnesium deficiency. *Biol. Trace Elem. Res.* 2011,

666 *142,* (3), 760-772.

47. Wang, Y.; Zhang, P.; Li, M.; Guo, Z.; Ullah, S.; Rui, Y.; Lynch, I., Alleviation of nitrogen stress
in rice (Oryza sativa) by ceria nanoparticles. *Environ. Sci.: Nano* 2020, 7, (10), 2930-2940.

669 48. Sun, L.; Zheng, M.; Liu, H.; Peng, S.; Huang, J.; Cui, K.; Nie, L., Water Management Practices

670 Affect Arsenic and Cadmium Accumulation in Rice Grains. *Sci. World J.* **2014**, *2014*, 596438.

49. Wang, X.; Sun, W.; Zhang, S.; Sharifan, H.; Ma, X., Elucidating the Effects of Cerium Oxide

672 Nanoparticles and Zinc Oxide Nanoparticles on Arsenic Uptake and Speciation in Rice (Oryza

673 sativa) in a Hydroponic System. *Environ. Sci. Technol.* **2018**, *52*, (17), 10040-10047.

50. Rossi, L.; Zhang, W.; Schwab, A. P.; Ma, X., Uptake, Accumulation, and in Planta Distribution

675 of Coexisting Cerium Oxide Nanoparticles and Cadmium in Glycine max (L.) Merr. . Environ. Sci.

676 *Technol.* **2017,** *51,* (21), 12815-12824.

677